Report No. 18

Final Report (Technical Report No. III) April 1, 1965 - March 31, 1968

# COMPARATIVE EVALUATION OF ABLATING MATERIALS IN ARC PLASMA JETS

Prepared for:

OFFICE OF RESEARCH GRANTS AND CONTRACTS
NATIONAL AERONAUTICS AND SPACE ADMINISTRATION
WASHINGTON, D.C. 20546
TASK ORDER CONTRACT NASr-49 (15)

STANFORD RESEARCH INSTITUTE

MENLO PARK, CALIFORNIA



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By: NEVIN K. HIESTER & CARROLL F. CLARK

SRI Project PRU-4512

Approved: A. G. BROWN, DIRECTOR

CHEMISTRY AND CHEMICAL ENGINEERING LABORATORIES



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#### ABSTRACT

In a round-robin ablation study monitored by Stanford Research Institute in 1964, Teflon and high-density phenolic-nylon models were evaluated at various enthalpies and heating rates under supersonic conditions. The results of that study, published in NASA Contractor Report CR-379, showed that the best description of the test environments for twelve different plasma arc heater facilities was given by the stagnation point heating rate, and pressure. The mass loss rates from all facilities could be correlated in terms of these parameters with a standard deviation of approximately 11 percent for both the Teflon and phenolic-nylon material.

The second phase of this study, described in this report, also involved twelve facilities, most of them the same as in the first phase. The same two high-density materials, Teflon and phenolic-nylon, were evaluated at stagnation pressures up to 10 atmospheres for the former and 30 atmospheres for the latter. The effect of model size was also evaluated using both hemispherical and flat-faced Teflon models having effective radii varying from one-quarter to four times the radius used previously. The mass loss rates again correlated with the results from the earlier study, except that high-density phenolic-nylon models showed a rapid increase in ablation above a stagnation pressure of about 2.5 atmospheres. These higher rates were also correlated, and found to agree, with literature data for this same pressure regime.

Five new low-density materials—Langley phenolic-nylon, Hughes phenolic-nylon, Avcoat epoxy-novalac-filled honeycomb, Langley silicone elastomer, and General Electric silicone elastomer—were also studied in the second phase at enthalpies from 2500 to 34,000 Btu/lb and stagnation pressures from 0.004 to 2 atmospheres. Mass loss rates, front surface, and internal temperatures were measured for these materials.

Mass loss rate correlations similar to those developed in the first phase of the study were satisfactory except for the silicone materials, which may suggest that the ablation mechanism varies for these in the range of conditions studied. Dimensional analysis was used to develop new correlations for interrelating front surface, internal temperature rise, and test environment. The resulting relations show that data from the various facilities could be satisfactorily compared.

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#### SYMBOLS

Constant-differs for each ablation material a Constant-differs for each ablation material a 0. Sensing area—ft<sup>2</sup> Α Nozzle throat area—ft<sup>2</sup> **A**\* b Constant-differs for each ablation material Constant-differs for each ablation material b<sub>1</sub> Constant-differs for each ablation material  $\mathbf{b}_{0}$ Constant-differs for each ablation material С c 0 Constant-differs for each ablation material  $C_{\mathbf{p}}$ Heat capacity—Btu/lb°F\* Temperature averaged heat capacity—Btu/lb°F  $C_{pAV}$ d Constant-varies for each ablation material Model diameter-ft D Constant-differs for each ablation material е Constant-differs for each ablation material f Conversion constant-2116 lb force/ft<sup>2</sup> atm  $\mathbf{F}_{\mathbf{p}}$ Constant-differs for each ablation material g Constant-differs for each ablation material g o Gravitational constant-32.17 lb ft/lb force sec<sup>2</sup> g c  $\Delta \mathbf{h}$ Enthalpy potential  $(h_t - h_w)$ —Btu/lb Stagnation enthalpy of gas-Btu/lb h s h, Total stream enthalpy—Btu/lb h w Wall enthalpy—Btu/lb  $H_{n}$ Heat of dissociation-Btu/lb  $\Delta H_{D}$ Overall heat of decomposition—Btu/lb Heff Effective heat of ablation =  $\dot{q}_{CW}/\dot{m}_t$ —Btu/lb J Conversion constant-778 ft lb force/Btu

 $<sup>^</sup>st$  When the symbol "lb" is used without the modifying word "force", it always means pound mass.

```
k __
           Catalytic surface activity—cm/sec
    K
           Conversion constant, see equation (E-3B)
           Constant-differs for each ablation material
           Mass loss rate—lb/ft<sup>2</sup> sec
    m
           Mass pyrolysis rate, see equation (19)—lb/ft<sup>2</sup> sec
   m n
           Total mass loss rate—lb/ft 2 sec
   m ,
           Mass of calorimeter slug-lb
    M
           Constant-differs for each ablation material
    n
           Lewis Number
  N
           Prandtl Number
  N_{Pr}
           Pressure-atm
    р
           Reservoir, arc chamber, or plenum pressure—atm
 P_{t}
           Model stagnation pressure-atm
           Heat transfer rate - Btu/ft 2 sec
    ġ
           Heat transfer rate to fully catalytic surface—Btu/ft<sup>2</sup> sec
\dot{q}_{,k} \underset{_{w}}{\sim} _{,\infty}
           Constant-differs for each ablation material
    r
           Model radius-ft
    R
 Reff
           Effective model radius—ft, see equation (15)
           Radius of flat-face cylinder-ft
  R_{FF}
   R_{\rm H}
           Radius of hemispherical shape—ft
           Constant-varies for each ablation material
    s
           Proportionality constant in Fay-Riddell relation—ft1.5
   S_{R}
                       sec atm^{0.5}/lb, see equation (5)
           Time-sec
    t
   \Delta \mathbf{t}
           Run time-sec
           Model front surface temperature-oF or oR
  T_{FS}
           Temperature rise—°F
   \Delta T
```

XII

Slug temperature rise rate—°F/sec

Constant-differs for each ablation material

Constant-differs for each ablation material

Constant-differs for each ablation material

 $\Delta T/\Delta t$ 

11

v

w

- w Cooling water flow rate-lb/sec
- W Gas mass flow rate—lb/sec

h. 1

- $W_{250}$  Weight of ablation material per unit area—lb/ft<sup>2</sup>, see equation (40)
  - x Linear distance from original model surface—in.
  - X Linear measurement along the model axis—ft
  - y Constant-differs for each ablation material
  - z Constant-differs for each ablation material
  - $\alpha$  Constant—heat necessary to raise the material to the ablation temperature and to decompose it—Btu/lb
  - $\beta$  Transpiration shielding factor
  - € Total surface emissivity
  - $\mu$  Viscosity—lb/ft sec
  - $\pi_{\rm A}$  Dimensionless group, involving char failure stress, see equation (12)
  - $\pi_{\mathrm{D}}$  Dimensionless group, involving density, see equation (49)
  - $\pi_{\mathbf{f}}$  Dimensionless group based on Fay-Riddell relation, see equation (E-7)
  - $\pi_{\rm h}$  Dimensionless group, involving enthalpy potential, see equation (29)
  - $\pi_{_{\mathrm{m}}}$  Dimensionless group, involving mass loss rate, see equation (E-4)
  - $\pi_{\rm p}$  Dimensionless group, involving stagnation pressure see equation (E-5)
  - $\pi_{q}$  Dimensionless group, involving heat transfer rate, see equation (E-6)
  - $\pi_{\rm s}$  Dimensionless group, involving front surface temperature, see equation (41)
  - $\pi_{\rm t}$  Dimensionless group, involving exposure time, see equation (48)
  - $\pi_{T}$  Dimensionless group, involving internal temperature rise, see equation (47)
  - $\pi_{\chi}$  Dimensionless group, involving position in model, see equation (46)
  - ρ Density—lb/ft<sup>3</sup>
  - $\rho_{\rm VR}$  Density of virgin ablation material—lb/ft<sup>3</sup>

- $\sigma$  Stefan-Boltzmann constant = 4.76 × 10<sup>-13</sup> Btu/ft<sup>2</sup> sec °R<sup>4</sup>
- 7 Mechanical stress on char—lb force/ft<sup>2</sup>
- $\psi({
  m N_{Le}})$  Function of the Lewis Number for the gas, see equation (E-2)

#### SUBSCRIPTS

250

AV	Average value
c	Char surface location
CALC	Calculated value
CR	Char thickness
CW	Cold wall condition
FAC	Instrument supplied by facility
meas	Measured value
P	High-density phenolic-nylon material
R	Char recession
RR	Phase I round robin
s	Stagnation condition
SRI	Instrument supplied by SRI
T	Teflon material
1.25FF	Refers to 1.25-indiameter, flat-faced models tested during Phases I and II round robins

Free stream condition

Internal temperature rise of 250°F

#### INTRODUCTION

The Office of Research Grants and Contracts, National Aeronautics and Space Administration, (NASA), in 1963 asked Stanford Research Institute to act as a program manager on a round-robin test study to determine whether ablation tests of representative materials at different plasma arc heater facilities would yield self-consistent results.

This work involved definition of the extent to which realistic environmental conditions are simulated by such devices; conduction of comparative ablation tests on standardized materials at selected organizations possessing suitable equipment; provision of the specialized instrumentation and test models required; and correlation of test results with analyses to determine the feasibility of developing a standardized method.

The twelve participating organizations, five government and seven industrial, tested over 170 models and, in addition, performed numerous calibration experiments. The resulting data, published in NASA Contractor Report CR-379, 1 showed that

- 1. A procedure for comparing ablation test results (on a given material) at each supersonic plasma arc heater facility is feasible through use of a standard mass loss rate, heating rate (or calculated enthalpy), and stagnation pressure correlation.
- 2. The applicability of the procedure outside the range of materials, model sizes, and arc heater operating conditions studied in the program needed further investigation.

The program was subsequently extended to assess the validity of the findings of the Phase I study and to determine their generality by providing a more detailed comparison of results over a wider range of ablation variables. This involved the study of more severe test conditions, changes in model geometry, new low-density materials, and more extensive measurements on the ablating models. Thus, the Phase II research, which was also to involve a round robin, fell naturally into four major categories—facility parameters, model parameters, measurements, and analysis of results. More specifically, the following studies were to be considered:

- 1. Facility Parameters
  - a. Higher stagnation pressures
  - b. Uniformity of plasma stream
- 2. Model Parameters
  - a. Geometry
  - b. New materials
- 3. Measurements
  - a. Front surface temperature
  - b. Internal temperature rise
  - c. Mass and length changes
  - d. Char behavior
- 4. Analysis of Results
  - a. Comparison of measurement techniques
  - b. Correlation of data

#### II SUMMARY

The ablation conditions studied in Phase I with the high-density Teflon (T)\* and phenolic-nylon (P) materials were extended to higher stagnation pressures and to models of different shapes and dimensions. In addition, five new low-density materials were evaluated during Phase II of the round robin, and more extensive front surface and internal temperature measurements were made with the models. The new materials were

- Langley phenolic nylon (PLL), density = 35.5 lb/ft<sup>3</sup>
- Hughes phenolic- nylon (PLH), density =  $35.7 \text{ lb/ft}^3$
- Avcoat epoxy-novalac-filled honeycomb (A), density =  $31 \text{ lb/ft}^3$
- $\bullet$  Langley silicone elastomer (SP), density = 33.5 lb/ft  $^3$
- General Electric silicone elastomer (SG), density = 36.8 lb/ft3

Insofar as possible the same test facilities were used in the new program. Several new organizations were added to replace those which could not be used and to provide capabilities at higher test pressures or larger model dimensions. The twelve participants, six government and six industrial, finally chosen were

- 1. Gas Dynamics Branch, Ames Research Center-NASA
- 2. Magneto Plasma Dynamics Branch, Ames Research Center-NASA
- 3. Applied Materials and Physics Division, Langley Research Center-NASA  $\dagger$
- 4. Entry Structures Branch, Langley Research Center-NASA
- 5. Manned Spacecraft Center, Subsonic Facility—NASA†
- 6. Manned Spacecraft Center, Supersonic Facility-NASA
- 7. Aerotherm Corporation
- 8. AVCO Corporation<sup>†</sup>
- 9. Giannini Scientific Corporation
- 10. Martin Company<sup>†</sup>

SRI designated code for these materials.

<sup>†</sup> Participant in Phase I round robin

- 11. Space General Corporation
- 12. Cornell Aeronautical Laboratory, Inc.

Calorimeters and ablation models, some instrumented with internal thermocouples, were supplied to each participant for use in the round-robin test program. Radiometers were also supplied for determining front surface temperature. In addition to model and calibration runs, each facility was asked to make a heating rate and stagnation pressure traverse of the plasma stream. The results showed that all the test facilities exhibited some nonuniformity of the stream, with "coring" occurring in several cases. However, the degree of uniformity in the region of the model was satisfactory in most cases.

The SRI and facility calorimeters when compared in the test environment had a standard deviation of 12 percent, which is slightly better than that found in the Phase I round robin. As in the earlier study, it was observed that when the plasma flow through the supersonic nozzle is far removed from equilibrium, such as with high expansion ratios or very low test pressures, surface catalytic effects on the calorimeter will influence the measured heating rate.

Comparison of the measured enthalpy to that calculated from the heating rate and stagnation pressure, through the Fay-Riddell relation, was not satisfactory. As in the Phase I study, those facilities having quite uniform stream traverses showed good agreement between the two values, indicating that in those cases the center-line enthalpy is probably close to the average enthalpy by the energy balance method. Stream enthalpy is a most important variable in material ablation studies, yet it is the most difficult to measure accurately.

The primary measurements made on the models were weight loss, recession, char depth, char density, front surface temperature, and internal temperature. These data, along with heating rate, stagnation pressure, and enthalpy, were the inputs for correlation of the data.

Initial interpretation involved further evaluation of the data from the Phase I round robin. Dimensional analysis suggested the following dimensionless relation, involving groups proportional to mass loss rate,  $\dot{\mathbf{m}}_{\rm t}$ , heating rate,  $\dot{\mathbf{q}}_{\rm CW}$ , and stagnation pressure,  $P_{\rm t_2}$ , and also containing the effective radius of the model,  $R_{\rm eff}$ , and the overall heat of decomposition,  $\Delta H_{\rm D}$ , for each material:

$$\pi_{\rm m} = a_0 \pi_{\rm q}^{\rm n} \pi_{\rm p}^{\rm m}$$

In the absence of a means to determine  $\triangle H_D$  independently, the dimensionless correlation was expanded into a dimensional form similar to those described in the Phase I report (NASA Report No. CR-379), namely,

$$\dot{m}_t = b(R_{eff})^{n+m-1}(\dot{q}_{CW})^n(P_{t_2})^m$$

A considerable amount of supersonic arc-jet test data on Teflon and high-density phenolic-nylon materials is available in the literature. Inclusion of these data for Teflon, along with the Phase I round-robin results, leads, by regression analysis, to the following values of b, n, and m:

$$(\dot{m}_t)_T = 0.0046(R_{eff})^{-0.21}(\dot{q}_{CW})^{0.55}(P_{t_2})^{0.24}$$

with a standard deviation of 10 percent. For high-density phenolic-nylon, the relation is

$$(\dot{m}_t)_p = 0.0010(R_{eff})^{-0.32}(\dot{q}_{CW})^{0.55}(P_{t_2})^{0.13}$$

with a standard deviation of 10 percent.

The results of the Phase II, high stagnation pressure experiments were interpreted in terms of the above dimensional correlations. The Teflon data were found to fit the relation up to the highest pressure used, 33 atm. On the other hand, at stagnation pressures above 2.7 atm, the high-density phenolic-nylon data showed higher mass loss rates than predicted by the above relation. These phenolic-nylon data, plus literature data obtained under similar conditions, fit a second correlation:

$$(\dot{m}_t)_P = 0.0010(\tau/2116)^{-0.75}(R_{eff})^{-0.32}(\dot{q}_{CW})^{0.55}(P_{t_2})^{0.13+0.75}$$

where  $\tau$  is the mechanical stress, in pounds force per square foot, at which failure of the char occurs. This relation, derived by dimensional analysis, permits determination of  $\tau$  from the intercept of the new correlation line.

The stress of the char at failure was found to be 5610~lb force/ft.<sup>2</sup> This assumption of char failure appears valid since the high-density phenolic-nylon models showed almost no char after exposure to the high stagnation pressure environments and thus had reduced thermal protection.

A limited number of studies were performed using variable radii models and calorimeters. The heat transfer data showed the proper inverse relationship with square root of calorimeter shroud radii. The mass loss data also showed the proper effect of radius but exhibited greater scatter than for the standard models. Plasma coring and stream blockage may be partially responsible.

The low-density materials ablated somewhat differently in terms of char appearance. With the low-density phenolic-nylon materials the char was cracked with a columnar structure oriented parallel to the direction of ablation. The silicone chars had two types of appearance. At low heat fluxes, they were black, but at higher heating rates the surface showed a grey, fused, inorganic coating, apparently due to the formation of SiO<sub>2</sub>. The Avcoat chars usually had a depression in the honeycomb filler material, with the web being slightly raised and with fused droplets of inorganic material at the model periphery.

The properties and composition of these chars are less reproducible than those of the high-density materials. This is also true of the appearance and dimensions of the charred core, the latter being difficult to measure.

The mass loss rate data for the five low-density materials all showed the same form of correlation as for the high-density materials, namely,

$$\dot{m}_{t} = a(\dot{q}_{CW})^{n}(P_{t}_{2})^{m}$$

The effective radius term has been combined with b in this case and replaced by "a" since no planned studies of these materials were made with varying radii models. The values of the constants are

MATERIAL	a	n	m	PERCENT STANDARD DEVIATION
PLL	0.0047	0.36	0.26	15
PLH	0.0039	0.36	0.19	14
Α	0.0036	0.47	0.33	16
SP	0.00032	0.81	0.19	24
SG	0.00019	1.03	0.28	36

Note that the standard deviation is higher than that found for the high-density materials in the Phase I round robin owing to the more difficult char measurements. The correlations for the silicone materials are particularly poor, suggesting that the mechanisms of ablation may vary in the range of conditions studied. Therefore, less credence should be given to the correlation constants found for these latter materials. Attempts to correlate mass loss rates with other variables were no more successful.

Front surface temperature, as measured with facility optical pyrometers, correlated well with pyrolysis rate and stagnation pressure, thus

$$T_{FS} = a(P_{t_2})^{v(\dot{m}_p)^{w}}$$

The values of the constants are

+ 1

MATERI AL	a	v	ŵ	PERCENT STANDARD DEVIATION
PLL	10,980	0.031	0.21	5
PLH	10,710	0.044	0.20	6
Α	10,040	0.039	0.18	6
SP	7,660	0.012	0.16	4
SG	5,210	0.028	0.072	.5

This relation was derived by dimensional analysis and then expanded into the above form.

The internal temperature profiles were correlated in terms of heating rate and stagnation pressure by adding the position and time at which a given temperature rise occurs. The correlation is

$$x = a(P_{t_2})^b(\dot{q}_{SRI})^c(t)^d(\Delta T)^e$$

The values of the constants are

MATERIAL	a	<b>b</b>	Ċ	d	e	STANDARD DEVIATION
PLL	0.034	0.053	0.30	0.63	-0.28	13
PLH	0.056	0.035	0.15	0.58	-0.24	9
Α	0.037	0.018	0.27	0.60	-0.26	14
SP	0.072	0.022	0.18	0.52	-0.30	12
SG	0.12	0.031	0.098	0.54	-0.28	8

DEBCENT

This relation compares with analogous relations and with the results in the literature. A markedly improved correlation is obtained when only the  $250\,^{\circ}\mathrm{F}$  temperature rise isotherm is considered. The relation in this case is

\* \_ \*

$$x_{250} = a(P_{t_2}) b(\dot{q}_{SRI}) c(t_{250}) d$$

and the constants are

а	ь	, <b>c</b>	ď	STANDARD DEVIATION
0.014 0.017 0.023 0.0082	0.083 0.079 0.105 0.016	0.18 0.14 0.15 0.26	0.61 0.60 0.62 0.54	8 5 6 9 2
	0.014 $0.017$ $0.023$	0.014     0.083       0.017     0.079       0.023     0.105       0.0082     0.016	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$	$\begin{array}{c ccccccccccccccccccccccccccccccccccc$

Both of the temperature rise correlations were derived by dimensional analysis.

The data obtained for the low-density materials can be used to compare their ablation performance. For environments leading to low front surface temperatures, the silicone materials show the lowest mass loss rates. However, at high front surface temperatures, the silicones perform much more poorly than the low-density phenolic-nylon materials. This marked difference undoubtedly relates to the chemical reactions involving silicon, oxygen, and carbon. Below the melting point of silica, the surface is protected by this material and some silicon carbide. Above the melting point, however, the silica reacts with carbon to form carbon monoxide and silicon monoxide which are rapidly lost as vapors. This change in ablation mechanism undoubtedly causes the difficulties in correlating the silicone mass loss data. The internal temperature rises show that the best insulator, on both a volume and a weight basis, is the Langley low-density phenolic-nylon.

#### III EXPERIMENTAL PROGRAM

The Phase II round-robin study was organized and undertaken in a very similar manner to the Phase I program. The following sections describe the study in more detail.

#### A. Scope and Participants

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The test environments, model responses, and ablation materials to be studied are outlined in the scope. Choice of the organizations to participate in the round robin were based on somewhat similar criteria to those used in the earlier study.

#### 1. Scope of Program

Representatives of the Ames Research Center, Langley Research Center, Manned Spacecraft Center, Jet Propulsion Laboratory, Advanced Research Programs Office of NASA, and Stanford Research Institute met early in the Phase II program to determine the test conditions, model dimensions, and materials to be evaluated. It was agreed that three separate areas should be studied in the new program, as follows:

- 1. Extension of previous test conditions
- 2. Variation from previous model dimensions
- 3. Addition of new low-density materials

In the first area, the major interest was in increased stagnation pressures. In order to minimize changes in other variables, model dimensions were kept the same as in Phase I, and the same materials were used for the study, namely,

- Teflon, type TFE 7, white variety, density =  $135.6 \text{ lb/ft}^3$
- Phenolic-nylon (50-50%), density =  $74.3 \text{ lb/ft}^3$

The second area involved changes in effective diameter, and both larger and smaller models than used previously were considered. Test conditions were kept the same as in Phase I, and Teflon was used as one of the materials. However, because of an insufficient quantity of

high-density phenolic-nylon polymer from the previous program, a low-density version was used instead, namely,

• Low-density phenolic-nylon (Hughes 5), density =  $35.7 \text{ lb/ft}^3$ 

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The major purpose of the third area of the program was to compare certain low-density, charring ablators. For this reason, test conditions and model dimensions were kept the same as in Phase I, but more extensive measurements were taken on the models. In addition to the Hughes low-density phenolic-nylon (noted above), the new materials involved were

- Low-density phenolic-nylon (Langley Scout R/4B) density = 35.5 lb/ft<sup>3</sup>
- Epoxy-novalac-filled honeycomb (Avcoat 5026-39 HC/G) density =  $31 \text{ lb/ft}^3$
- Silicone elastomer (Langley Modified Purple Blend E4A1) density = 33.5 lb/ft<sup>3</sup>
- Silicone elastomer (G.E. ESM 1004AP), density = 36.8 lb/ft<sup>3</sup>.

Phase II of the round robin thus consisted of the exposure of the above models under the appropriate conditions at various arc-heated plasma jet facilities. The participants supplied information about test conditions and the Institute measured the physical and chemical changes in the models.

#### 2. Selection of Participating Organizations

Several factors governed the selection of supersonic testing facilities to participate in the new round-robin program. These were: (1) that the test models could be accommodated in the plasma stream, (2) that the facility operate in the range of test conditions desired, and (3) that insofar as possible the facilities used were either participants in the Phase I round robin or would bring a new capability to the study.

All twelve of the organizations used in the Phase I round robin were contacted and asked to indicate their interest in further work. Lack of facility time or high preliminary cost estimates eliminated five of the twelve. Other organizations were then contacted and their facilities assessed, using the criteria mentioned in the Phase I report. Inspection trips were made to the new facilities.

Of particular interest was the ability of supersonic plasma arc jets to test models as large as 5 inches in diameter or at stagnation pressures up to 30 atm. These facilities would be used to extend the Phase I studies on Teflon and high-density phenolic-nylon materials.

As in the Phase I round robin, availability of funds determined the number of commercial participants. Two of these had high stagnation pressure capabilities (AVCO Corporation and Cornell Aeronautical Laboratory), and one had a capability for testing large models (Martin Company). After including these organizations in the program, the remaining funds permitted the selection of four participants for the studies on the ablation of low-density materials, namely

- Aerotherm Corporation
- Giannini Scientific Corporation
- Martin Company

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Space General Corporation

Three of the commercial organizations had participated in the Phase I round robin: Giannini, Martin, and AVCO.

Five government organizations also agreed to participate in the studies on the ablation of low-density materials, namely,

- Gas Dynamics Branch, Ames Research Center-NASA
- Magneto Plasma Dynamics Branch, Ames Research Center— NASA
- Applied Materials and Physics Division, Langley Research Center—NASA
- Entry Structures Branch, Langley Research Center-NASA
- Manned Spacecraft Center -- NASA

The last of these, Manned Spacecraft Center, has a subsonic facility which was included to provide a comparison with supersonic facility results.

The Entry Structures Branch had a higher enthalpy facility under construction and had planned to use it as well as the low enthalpy arc jet used in the Phase I round robin. Delay in completing the new facility prevented the performance of any tests on it. The Applied Materials and Physics Division also had a capability for testing large models and therefore participated in that portion of the program.

#### B. Models and Instrumentation

Approximately thirty ablation models, one calorimeter, and one total radiation pyrometer were furnished by SRI to each participant. Each facility also provided instrumentation to monitor the test environments.

#### 1. Models

In general, the models had the same configuration as in the earlier study. The new materials introduced additional fabrication problems and the need for thermocouple instrumentation.

#### a. Fabrication

The ablation models used in the second NASA round robin were machined from the materials listed in Table I.

The Langley low-density phenolic-nylon was supplied as two 12-in.-diameter × 4-in.-thick billets, and the Hughes phenolic-nylon was furnished in the form of five 12-in.-diameter × 1.5-in.-thick pieces. One-quarter inch of material was discarded from the periphery of all low-density phenolic-nylon billets to ensure uniform models.

The Avcoat material was supplied as two  $12 \times 12 \times 2$ -in. sheets, and model cores of this material were cut with a single honeycomb centered in the core face. The Avcoat and low-density phenolic-nylon materials were fabricated with high-speed cutting techniques.

The Modified Purple Blend models were machined approximately ten percent oversize from a 16-in.-diameter  $\times$  4-in. billet, and then cured at  $100^{\circ}\text{C}$  for four hours. The heat-up rate before the start of cure was  $150^{\circ}\text{C}$  per hour, and the cool-down rate after cure was  $75^{\circ}\text{C}$  per hour. After cure, the models were machined to size by high-speed cutting and grinding.

The General Electric silicone material was supplied in the form of two  $24 \times 24 \times 1$ -in. sheets. Models of this material were fabricated by rough, high-speed cutting followed by high-speed grinding.

The 1.25-in.-diameter Teflon and high-density phenolic-nylon models were fabricated from the same material used in the Phase I round-robin tests, as described in the report on that study. The large-diameter Teflon models were prepared from the identical grade of material provided by the same supplier used in the first round robin.

Table I

ABLATION MATERIALS EVALUATED

	ORGANIZAT	ATION	IDENTIFICATION	SRI	DENSITY	MOTHER CHILDS
MATERIAL	Manufacturer	Supplier	CODE	PREFIX DESIGNATION*	(lb/ft <sup>3</sup> )	APPROXIMALE COMPOSITION
Low-Density Phenolic-Nylon	Langley	Langley	Scout R/4B FB/30 AM/30 Scout R/4B FB/31 AM/31	PLL PLL	35.5 35.5 35.5	Phenolic resin, 25%; phenolic Microballoons, 25%; nylon powder, 50%
Low-Density Phenolic-Nylon	Hughes Aircraft	Amès	Hughes 5 - No. 3 Hughes 5 - No. 4 Hughes 5 - No. 5 Hughes 5 - No. 6 Hughes 5 - No. 7	PLH PLH PLH PLH	36.7 35.7 35.6 35.8 35.6	Hexa-cured phenolic-novalac, phenolic Microballoons, 23%; nylon powder, 40%
Avcoat 5026-39 HC/G	AVCO	Manned Spacecraft Center	K5-1 3002-7 N5-80070	A A	30.8 31.2 31	Proprietary filler; epoxy-novalac resin, chopped Fiberglass, phenolic Microballoons; in Fiberglass honeycomb
Modified Purple Blend Silicone	Langley	Langley	E4Al Elastomer, RDY 150, H3365	SP	33.5	Sylgard 182 silicone, 75%; Eccospheres, 15%; phenolic Microballoons, 10%
General Electric Silicone	General Electric	Jet Propulsion Laboratory	ESM 1004AP	SG	36.8	Methylphenyl silicone elastomer; aluminum silicate fibers, 12%; small amount of iron oxide
Teflon	Du Pont	R. S. Hughes	TFE 7, Teflon	T	135.6	Polytetrafluoroethylene
High-Density Phenolic-Nylon	Ames	Ames	PI, Pl2	þ	74.3	Phenolic resin, 50%; nylon powder, 50%

\* These letter symbols will be used hereafter in text, tables, figures, and appendices to denote the materials investigated.

All models were constructed with removable cores to simulate one-dimensional heat flow to the model stagnation region. Since the low-density materials being evaluated were porous in structure, the back face of the core and the inside surface of the shroud were coated with a thin layer of RTV silicon to prevent hot gas from passing through the material. This simulates having the material bonded to a substrate. The core diameter was one half of the total model diameter for all sizes.

#### b. Thermocouple Instrumentation

The cores of approximately sixty models were instrumented with four 36-gauge Chromel-Alumel thermocouples spaced at 0.1-in. intervals back from the model front surface. The thermocouples were formed with a Dynatec thermocouple welder.

In the case of the two elastomeric materials, the thermocouples were inserted into the model cores with a hypodermic needle using a positioning jig. The low-density phenolic-nylon and Avcoat cores were instrumented by drilling 0.007-in.-diameter longitudinal holes and drawing in the thermocouples. After assembly, the instrumented models were X-rayed at  $90^{\circ}$  planes, and the thermocouple positions measured on the X-ray films with a Telereadex viewer.

Sketches of the instrumented and uninstrumented models are shown in Fig. 1. The assembled 1.25-in.-diameter models, with their plastic shipping containers, are shown in Fig. 2. The uninstrumented Teflon and low-density phenolic-nylon models with diameters ranging from a 1-in. hemisphere to 5 inches flat face are shown in Fig. 3.

#### 2. Instrumentation

Two instruments, a calorimeter and a radiometer, were supplied by SRI for use in the experiments. All other instrumentation at the test facility was made available by the participating organization.

#### a. SRI Calorimeter

The SRI calorimeter supplied to each facility was identical in dimensions and shape to the calorimeter used during the Phase I round robin. The calorimeter was a transient, slug type based on a design developed at Ames Research Center, NASA. The slug was oxygen-free copper with a 0.5 mil-thick nickel plating on the front face. The slug was isolated from the copper shroud by three sapphire bearings, as shown in Fig. 4. The slug diameter was 0.625 in., which was equal to the sample

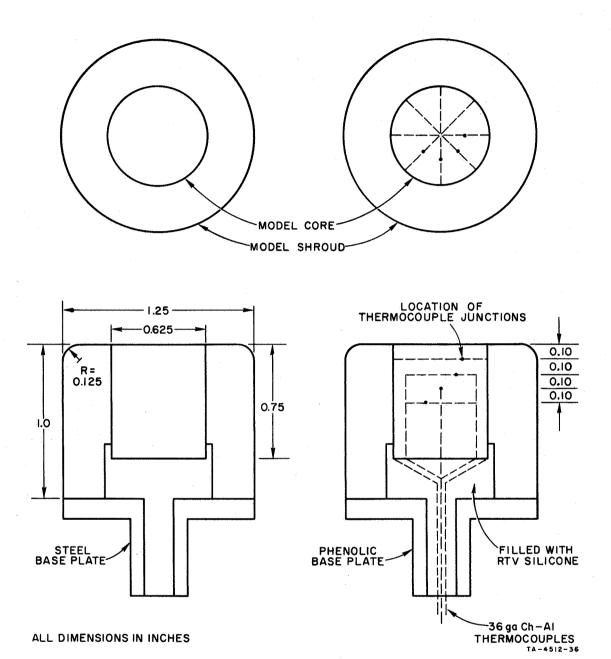


FIG. 1 DIMENSIONS OF MODELS

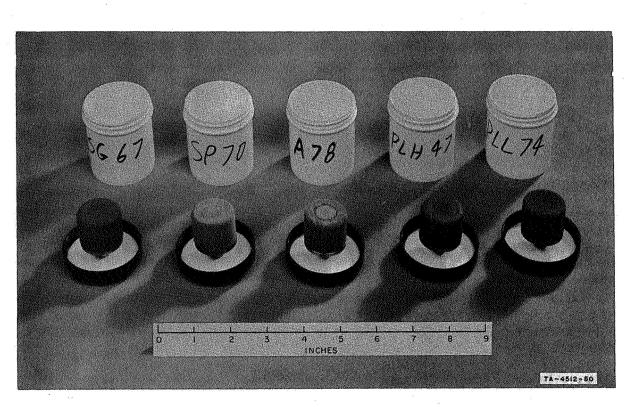


FIG. 2 MODELS FABRICATED FROM LOW-DENSITY MATERIALS

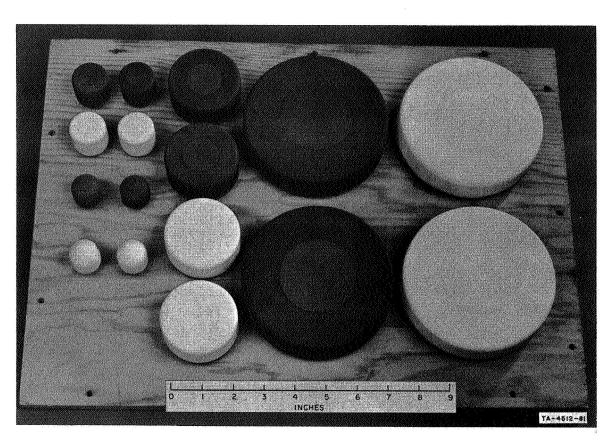


FIG. 3 MODELS OF VARIABLE RADIUS

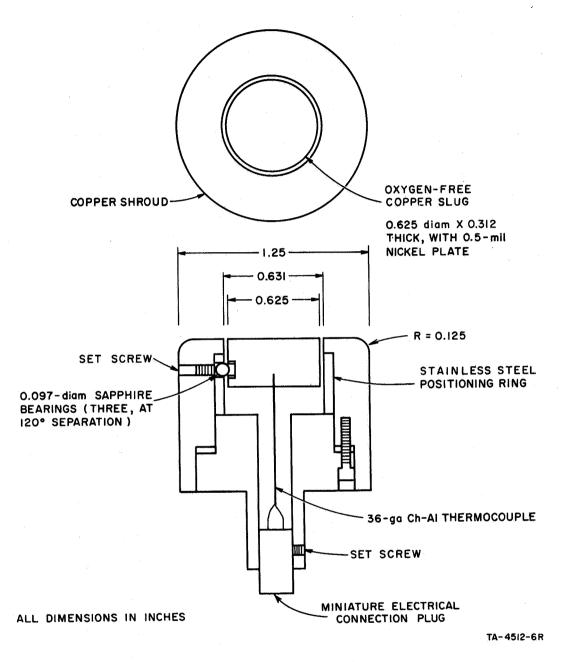


FIG. 4 DESIGN AND DIMENSIONS OF SRI CALORIMETER

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core diameter in the 1.25-in.-diameter models. The average slug temperature was sensed by two parallel, 36-gauge Chromel-Alumel thermocouples peened into holes in the slug base.

The weight of each slug in pounds was stamped on the calorimeter base, and each facility was provided with a graph of the slug specific heat versus temperature. The heat flux was calculated by the facility, using the following relationship:

$$q_{\substack{SRI \\ CW}} = \frac{MC_{PAV}}{A} \frac{\Delta T}{\Delta t}$$
 (1)

where

 $\dot{q}_{SRI}$  = heat transfer rate, (cold wall), SRI calorimeter—Btu/ft<sup>2</sup> sec

M = mass of the calorimeter slug—lb

A = calorimeter sensing area—0.00213 ft<sup>2</sup>

C = temperature averaged heat capacity of copper—Btu/lb°F

 $\frac{\Delta T}{\Delta t}$  = slug temperature rise rate—°F/sec

#### b. SRI Radiometer

In an effort to minimize the scatter in front surface temperature data that was observed in the Phase I round robin when each facility had its own special pyrometer, each facility was provided with an identical reference pyrometer for measuring the front surface temperature of the model. The selection of this pyrometer was governed by the necessity for a moderate cost, durable instrument; and a radiometer, or total radiation-type pyrometer, was chosen as best satisfying these requirements. The instrument selected, which will later be referred to as the SRI radiometer, was a Honeywell Radiamatic Detector, Model R12-354546-7. This instrument had a fused silica lens and viewed the total radiation over the range of wavelengths from 0.3 to 3.9 microns. The instrument required a 0.5-in... diameter target with a 24-in. sighting distance to the target; it required a larger size target for greater sighting distances.

The millivolt output of the SRI radiometers was calibrated over a range of temperatures from 2000 to 4600°F by viewing an inductively heated graphite black body. The black body cavity was 1.5 in. internal

diameter × 2.0 in. height, with an 0.75-in-diameter top viewing port. The viewed bottom of the cavity was covered with a layer of lamp black. The entire graphite block was surrounded by two molybdenum radiation shields and an argon-filled chamber with the glass viewing port removed. The cavity temperature was monitored with a Micro Optical Pyrometer No. 95 and a Leeds-Northrup optical pyrometer. Each facility was provided with plots of the radiometer output versus temperature plus instructions for mounting the radiometer.

#### c. Facility Instruments

The equipment and instruments that were used by each facility for the Phase II round-robin ablation tests are summarized in Appendix A. This information was based on data collected at the time the model tests were witnessed. A detailed description of each facility is beyond the scope of this report, and the information contained in Appendix A is intended only as a brief summary of this equipment.

#### C. Experimental Procedures

Each participating facility received a run plan, specifying three tunnel test conditions and the test run times. Tunnel test conditions were selected that would be within the capability envelope for each facility and at the same time provide testing of the ablation materials over the widest range of conditions. The tunnel conditions specified were enthalpy, heating rate, and stagnation pressure. The range of test parameters and the number of models involved in the Phase II round robin are given in Table II.

Three models were tested at varying run times at the tunnel condition giving the lowest heating rate. The run times were set to give total heat loads  $(\dot{q}_{\text{CW}}\Delta t)$  of 1500, 3000, and 5000 Btu/ft<sup>2</sup>. Two of the three models were uninstrumented, and the remaining model was instrumented with four thermocouples. The instrumented model was usually run at the highest heat load condition.

The two uninstrumented models were tested at varying run times for each of the other two tunnel conditions. The total heat loads for these conditions were 2000 and 5000 Btu/ft $^2$ .

Table II

TEST PARAMETERS FOR NASA ROUND-ROBIN ABLATION PROGRAM

		· :	PHASE II - ROUND ROBIN		
	DENSITY lb/Ft <sup>3</sup>	PHASE I - ROUND ROBIN	TASK A HIGH PRESSURE	TASK B VARYING R <sub>eff</sub>	TASK C NEW MATERIALS
TUNNEL CONDITIONS		1,500,100,000	1500 5000	0500 10 000	2000 24 000
Enthalpy Range (Btu/lb)		1500-16,000	1700-7000	9500-12,000	3000-34,000
Heating Rate Range (Btu/ft <sup>2</sup> sec)		40-700	500-3300	75-450	50-1100
Model Stagnation Pressure Range (atm)	:	0.006-1.0	0.3-30	0.02	0.004-1.0
Model R <sub>eff</sub> (ft)		0.172	0.172	0.0416-0688	0.172
NUMBER TESTED					
Langley P-N Scout R/4B	35.5				.65
Hughes P-N H-5	35.7			16	73
Avcoat 5026-39 HC/G	31				70
Modified Purple Blend E4Al	33.5				56
G.E. Silicone ESM1004AP	36.8				55
Teflon	135.6	76	7	16	5
High-Density Phenolic-Nylon	74.3	97	8		5
		PARTICIPANTS	<u> </u>	<u> </u>	

MPDB, Ames - II C General Dynamics - I General Electric - I AMPD, Langley - I, II B, II C Giannini - I, II C ESB, Langley - I, II C MSC, Houston (supersonic) - II C Martin - I, II B, II C MSC, Houston (subsonic) - I, II C North American - I Space General - II C FMD, Wright-Patterson AFB - I Aerotherm - II C Cornell - II A I = Phase I, II A = Phase II - Task A AVCO - I, II A, II C

The participating organization then provided information on the test environment and the model response. Each facility was also requested to make a heating rate and pressure profile survey of the jet stream for each of the three test conditions.

# 1. Measurement of Test Environment

The tunnel operating conditions that were used for each ablation test are tabulated in Appendix B. The tables contain all data reported by the facilities in their original form; that is, if the facility reported the tunnel calibration data separately, they are listed separately in Appendix B. Some facilities with limited insertion capability combined calibration and run data, although these were obtained at different times. The tables contain pertinent footnotes on the facility measurement techniques.

Although the tunnel conditions and run times were specified by SRI, an effort was made not to influence the measurement techniques and methods used by each facility. The only instructions issued by the Institute covered the use of the SRI calorimeter and radiometer.

## a. Enthalpy

Eight of the twelve participating facilities measured the average total enthalpy of the plasma stream with a single technique; three organizations used two methods; and one group used three methods.

Ten of the facilities measured the mean enthalpy of the plasma stream by the energy balance method; three groups used the sonic flow method; and two groups calculated a local enthalpy from the heat transfer data. Cornell calculated the enthalpy of the test gas in the Wave Superheater from the temperature and pressure of the helium driver gas.

None of the groups used an enthalpy probe to determine localized enthalpy, and while interest in this type of instrument is continuing, the results to date have been somewhat discouraging. The local enthalpy in the vicinity of the model can be inferred from the Fay-Riddell relation when flow is supersonic, and it was calculated from the heating rate and stagnation pressure traverses that were made at each facility. These traverses are reported in Section IV-A.

Some of the problems and difficulties encountered when measuring enthalpy by various methods are detailed below.

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- (1) Energy Balance Enthalpy. The majority of participants preferred the energy balance method of measuring enthalpy. The average enthalpy of the plasma stream was calculated by subtracting the heat losses in the arc generator and nozzle from the total input power and dividing the resulting net power by the mass gas flow. The power losses were determined by measuring the cooling water flow rates and the small temperature rise of the water as it passes through the apparatus. Some facilities used a thermopile arrangement of the thermocouples to increase the accuracy of the water temperature rise measurements. The energy balance method is simple in concept but may require from five to ten separate readings, each with its attendant error, and the accumulated errors can be considerable.
- (2) Sonic Flow Enthalpy. The mean total enthalpy,  $h_t$ , of the jet can be calculated from the gas mass flow rate, W, the reservoir pressure,  $P_{t_1}$ , and nozzle throat area,  $A^*$ , according to the following relationship: <sup>2</sup>

$$h_{t} = (280P_{t_{1}}A^{*}/W)^{2.5}$$
 (2)

One problem encountered with this method is the difficulty of measuring a true static chamber pressure, since most arc heaters are vortex or magnetically stabilized, which can result in a dynamic pressure component. Any measurement error is magnified when raised to the power indicated in equation (2). Another difficulty arises when the stream is not in chemical and thermodynamic equilibrium. A correction for frozen flow that increases with increasing enthalpy must then be added to the above relationship.

A modification of the sonic flow method was developed by R.Pope<sup>3</sup> of the Gas Dynamics Branch at Ames Research Center whereby the temperature of the gas in the reservoir prior to expansion in the nozzle is calculated. The calculation then permits the enthalpy of the plasma stream in the center-line area of the model to be determined. 4-6

(3) Heat Flux Enthalpy. A local enthalpy of the plasma stream can be calculated from the cold wall heat flux using the relationships of Fay-Riddell<sup>7</sup> or Lees. 8 This method has the advantage of indicating an enthalpy in the same area of the stream as the ablating samples are

exposed to and will be described in Sec. IV-A. Its disadvantages are variations in the heat flux measurement resulting from geometry and recombination effects; these will be discussed later.

## b. Heat Flux

The calorimeters that were used by the various facilities are described in Appendix A and in the footnotes to Appendix B. The details of these calorimeters are summarized in Table III. Five of the calorimeters were commercially available designs, primarily of the Gardon type, and seven were "in-house" designs. Five of the calorimeters were hemispherically shaped and seven were flat faced. A wide range of total diameter and sensing area diameters was present in the facility calorimeters. The calorimeter sensing areas were constructed of four different metals.

Table III
FACILITY CALORIMETER DESCRIPTION

FACILITY	CALORIMETER TYPE	CALORIMETER SHAPE	SURFACE MATERIAL	TOTAL DIAMETER (in.)	SENSING DIAMETER (in.)
SRI	Transient slug	Flat face	Nickel plate on copper	1.25	0,625
Ames Research Center- GDB	Transient slug	Hemisphere	Teflon coating on copper	0.75	0,313
Ames Research Center- MPDB	Transient slug	Hemisphere	Gold plate on copper	1.25	0.375
Langley Research Center- AMPD	Transient, thin-shell multiple TC's	Hemisphere	Stainless steel	2.0	2.0
Langley Research Center- ESB	Transient, thin-shell multiple TC's	Hemisphere	Stainless steel	1.5	1.5
Manned Spacecraft Center	Hy-Cal, asymptotic	Flat face	Constantan	1.25	0.15
Aerotherm Corp.	Hy-Cal, asymptotic	Flat face	Constantan	1.5	0.20
AVCO Corp.	Transient, long slug	Flat face	Copper	1.25	0.375
Giannini Scientific Corp.	Steady state	Hemisphere	Copper	0.625	0.625
Martin Co.	Thermogage, asymptotic	Flat face	Constantan	1.25	0.10
Space General Corp.	Hy-Cal, asymptotic	Flat face	Constantan	1.25	0.10
Cornell Aeronautical Lab.	Transient slug	Hemisphere with conical skirt	Copper	0.6	0.090

As pointed out earlier, each facility conducted heat flux traverses for each tunnel condition; the results are reported in Sec. IV-A.

(1) Transient Calorimeters. The majority of the transient calorimeters used during this study could be categorized as "medium-length" slug calorimeters, i.e., slug length of one-half to one times the slug diameter. These calorimeters were exposed to the plasma stream for a few seconds; the heat flux was determined from the slug temperature rise rate by a relation analogous to equation (1).

The two Langley facilities used a thin-walled, slug-type calorimeter containing a thin, stainless steel hemisphere instrumented with a number of thermocouples. This arrangement permitted determination of the heat flux distribution not only at the stagnation point but also over the hemisphere. The AVCO calorimeter was a specially designed, long-slug calorimeter, in which the thermocouple was mounted in a 1.5-in.-long copper slug, 0.020 in. from the front sensing surface. The temperature rise rate was evaluated with a computer program to calculate the cold wall heat flux.

(2) Steady-State Calorimeters. The steady-state calorimeter used by Giannini was a water-cooled, temperature-rise type. The heat flux was calculated with the relation:

$$\dot{q}_{CW} = (\dot{w}C_p \Delta T)/A \tag{3}$$

where

A 8

w = cooling water flow rate-lb/sec

 $C_p$  = heat capacity of water—Btu/lb°F

 $\Delta T$  = temperature rise of the cooling water—°F

 $A = sensing area - Ft^2$ 

Since the sensing area covered the entire hemisphere, they corrected the average heat flux to center-line stagnation conditions with the special relation  $\dot{q}_{\text{CW}}$  =  $2.1\dot{q}_{\text{AV}}$ .

The majority of the steady-state calorimeters used during this study were of the Gardon or asymptotic types manufactured by either Hy-Cal or Thermogage. The heat flux was determined by measuring the temperature difference between the center and the cooled periphery of a thin constantan disc. A small-diameter copper wire was connected to the center of the disc and the disc periphery was welded to the cooled copper shroud, forming the hot and cold thermocouple junctions. The radial temperature difference on the disc is a function of heat flux, disc thickness, diameter, and thermal properties. Since the last three factors are constant for a given instrument, the heat flux can be calculated from the millivolt difference between the two thermocouple junctions.

#### c. Pressure

Because the Phase I study revealed a good correlation between the SRI uncooled pitot probe and the various facility probes, it was decided not to include an SRI pressure probe in this study. Therefore, all model stagnation values listed in Appendix A were measured with the facility pitot probes and pressure gauges or transducers. The majority of the pitot probes were water-cooled, flat-faced cylinders ranging in size from 0.375 to 0.75 in. in diameter. The stagnation pressure  $P_{\rm t_2}$ , was measured with a wide variety of gauges and transducers, as described in Appendix A. Stagnation pressure traverses of the plasma jet were made at each facility for each tunnel condition; the results are reported in Section IV-A.

The expansion of the jet through the nozzle was controlled at most facilities by bleeding air into the test section or by throttling the vacuum line. Some facilities monitored the jet expansion by matching the test chamber pressure to the nozzle exit pressure, and the remainder of the groups monitored the stream visually.

#### 2. Measurement of Model Response

Measurements of model response were made both during the run and after its completion. These include model temperatures as well as physical changes in the model.

## a. Front Surface and Internal Temperature

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To reduce the scatter of front surface temperature data that was experienced during the first round robin, eleven facilities were supplied with identical, calibrated total radiation pyrometers. They were also sent suggestions for mounting the pyrometer in the tunnel and instructions for the use of the instrument.

In most cases, a facility pyrometer was used in addition to the SRI radiometer; however, Ames-GDB also used a total radiation pyrometer. Most facility instruments were either manual or automatic monochromatic optical pyrometers. All the instruments measured the brightness temperature of the model surface, and the results were reported assuming an emissivity of unity.

The Langley-AMPD group used a photographic pyrometer that viewed the entire model surface and made exposures at frequent intervals. The surface temperatures were then measured from densitometer traces of the developed film. Internal temperatures were measured at eight facilities which had the capability of connecting the model thermocouples to instrument leads in the insertion probe. The output of the thermocouples was then fed into a continuous multichannel recorder.

These model temperatures were received from the facilities in the form of graphs of temperature versus time. Since reproduction of these graphs in their entirety was impractical, sufficient data were taken from them to allow redrawing of the original curves. These data are tabulated in Appendix C.

#### b. Mass and Length Changes

A preliminary check indicated that the model core weights of the low-density materials were not constant under varying ambient conditions. Consequently, a study was made of the equilibrium water content of the five low-density materials at various relative humidities; the results are shown in Fig. 5. As a result of the study, it was decided to equilibrate the model cores to 50 percent relative humidity before and after testing. The length and diameter of all model cores were measured and the cores conditioned for 24 hours at 50 percent relative humidity and 70-75°F before weighing on an analytical balance. The model was then assembled, reweighed, and its total length determined with a dial micrometer.

The facility determined the model recession and the total model weight loss after completion of the test. The models were returned to the Institute, and the total model weight loss and front surface recession were again measured. Model base plates were removed and the recession of the front surface of the core rechecked. The model core was pressed out of the shroud, reconditioned as described above, and the weight loss of the core determined. The core char cap was removed and the substrate scraped back to the start of the pyrolysis zone. The cores were reweighed and measured so that char weight, thickness, and density could be calculated.

The measurements made at SRI on the models are listed in the last five columns of the tables in Appendix B. The weights listed in the tables are for the 0.625-in.-diameter  $(0.00213~{\rm ft^2~cross}$ -sectioned area) cores, except where noted. Mass loss rates were determined for each material and each tunnel condition and are listed in Appendix D with other derived information. For cases when two models with varying run times were tested, the mass loss rate was calculated as the slope between the two data points. When three or more samples were run, the mass loss rate was determined from the slope of the best straight line through the data. In a very few instances the mass loss rate was determined from a single run, and for these cases the slope was assumed to pass through zero.

## c. Char Density

The variation in char density from the front surface to the virgin-material interface was measured on a few samples using an X-ray measurement technique that was developed at SRI. A 0.5-in.-wide  $\times$  0.1-in.-thick sample including the char was cut along the center line from the front to the back of the ablated model core. The char layer was then scanned from the front surface to the virgin material with a  $0.250 \times 0.003$ -in.-thick X-ray beam normal to the original model core axis. Attenuation of the X-ray beam indicated the char density profile.

#### 3. Test Procedure

The tunnel operating variables such as power and gas flow rate were established by trial and error at each facility to meet the tunnel condition requested by the Institute. The facility was allowed to match either the requested enthalpy and stagnation pressure or a specified heat flux and pressure.

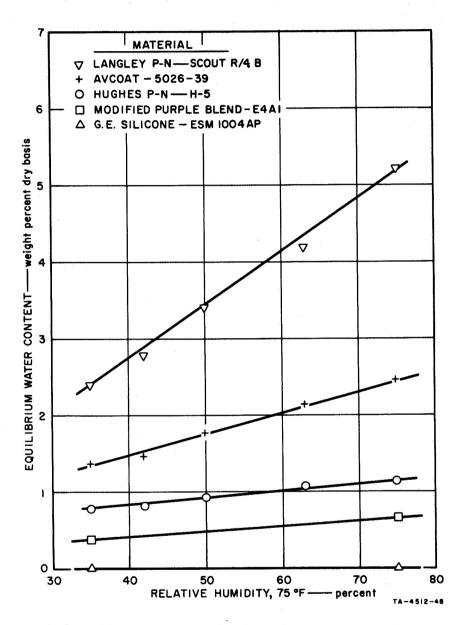


FIG. 5 EFFECT OF HUMIDITY ON WEIGHT OF MATERIALS

The sequence followed by the facility in measuring the requested tunnel variables was largely dictated by the number of instruments that could be sequentially inserted into the plasma stream during a single run. We have termed this the "tunnel insertion capability," and it refers to the number of model supports in the test chamber. Tunnels with four supports could make all requested measurements during a single start-up; facilities with fewer supports required progressively more runs to obtain the required information. The relative reproducibility of a facility's results is, of course, dependent on the run-to-run variation in tunnel conditions compared to the variations during a single-run.

Table IV is a summary of the operating sequence followed at each facility for calibrating the tunnel conditions and testing the ablation models.

Table IV
SEOUENTIAL ORDER OF TEST MEASUREMENTS

			DATA DETERMINED DURING SAME RUN			
FACILITY	REFERENCE (APPENDIX TABLE NUMBER)	NUMBER OF INSERTIONS	<sup>†</sup> t  qSRI  qFAC  P  t 2	<sup>m</sup> t <sup>q</sup> SRI P <sub>t</sub>	<sup>†</sup> t <sup>q</sup> FAC P <sub>t</sub> 2	<sup>q</sup> SRI <sup>q</sup> FAC P <sub>t</sub> 2
Ames-GDB	B-1	8	М		_ :	С
Ames-MPDB	B-2	5	М			С
Langley-AMPD	B-3	2		M		
Langley-ESB	B-4	2	I.			
Manned Spacecraft Center	B-5	2			M	С
Aerotherm Corp.	B-7	5	М			С
AVCO Corp.	B-8	1	ΙΙ			
Giannini Scientific Corp.	B-9	-3			M	С
Martin Co.	B-10	5	M			
Space General Corp.	B-11	4			М	С
Cornell Aeronautical Lab.	B-12	1	11			

M - Model runs.

C - Calibration runs.

I -  $\dot{q}_{
m SRI}$ ,  ${
m P}_{
m t_9}$  estimated and reported from calibration runs.

II -  $\dot{q}_{FAC}$ ,  $\dot{q}_{SRI}$ ,  $P_{t_2}$  estimated and reported from calibration runs.

#### IV EXPERIMENTAL RESULTS

The results of the experimental program covered three broad areas: test environment, high-density materials, and low-density materials, these are covered in the following sections. The test environment is not only measured, but the various techniques and instruments for determining its parameters are cross-compared. The ablation behavior of the high-density materials is described and correlated with the results from the Phase I program. The ablation behavior of the low-density materials is described in more detail, and correlations for these results are suggested.

#### A. Evaluation of Test Conditions

The matrix of test conditions for the second round robin was designed by selecting three test conditions for each participating group that would utilize the full range capability of the facility and at the same time provided the widest distribution of test conditions for all facilities. The distribution of test conditions used in the Phase II round robin is shown in Fig. 6. Also shown in the figure is the envelope of the conditions for the Phase I round robin.

Since the ablation of Teflon and high-density phenolic-nylon had been investigated during the first round robin in the low and medium pressure range, testing of these materials was restricted primarily to the 0.3 to 30 atm stagnation pressure range. The five new low-density materials had been designed for low pressure applications and were therefore tested primarily in the 0.004 to 0.7 atm pressure range.

During the first round robin, the model stagnation pressure measured with an SRI pitot probe of the same geometry as the ablation models was compared to the stagnation pressure measured with the facility pitot probe. The results were in excellent agreement and therefore this comparison was not in the Phase II round robin. Instead a stagnation pressure and heating rate traverse of the plasma streams was substituted.

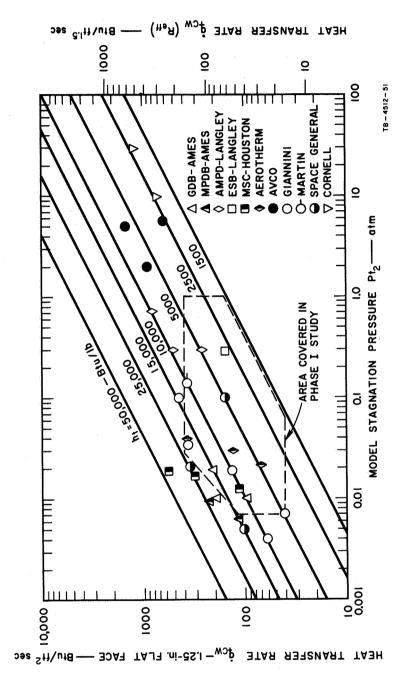


FIG. 6 TEST CONDITIONS FOR PHASE II ROUND ROBIN

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## 1. Plasma Stream Uniformity

The results of the plasma stream traverse of heating rate and model stagnation pressure at each facility are shown in Figs. 7 through 16. These plots were prepared by normalizing the local measured heating rates at various distances from the nozzle center line in terms of the measured heating rate at the center-line position. The same procedure was followed for the model stagnation pressures.

A 1.25-in.-diameter ablation model, drawn to the same scale as the nozzle exit diameter, is shown at the top of each plot to indicate the stream uniformity in the area of the model and core. A scale sketch of the calorimeter, showing its shape, total diameter, and sensing diameter, is also included at the top of each plot.

The nonuniformity of the plasma stream can result from a variety of causes such as heat losses to the nozzle wall, nozzle expansion characteristics, pressure mismatch between the nozzle exit and the test chamber, method used to stabilize the arc, and the position of the measuring instrument. It is impossible to generalize on the causes for the stream nonuniformities shown in Figs. 7-16. One may only state that these were the measured heating rates and pressures for a particular apparatus, tunnel operating condition, and model geometry.

Actually, for this particular series of tests, the plasma streams were apparently quite uniform in the center-line area where the model cores were located. An average of all the participating groups indicated that the heat flux at the model core outer diameter (0.625 in.) was 99 percent of the center-line heat flux. The stagnation pressure at the same point was 97 percent of the center-line value. At the model outer, or shroud diameter (1.25 in.), location the average heat fluxes were 89 percent of center-line values and average pressures were 85 percent of those at the center. The dropoff in heating rate when moving out of the center of the plasma stream is somewhat compensated for with the flat-faced shape which gives a higher heat flux at its periphery. This is indicated in Fig. 17 (Graph A) which shows the heat flux at various positions on the hemispherical and flat-face calorimeters used by the Martin Company in the study of varying model diameter. This plot is in reasonable agreement with the results of Marvin and Sinclair. 9 Graph B of Fig. 17 shows the plasma stream uniformity at the same facility.

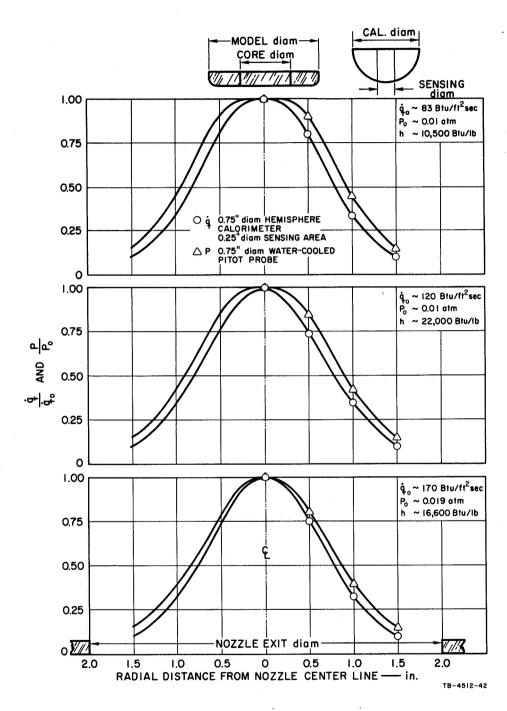


FIG. 7 PLASMA STREAM UNIFORMITY AT GAS DYNAMICS BRANCH, AMES RESEARCH CENTER

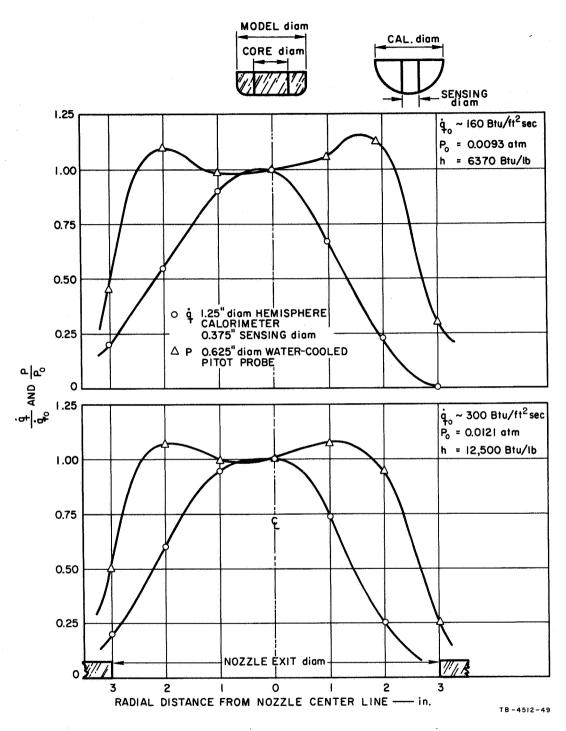


FIG. 8 PLASMA STREAM UNIFORMITY AT MAGNETO PLASMA DYNAMICS BRANCH, AMES RESEARCH CENTER

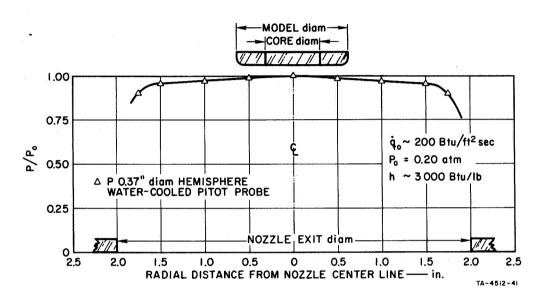


FIG. 9 PLASMA STREAM UNIFORMITY AT ENTRY STRUCTURES BRANCH, LANGLEY RESEARCH CENTER

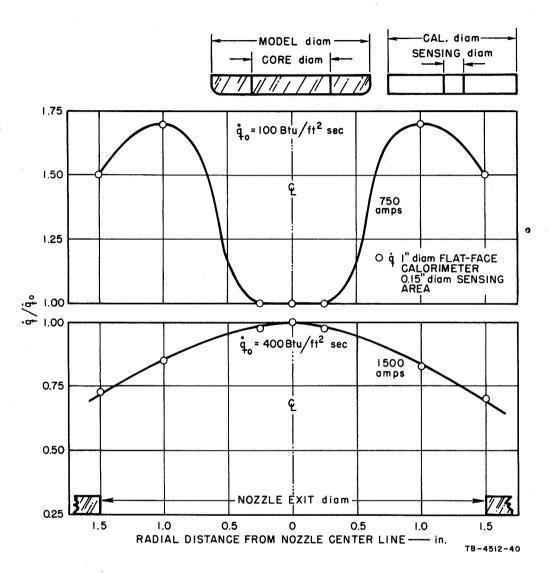


FIG. 10 PLASMA STREAM UNIFORMITY AT MANNED SPACECRAFT CENTER (Subsonic Facility)

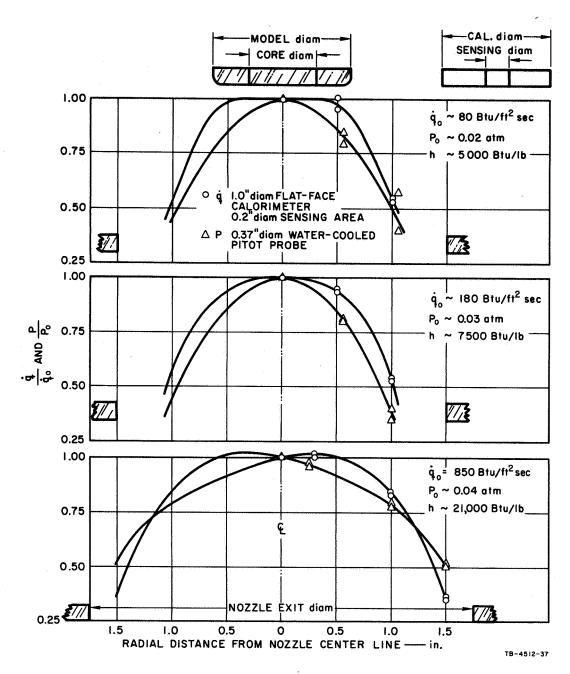


FIG. 11 PLASMA STREAM UNIFORMITY AT AEROTHERM CORPORATION

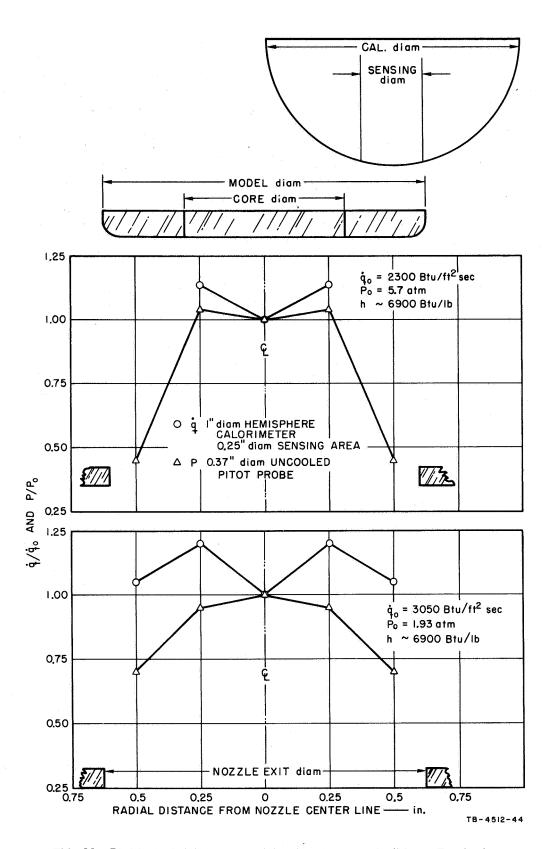


FIG. 12 PLASMA STREAM UNIFORMITY AT AVCO (10-Mw Facility)

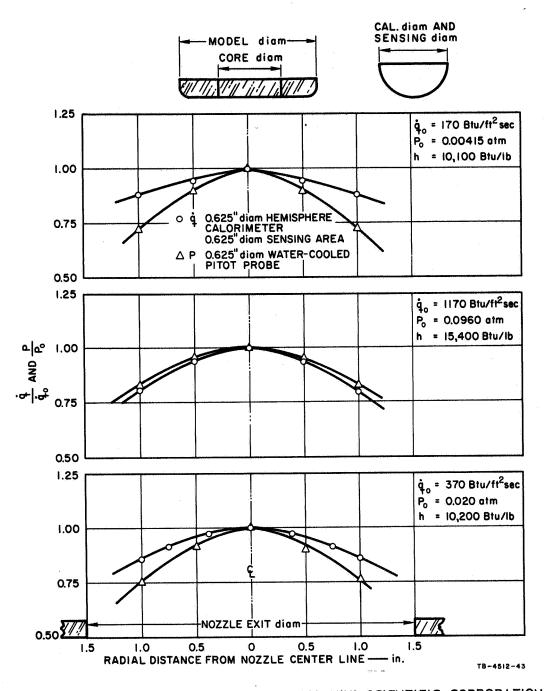


FIG. 13 PLASMA STREAM UNIFORMITY AT GIANNINI SCIENTIFIC CORPORATION

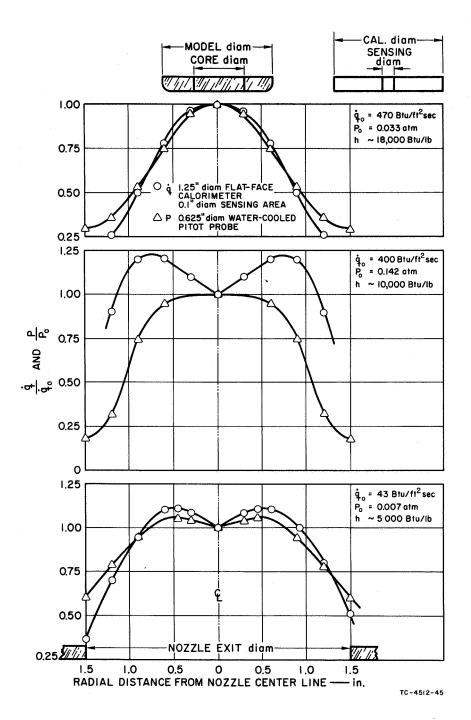


FIG. 14 PLASMA STREAM UNIFORMITY AT MARTIN COMPANY

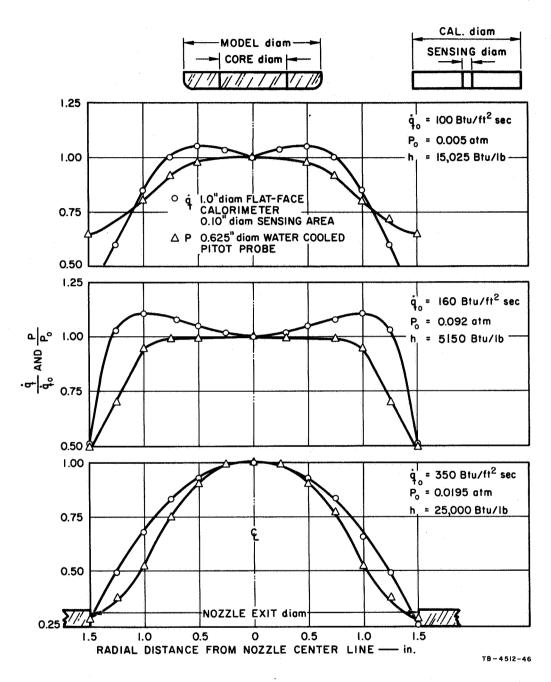
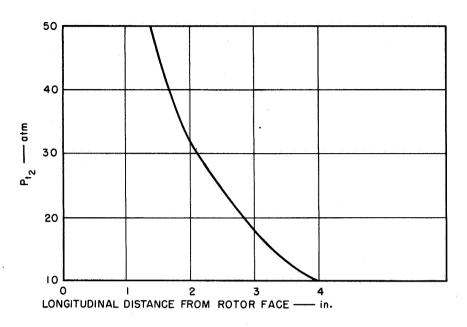


FIG. 15 PLASMA STREAM UNIFORMITY AT SPACE GENERAL CORPORATION



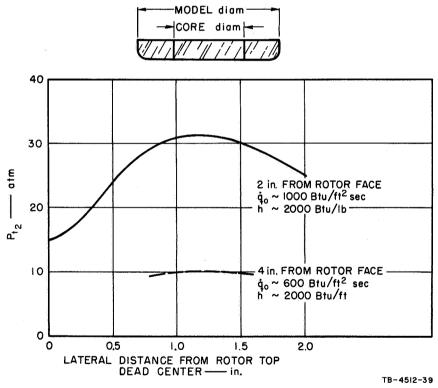
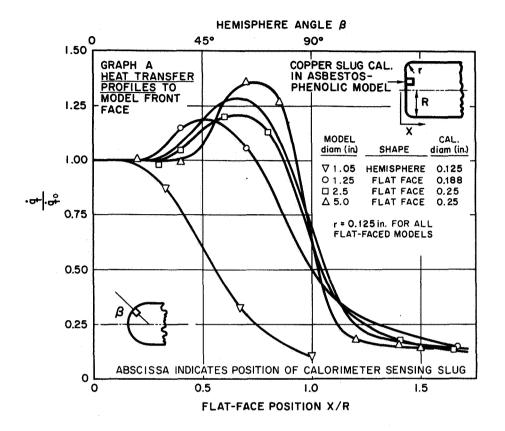


FIG. 16 PLASMA STREAM UNIFORMITY AT CORNELL AERONAUTICAL LABORATORY



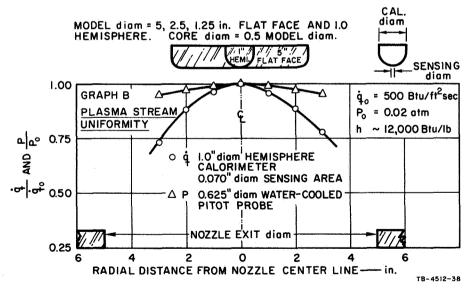


FIG. 17 HEAT TRANSFER PROFILE ACROSS MODEL, AND PLASMA STREAM UNIFORMITY AT THE MARTIN COMPANY

The chemical nonuniformity of the plasma streams was not studied during this work, but the Giannini group, which has conducted such studies, reported that a varying oxygen-to-nitrogen ratio can exist across the stream depending on where and how the oxygen enters the stream. The oxygen level of the test stream will of course have a marked effect on the material ablation rate.

Probably the greatest significance of the plasma stream traverse is that the average stream enthalpy measured by an energy balance does not represent the center-line enthalpy where the model is being tested.

#### 2. Stagnation Point Heating Rate

As stated previously, the cold wall heat flux was measured at most facilities with both a facility calorimeter and the SRI calorimeter. The calorimeter designs differed both from facility to facility and from the SRI design. The main differences in calorimeter designs were shape, total diameter, sensing diameter, and the surface material of the sensing area.

## a. Effect of Calorimeter Design

Ideally, the calorimeter used to measure cold wall heat flux should have the same shape and dimensions as the ablation models being tested. Unfortunately each set of ablation models may differ, and the usual practice is to recalculate the measured calorimeter heat flux to conform to the model shape and size. Selection of a calorimeter is further complicated because for a task such as a stream traverse it might be desirable to have a small diameter hemispherical shape, whereas for model testing, and to reduce the surface catalytic effect, it would be desirable to have a larger diameter calorimeter.

(1) Shape and Diameter Corrections. The shape and diameter of a calorimeter determine the velocity gradients over its surface and thus the heat flux to the surface. It is generally accepted that under supersonic conditions the heat flux to different sized calorimeters with the same shape will vary inversely with the square root of the calorimeter radius, R, or diameter, D, according to the following relation:

$$\dot{q}_1/\dot{q}_2 = (R_2/R_1)^{0.5} = (D_2/D_1)^{0.5}$$
 (4)

where the subscripts designate two different calorimeters.

The above relationship was used to correct the facility heat flux data when the facility calorimeter was flat faced and had a different diameter than the SRI model.

Heat transfer relationships such as the one proposed by Fay-Riddell  $^7$  are based on the heat flux to a hemispherical shape. Thus, the effective radius,  $R_{\rm eff}$ , equals the hemispherical radius. Heat transfer to other shapes may be expressed as some fraction of the heat flux to an equal radius but hemispherical body. Equivalently a correction may be made to the actual radius to give the  $R_{\rm eff}$ .

At the completion of the first round robin, the facility heat flux data for hemispherical calorimeters were compared to the SRI flatfaced calorimeter and were found to effectively follow the relations:

$$\dot{q}_{FF} = 0.55 \dot{q}_{H}$$

$$R_{eff} = 3.3 R_{FF}$$

These results agreed well with the data of Stoney and Markly 10 and were used to adjust facility hemispherical calorimeter results to the SRI shape.

In the Phase II round robin most facility calorimeters were flat faced and required only diameter corrections. In addition, the two Ames facilities corrected their hemispherical calorimeter results with factors that they had previously established experimentally. The few remaining facility hemispherical calorimeters were corrected using the same factors that were used in the Phase I round robin.

uation, the plasma arc has been the most versatile test device developed for reproducing free flight heating conditions. There are, however, obvious differences between ground test conditions and free flight conditions. In free flight the air preceding the vehicles shock wave is at rest and at chemical equilibrium, except at extreme altitudes. In arc plasma testing, the model is stationary, and the test gas preceding the model shock wave has been preheated to a very high temperature level and then expanded to low pressure to simulate free flight conditions. The high gas tempererature, together with this expansion through a supersonic nozzle to obtain high velocity, can give a plasma stream that is not in chemical

equilibrium. This is particularly true with large expansion ratios. Recombination of the dissociated gas molecules behind the model shock wave thereby influences the heat flux to the calorimeter or model. The recombination mechanism has not been fully quantified but is known to be a function of the atomic concentration and gas density in the boundary layer, the wall temperature, model geometry, and wall catalytic activity. 11

The amount of heat released by catalytic recombination becomes important when the heat flux measurements are used to calculate the enthalpy in the center of the nonequilibrium plasma stream at the model location. Heat transfer relationships such as Fay-Riddell assume an infinitely catalytic surface and complete recovery of all energy. Metal calorimeter surfaces have varying finite catalytic reaction rate constants, and the measured heat flux will be less than that for infinitely catalytic surfaces. Further, for a given surface material, the ratio of measured heat flux to the heat flux at a fully catalytic surface will increase with increasing stream density and calorimeter diameter, and the ratio will decrease with increasing enthalpy and wall temperature.

The Gas Dynamics Branch of Ames Research Center conducted a study of the effect, on the measured heat flux, of calorimeter surface catalytic activity and some of the other variables noted above. During the study, the Ames copper-surface calorimeter and the SRI nickel-surface calorimeter were exposed to a range of enthalpies (8000 Btu/lb and greater) at two stagnation pressures. Identical calorimeters that had been sprayed with a thin coating of Teflon were also exposed to the same conditions. These tests were performed at a relatively high expansion ratio.

The results of the study are shown in Fig. 18, in which the ratio of measured heat flux,  $\dot{q}_{\text{meas}}$ , to the heat flux for an infinitely catalytic surface,  $\dot{q}_{k}\sim_{\infty}$ , is plotted versus the total stream enthalpy as determined by the modified sonic flow method (see footnote 2, Appendix B-1). The value of  $\dot{q}_{k}\sim_{\infty}$  was calculated using the Fay-Riddell relation and the total stream enthalpy reported by Ames with their experimental relation of  $R_{eff}$  = 2.91  $R_{FF}$ . It should be noted that the higher pressure runs shown in the figure were made by entering part of the gas at the plenum location, thereby changing the equilibrium condition and giving an accentuated effect of stagnation pressure on the heat flux ratio,  $\dot{q}_{\text{meas}}/\dot{q}_{k}\sim_{\infty}$ .

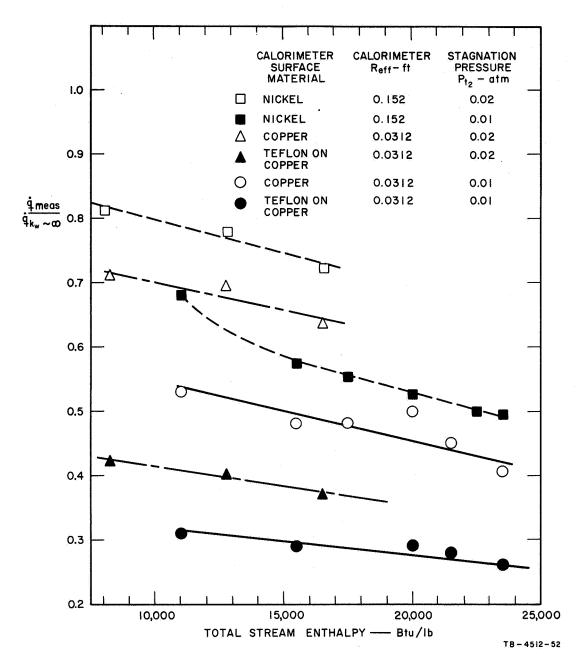


FIG. 18 EFFECT OF CALORIMETER DESIGN ON HEAT FLUX MEASUREMENTS IN A NONEQUILIBRIUM STREAM

The data were also used by Ames to estimate the catalytic reaction rate constant,  $k_{\rm w}$ , for each surface. The  $k_{\rm w}$  was found to be 500 to 700 cm/sec for copper, 300 cm/sec for nickle, and to be much lower for Teflon; the values for the metals agree with those of Goulard.  $^{\rm 11}$ 

The results indicate the importance of calorimeter surface catalytic activity, calorimeter geometry, and stream conditions on the measured heat flux. No simple correlation of all variables has been developed to date; however, studies in this field are now under way. The general conclusions at this time are that, when the arc generator-nozzle system tends to lead to a nonequilibrium plasma, the calorimeter surface should be a clean metal having high catalytic activity such as silver, copper, or nickel. Further, the calorimeter diameter should be as large as is practicable. Finally, an indication of the stream nonequilibrium condition can be obtained by comparing the measured flux to a catalytic metal surface to the heat flux measured with an identical calorimeter that has been sprayed with a thin coating of Teflon to give a noncatalytic surface. This, of course, is not possible at high heat fluxes where the Teflon would sublime rapidly.

#### b. Comparison of Results

Using the correction techniques discussed above, the measured facility heat flux data were adjusted to the 1.25-in.-diameter SRI model shape. The facility and SRI calorimeter results are compared in Fig. 19, the standard deviation was found to be 13 percent. These results are a slight improvement over the first round-robin data, which showed a standard deviation of 16 percent.

The Cornell data gave the greatest deviation, with the SRI calorimeter reading about 1.6 times the Cornell value. Cornell reported
that they have previously experienced even higher readings at high pressure
conditions with calorimeters which are similar to the SRI design but which
have an air gap surrounding the slug. Apparently the high pressure gases
flow through this air gap and can preferentially heat the thermocouple
junction, thereby giving a high temperature rise rate. The problem may
be further accentuated by the nonsymmetry of the stream at this facility.
Cornell solves this problem by filling a short section of the air gap
with a refractory cement. However, this solution is not completely
satisfactory, since increased contact between the slug and the shroud

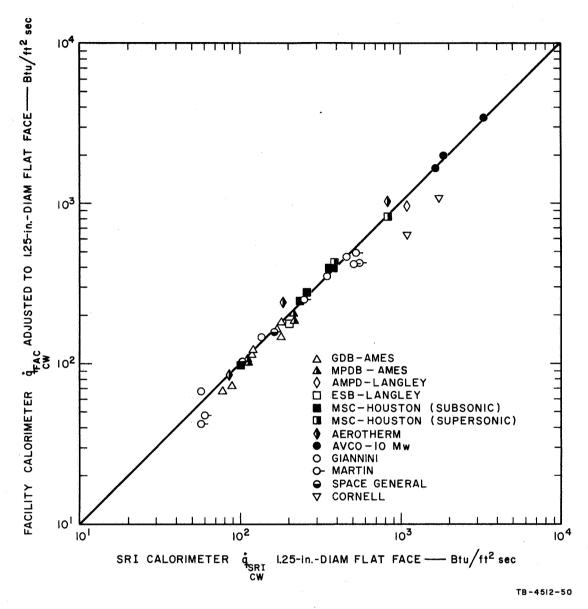


FIG. 19 COMPARISON OF FACILITY AND SRI CALORIMETERS

can lower the measured heat flux. Perhaps the best solution is to seal the calorimeter for pressure conditions considerably above 1.0 atm and to isolate the slug with an air gap for lower pressures.

# 3. Prediction of Stagnation Point Enthalpy

The stagnation point enthalpy in the vicinity of the model can be calculated from the model stagnation pressure and the cold wall heat flux values. The following form of the Fay-Riddell equation was used to calculate the stagnation enthalpy potential,  $\triangle h$ , for each facility from the  $\dot{q}_{\text{SRI}}$  and  $P_{\text{t}_2}$  data:

$$\Delta h_{calc} = S_{R}\dot{q}(R_{eff})^{0.5}/(P_{t_2})^{0.5}$$
 (5)

where  $S_{\rm R}$  is 24 as shown in Appendix E, Sec. A.

The above relation assumes air at chemical and thermodynamic equilibrium as the test gas with an invariant Lewis number equal to 1 and a Prandtl number equal to 0.72. The value of  $R_{\rm eff}$  was taken as 0.172 ft based on the 1.25-in.-diameter flat-faced shape and  $R_{\rm eff}$  = 3.3  $R_{\rm FF}$ . The resulting values of enthalpy have been tabulated in Appendix D and are compared in Fig. 20 to the reported enthalpy as measured by the technique preferred by the facility.

Figure 20 shows a prepondence of data above the correlation line indicating that the center-line enthalpy in the area of the model was probably higher than the average measured enthalpy reported by some of the facilities. Facilities such as AMPD-Langley and ESB-Langley that prefer the heat flux method for measuring enthalpy gave a good correlation as would be expected. The differences for these two facilities result from different calorimeters and calculation methods. Giannini's and Space General's measured enthalpies agreed well with the calculated values. These two facilities also reported quite uniform stream traverses, indicating that the center-line enthalpy is probably close to the average enthalpy by the energy balance method.

Personnel at GBD-Ames feel that for moderate to high pressure nonuniform streams the heat flux enthalpy is preferable to other methods of measuring average enthalpy, but they also believe that this enthalpy can

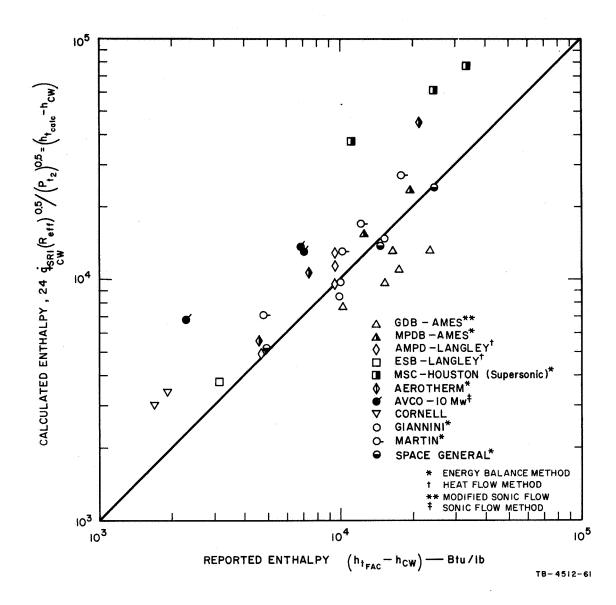


FIG. 20 COMPARISON OF CALCULATED AND REPORTED ENTHALPIES

be severely in error on the low side, as shown by their data in Fig. 20, when used for low pressure, nonequilibrium streams. The problem results from the necessity of using calorimetric surfaces with less than infinite catalytic activity, as discussed in the previous section. The group at GDB-Ames has therefore developed a modified sonic flow method which allows them to calculate the stream temperature and resulting enthalpy in the model area. The enthalpies calculated by the modified sonic flow method can be 1.5 times higher than the heat flux method as seen in Fig. 20.

It appears that although the stream enthalpy is undoubtedly the most important variable in material ablation studies, it is also the most difficult to measure accurately.

## B. Performance of High-Density Ablation Materials

In order that the effects of extended test conditions, and especially of varying dimensions, could be evaluated, a dimensional analysis of ablation variables was undertaken. This analysis and its use to correlate ablation data for the high-density Teflon and phenolic-nylon materials used in the Phase I study is covered in Appendix E.

In the present round-robin program (Phase II), two groups of experiments were performed to extend the variables studied in Phase I. The first was the use of high-density Teflon and phenolic-nylon models of the standard size (same as in Phase I) but exposed to considerably higher stagnation pressures. The second group involved Teflon and low-density phenolic-nylon models having effective radii varying from four times as large to about four times as small as the standard models. The results of these experiments and how they fit the correlations are discussed below.

#### 1. High Stagnation Pressure Environments

The lack of fit of the data obtained by Walberg at high stagnation pressures, as shown in Fig. E-2 of Appendix E, suggests that the correlation does not properly take into account such environmental conditions. This was confirmed, at least for the high-density phenolic-nylon models, when the data from the Phase II round robin experiments at high stagnation pressures were checked against that figure and were shown to have the same displacement. Reconsideration of these relations was therefore in order, and, at this point, it was decided to use the separate correlations

represented by Equations (E-18A) and (E-18B) in Appendix E since there appeared to be a difference in the behavior of Teflon and phenolic-nylon at high stagnation pressures.

## a. Behavior of Teflon

The approach tried was to separate the pressure term from the rest of the relation. Rearrangement of Equation (E-18A) in this manner leads to

$$\dot{m}_{t}(R_{eff})^{0.18}/(\dot{q}_{CW})^{0.57} = 0.0044(P_{t_{2}})^{0.25}$$
(6)

A plot of the left-hand side of this relation against stagnation pressure,  $P_{t_2}$ , on logarithmic coordinates should show the indicated slope of 0.25 and intercept of 0.0044 for the right-hand term. It should be remembered that Equation (6) is based on the Phase I round-robin results. When such a plot was made with the Teflon literature data given in Table E-1 (Appendix E), the best correlation line showed a slightly different slope and intercept. These data were therefore correlated by the regression program in terms of relation (E-16B), namely,

$$\dot{m}_{t}(R_{eff}) = b(\dot{q}_{CW}R_{eff})^{n}(P_{t_{2}}R_{eff})^{m}$$
 (7)

The computer gave the following values for the constants using the Teflon literature data:

The multiple correlation coefficient is maximized by the regression analysis. The closer this coefficient is to unity, the more significant is the correlation.

Averaging these constants with the Phase I round-robin constants of 0.0044, 0.57, and 0.25 and giving the latter ones slightly more credence, since they represent more data points, the corrected form of Equation (6) would be

$$\dot{m}_{t}(R_{eff})^{0.21/(\dot{q}_{CW})^{0.55}} = 0.0046(P_{t_{2}})^{0.24}$$
 (8)

This corrected form is plotted in Fig. 21 and is based not only on the Phase I round-robin results and literature data, but also on the Phase II round-robin high stagnation pressure runs performed at AVCO and Cornell. As can be seen in the figure there appears to be no effect of the stagnation pressure on the correlation, at least to pressures of 33 atm. This is not unexpected since Teflon ablates by sublimation and thus should be little affected by mechanical forces.

# b. Behavior of High-Density Phenolic-Nylon

A similar approach was used in evaluating the phenolic-nylon data. The rearranged Equation (E-18B) gave

$$\dot{m}_{t}(R_{eff})^{0.32}/(\dot{q}_{CW})^{0.55} = 0.0010(P_{t_{2}})^{0.13}$$
 (9)

The effect of the literature data on the constants was not checked, since there was insufficient information for use in a regression analysis. A plot of Equation (9) is shown in Fig. 22 and is based on the Phase I round-robin results and literature data, as well as on the Phase II round-robin, high stagnation pressure runs at Cornell and AVCO.

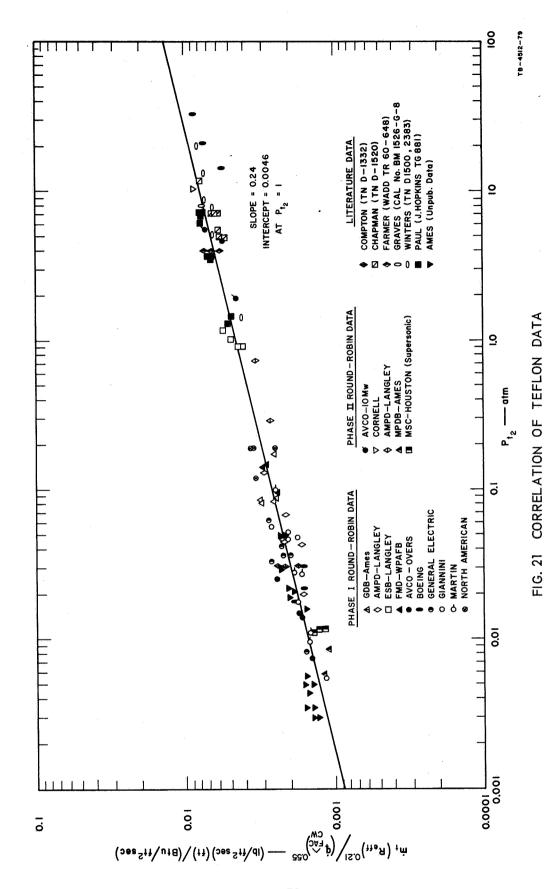
This plot shows that at high stagnation pressure the phenolic-nylon models exhibit higher mass loss rates than would be predicted by the Phase I round-robin correlation. However, these higher rate data do fit a correlation line of steeper slope, with a transition between the two correlations occurring at a stagnation pressure of about 2.7 atm. Thus, the correlations for high-density phenolic-nylon might be expressed for  $P_{\rm t_2} \leq 2.7$  atm as

$$\dot{m}_{t} = 0.0010(R_{eff})^{-0.32}(\dot{q}_{CW})^{0.55}(P_{t_{2}})^{0.13}$$
 (10)

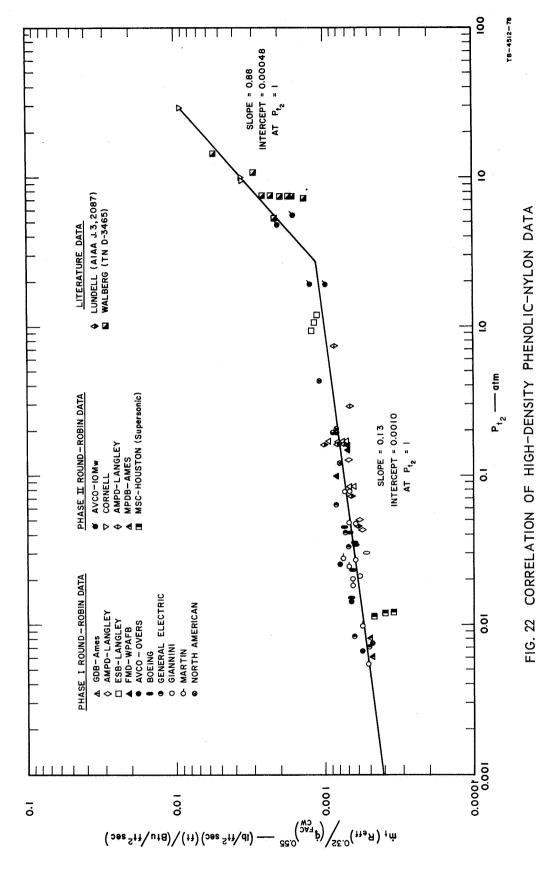
and for  $P_{t_2} \ge 2.7$  atm (and at least up to 29 atm) as

$$\dot{m}_{t} = 0.00048 (R_{eff})^{-0.32} (\dot{q}_{SRI})^{0.55} (P_{t_{2}})^{0.13+0.75}$$
 (11)

This variation in behavior might have been predicted because phenolic-nylon ablates by a charring mechanism which is particularly sensitive to the mechanical stresses brought on by high stagnation pressures.



\$5



This is verified by the fact that the phenolic-nylon models from these tests showed almost no char after exposure to the high stagnation pressure environments, and thus had reduced thermal protection.

The use of the double exponent on the pressure term in Equation (11) is based on this fact of char failure. The dimensionless form of Equation (9) is given in expanded form by Equation (E-10) of This form would imply that the exponent on the effective Appendix E. radius would have to increase to a positive number with this large an exponent on the stagnation pressure. There is no evidence for this behavior and it appears more logical that one additional dimensionless group should be added to Equation (E-9) based on a new variable ,  $\tau$ , the failure stress of the char. Normal units for this variable are pound force per square foot. Converted to the pound-foot-second system, it becomes  $au_{\mathrm{c}}$  with converted units of lb/ft sec2. This has the same converted units as stagnation pressure,  $P_{t}$ . Section A of Appendix E shows the converted form of Thus, the simplest form of the new dimensionless group this as P<sub>t</sub>F<sub>p</sub>g<sub>c</sub> would be

$$\pi_{A} = P_{t_2} F_{p} / \tau \tag{12}$$

Equation (E-9) then becomes

$$\pi_{\rm m} = a_0 \pi_{\rm q}^{\rm n} \pi_{\rm p}^{\rm m} \pi_{\rm A}^{\rm r} \tag{13}$$

Expanded into dimensional form, this equation becomes

$$\dot{m}_{t} = b(F_{p}/\tau)^{r}(R_{eff})^{n+m-1}(\dot{q}_{CW})^{n}(P_{t_{2}})^{m+r}$$
 (14)

where b is as defined in Equation (E-17) and is equal to 0.0010 as shown in Equation (10).

This is identical to Equation (11) with r equal to 0.75 and the value of the constant there can be compared with b from Equation (10) to determine the value of  $\tau$  from Equation (14). Hence,  $\tau$  is found to be 5610 lb force/ft<sup>2</sup> or 2.65 atm.

#### 2. Models with Variable Radii

#### a. Phase II Round-Robin Data

The Phase II studies with models of varying radii involved both heat transfer and mass loss measurements, thus permitting a check of the

radius effect in both cases. It should be remembered, however, that only a limited amount of effort was put into these studies.

(1) <u>Heat Transfer Rate</u>. It is generally accepted that under supersonic conditions the heat flux to different sized calorimeters with the same shape will vary inversely with the square root of the calorimeter radius. Further, the dimensions of different shaped models can be expressed in terms of their effective radii, and, as pointed out in Equation (13) of the Phase I report, <sup>1</sup> this is

$$R_{eff} = R_{H} = 3.3 R_{FF} \tag{15}$$

Thus, the heat flux will decrease with increasing calorimeter size according to the relation

$$\dot{q}_{CW}/(\dot{q}_{CW})_{1.25FF} = [(R_{eff})_{1.25FF}/R_{eff}]^{0.5}$$
 (16)

A logarithmic plot of the heat flux ratio versus the effective radius is shown in Fig. 23.

The calorimeters used by AMPD-Langley and Martin were those that best matched the SRI calorimeters. The AMPD facility used a four-inch shroud for the flat-face model and calorimeter rather than a five-inch shroud; this was done to minimize stream blockage at the diffuser.\*

As can be seen from the plot, a slope of -0.5 fits the data well, and the intercept at an effective radius of unity is equal to the square root of the effective radius of a 1.25-inch, flat-face calorimeter, namely, 0.415. Except for the Martin high point at the five-inch, flat-face effective radius, where stream blockage may be occurring, and a low point for the one-inch hemispherical calorimeter, the data confirm the inverse square root relation.

(2) <u>Mass Loss Rate</u>. The predicted effect of model radius on Teflon mass loss rate is shown by Equation (E-18A). Rearrangement of this equation in terms of effective radius gives

$$\dot{m}_{t}/(\dot{q}_{CW})^{0.57}(P_{t_{2}})^{0.25} = 0.0044(R_{eff})^{-0.18}$$
 (17)

The core diameter, however, remained at 2.5 inches. Also the corner radius on the AMPD models was made one-fifth of the shroud radius to reduce stream blockage. The corner radius on the Martin models was constant at one-eighth inch regardless of shroud radius so that the ratio varied, as shown in Graph A of Fig. 17.

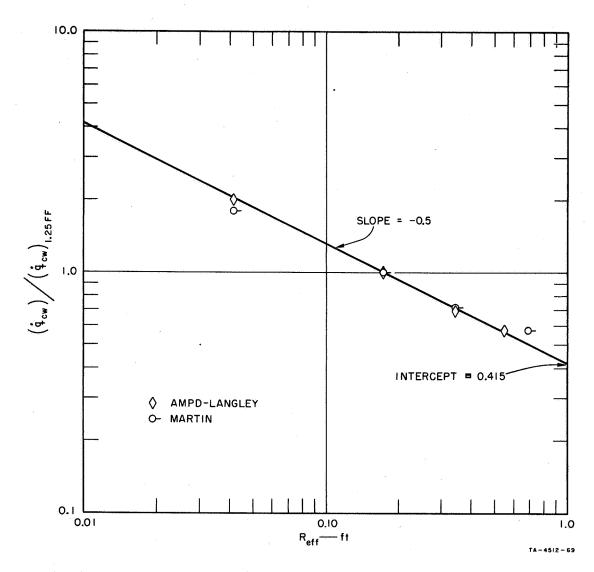


FIG. 23 VARIATION OF HEAT TRANSFER RATE WITH EFFECTIVE RADIUS

Using the data from the Phase II experiments of variable radii Teflon models, the left-hand side of this relation was calculated and plotted against effective radius in the top half of Fig. 24. The slope of the line is -0.18 and the intercept is 0.0044 at  $R_{\rm eff}$  = 1 ft. It appears that the data could be fitted with a line of this slope but at a lower intercept. The reason for this is unknown but must relate to the way in which the experiment was performed and in which the measurements were made, since data obtained at this facility using other standard 1.25-inch flat-face models fit the general correlation well, yet the data obtained with a standard model in this series of experiments are also displaced downward.

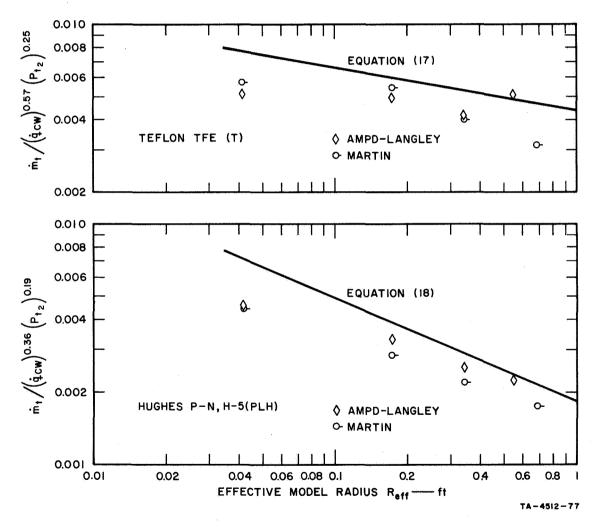


FIG. 24 VARIATION OF MASS LOSS RATE WITH EFFECTIVE RADIUS

One possible explanation relates to the fact that as the model diameters were increased, the core diameters were increased in the same proportions. However, the calorimeters kept the same sensing area and merely increased the shroud diameter. With the plasma coring exhibited by the facilities involved this could provide misleading information about the thermal environment to which the core was exposed.

The same pattern was found with Phase II variable radii data for the Hughes low-density phenolic-nylon models. As will be seen in Sec. IV-C, the mass loss rate correlation, cast into the same form as Equation (17), is

$$\dot{m}_{t}/(\dot{q}_{CW})^{0.36}(P_{t_{2}})^{0.19} = 0.0018(R_{eff})^{-0.45}$$
 (18)

This is plotted in the bottom half of Fig. 24.

## b. Literature Data

The correlation of the literature data from supersonic facilities in Appendix E, Sec. C, covered radii varying from 0.0156 to 0.55 ft and had a very low standard deviation. The effect of radius on mass loss rate, predicted by the dimensionless correlation (see Equation (E-10), or its revised form (E-16A)), has therefore been well verified.

# C. Performance of Low-Density Ablation Materials

More extensive measurements were made on the low-density materials during the ablation experiments. These permitted a more detailed determination of the response of these materials to the test environment. As a result, additional correlations were considered for interpreting these data.

### 1. Ablation Behavior

This section contains information on temperature measurements and physical changes in the low-density materials evaluated during the Phase II round-robin.

### a. Front Surface Temperature

The front surface temperature data from the first round robin showed considerable scatter. To avoid this, the facilities were supplied with identical, calibrated total radiation pyrometers. A description of the pyrometers and calibration technique is given in Sec. III-B. The facilities also received suggestions for mounting the pyrometer in the tunnel and instructions for the use of the instrument.

A comparison of reported front surface temperatures as measured with the SRI-supplied radiation pyrometer and the facility optical pyrometer is presented in Fig. 25. The data shown in this figure are from Appendix B and are the surface temperatures measured on the Langley phenolic-nylon (PLL) material. The same pattern of data was also evident on the other materials that were evaluated.

The measured surface temperatures given in Fig. 25 are "brightness" temperatures, assuming a surface emissivity of unity. Since the actual emissivity is less than one, the true-surface temperatures are higher than those indicated. If an emissivity of 0.8 is assumed for the (PLL) material, the following corrections must be added to the measured values to give the true temperatures:

MEASURED TEMPERATURE (°F)	RADIATION PYROMETER CORRECTION (°F)	OPTICAL PYROMETER CORRECTION AT 0.655μ (°F)
2000	+100	+35
3000	+160	+70
4000	+220	+130

The effect of these corrections is shown by the correlation line labeled  $\epsilon$  = 0.8 in the figure.

Examination of Fig. 25 indicates that some facilities such as GBD-Ames, MSC-Houston (subsonic), Giannini, and Martin had good agreement between the facility and SRI pyrometers when an emissivity of 0.8 - 0.9 was assumed. AMPD-Langley, Aerotherm, and Space General did not display as good an agreement in surface temperature. Part of the lack of agreement seemed to have resulted from the radiometer mounting location and optical path to the radiation pyrometer, as described briefly in Appendix A. When the radiometer was located outside the tunnel, with narrow grazing angles off intervening windows, the attenuation of the optical signal resulted in low surface temperatures of the model. In a few cases the radiometer was mounted with a narrow viewing angle to the model front surface and the model rapidly ablated out of focus.

The group at GDB-Ames has used radiation pyrometers extensively and is aware of the precautions that must be followed in their use. It

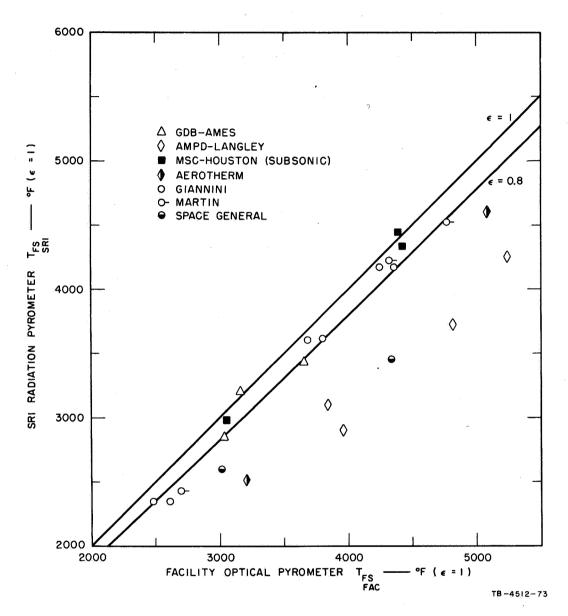


FIG. 25 COMPARISON OF SRI RADIOMETER AND FACILITY OPTICAL PYROMETERS

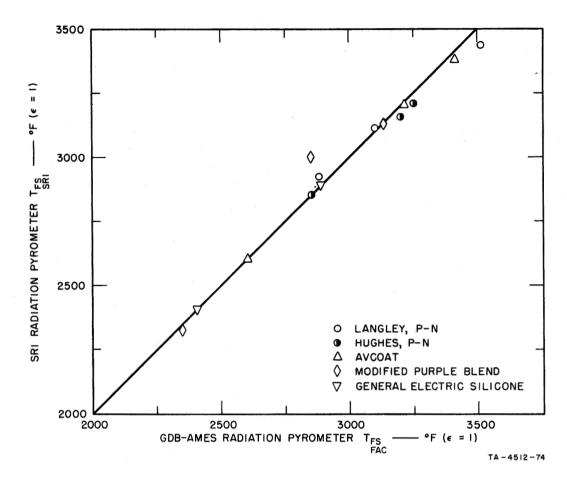


FIG. 26 COMPARISON OF SRI AND GDB-AMES RADIOMETERS

is probably for this reason that they obtained a good correlation between their radiation pyrometer and the SRI radiometer, as indicated in Fig. 26.

Since the radiometers were not calibrated in position on the tunnels, it was decided that the facility pyrometer temperatures were more reliable and were therefore used in all front surface temperature correlations. This points up the general problem of using identical calibrated instruments to cross-correlate facilities. Either their use must be rigidly specified and followed, or they must be further calibrated in position on the tunnel.

### b. Internal Temperature Rise

A plot of the temperature data for the run on model PLL 96 performed at GDB-Ames is given in Fig. 27. This figure shows the temperature rise of the four internal thermocouples and the model front surface

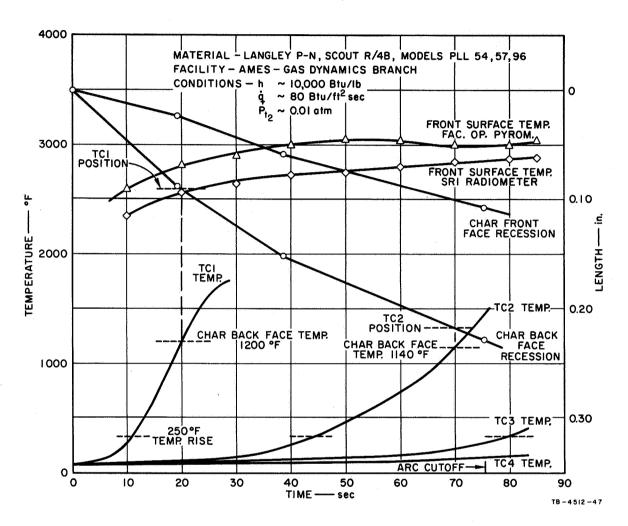


FIG. 27 MODEL TEMPERATURES AND RECESSION DURING ABLATION RUN

temperature as measured with a facility optical pyrometer and the SRI radiometer and is based on the data in Appendix C. A length scale was added to the right-hand ordinate of each temperature graph, and the model recession and char thickness data from Appendix B were plotted for the three varying run times used on models PLL 54, 57, and 96. The initial thermocouple distances were added to the graphs and the time noted when the char-virgin material interface passed each thermocouple position. The temperature of the thermocouple at the above noted time was designated as the char-virgin material interface temperature and is recorded in Appendix C.

Although the data showed considerable scatter, they also followed a particular pattern indicating that the char-virgin material interface temperature increases with increasing mass pyrolysis rates. The scatter probably resulted from the difficulty of measuring the position and temperature of a receding boundary layer and the fact that the temperature gradient in the material is very steep at the char in the decomposition zone. The temperature gradients at the interface ranged upwards to  $35^{\circ}F/0.001$  in. for the high pyrolysis rate condition, indicating the importance of small thermocouple wire diameter and of position measurement.

The instantaneous mass pyrolysis rate was calculated at the time the char interface passed each thermocouple position. It was assumed that all material back to the interface had been pyrolized and consumed. Comparison of the data for the mass pyrolysis rate versus char interface temperature for the Avcoat and the two low-density phenolic-nylon materials indicated that above a minimum temperature of about  $800\,^{\circ}\mathrm{F}$  the pyrolysis rate was approximately a function of the fourth power of the interface temperature.

#### c. Mass Loss Rates

The model core weight losses were determined on all models returned to SRI. The methods are outlined in Sec. III-C. The equivalent mass loss per area, in pounds per square feet, was calculated and plotted against time for each material. Examples of these plots are given in Figs. 28 and 29 for Langley low-density phenolic-nylon and Avcoat material.

The plots show a typical higher initial mass loss rate of charring ablation as the char is established and the front surface temperature increases. This is followed by a period of slightly lower mass loss and the

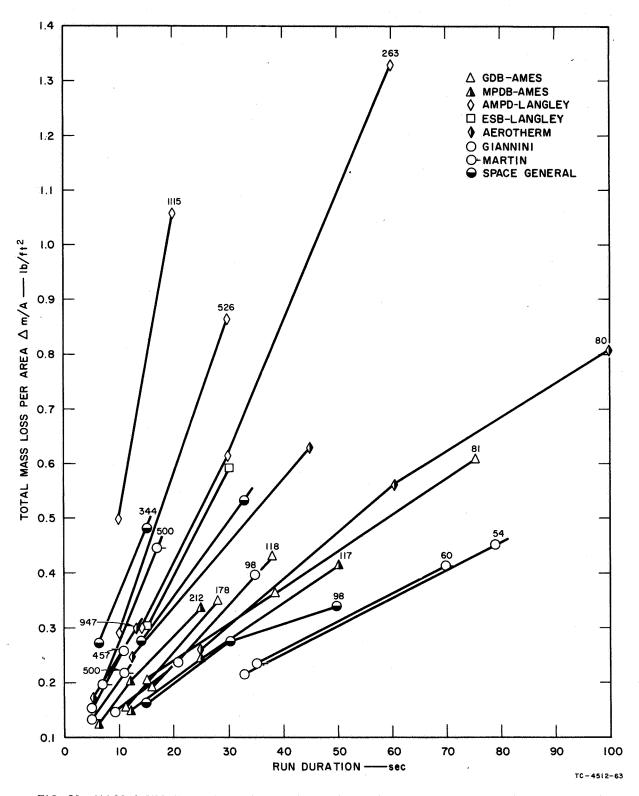


FIG. 28 MASS LOSS OF LANGLEY LOW-DENSITY PHENOLIC-NYLON AS A FUNCTION OF RUN DURATION (Heat Transfer Rate Indicated for Each Facility)

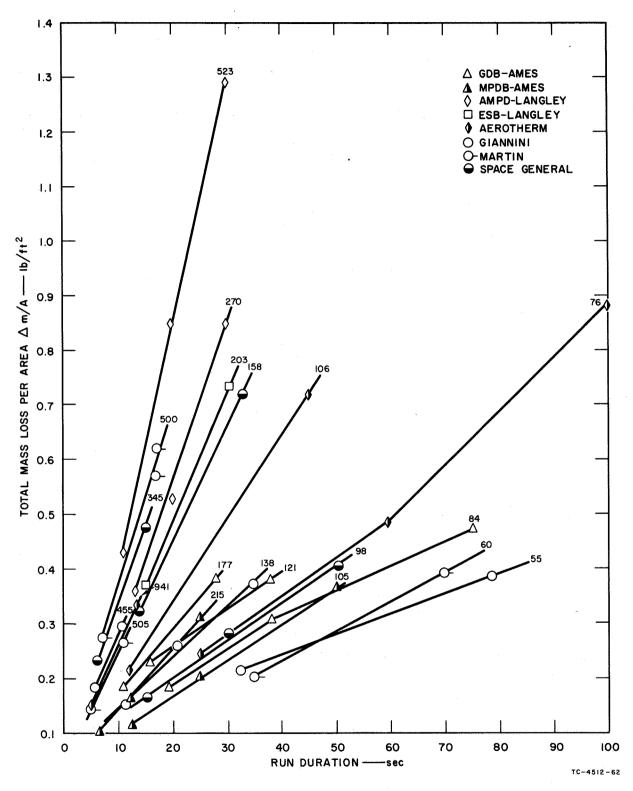


FIG. 29 MASS LOSS OF AVCOAT MATERIAL AS A FUNCTION OF RUN DURATION (Heat Transfer Rate Indicated for Each Facility)

establishment of a quasi steady-state mass loss rate. The induction period ranged from a fraction of a second at the very high fluxes to approximately eight seconds for the lowest fluxes.

As stated in Sec. III-C, when more than two models were tested, the mass loss rate,  $\dot{m}_t$ , was determined from the slope of the best straight line through the data. When two models were tested the mass loss rate was determined from the slope between the two data points, and for the occasional runs involving one model, the slope was assumed to intersect zero. The calculated mass loss rates have been tabulated in Appendix D.

The mass pyrolysis rate,  $\dot{m}_p,$  was also calculated, as follows, for all tests involving two or more models:

$$\dot{m}_{p} = \rho_{VR} [(X_R + X_{CR})_2 - (X_R + X_{CR})_1] / (t_2 - t_1)$$
 (19)

Here,  $\mathbf{X}_{\mathrm{R}}$  and  $\mathbf{X}_{\mathrm{CR}}$  are the recession of the front surface of the char and char thickness, respectively;  $\rho_{\mathrm{VR}}$  is the virgin material density; and the subscripts 2 and 1 denote long and short duration runs, respectively. The mass pyrolysis rate thus represents the rate at which the char-virgin material interface is moving into the model. These results are also tabulated in Appendix D.

The mass pyrolysis rate described above is defined slightly differently than the mass pyrolysis rate reported in the Phase I report, but it is consistent with Lundell's definition.  $^{12}$  The Phase I data can be corrected to equal the Phase II data by adding the  $\dot{m}_{CR}$  listed in Appendix C of the Phase I report  $^1$  to the listed values of  $\dot{m}_p$ .

Although the char removal rate,  $\dot{m}_{CR}$ , is not tabulated in Appendix D for the Phase II data, it was used to determine how closely the ablation of the low-density materials correlates with Scala's predicted regimes for the combustion of graphite. <sup>13</sup> This involved calculation of  $\dot{m}_{CR}/(P_{t_2}/R_{eff})^{0.5}$  and the plotting of this against front surface temperature,  $T_{FS}$ , in degrees Rankine.

The high-density phenolic-nylon data from Phase I agree well with theory in showing a diffusion-rate-controlled plateau above  $3000^{\circ}R$ . On the other hand, the high stagnation pressure data from Phase II for this same material show considerably higher rates. The five low-density materials showed no plateau above  $3000^{\circ}R$ , in fact, varied as much as sixfold without any discernible pattern.

## d. Char Properties

Chars on the low-density phenolic-nylon materials had a cracked appearance on the surface and a columnar structure oriented parallel to the direction of ablation. This could indicate that the pyrolysis gases take preferential paths to the surface. The pyrolysis zone, as indicated by the slight change in color of the virgin material, was very narrow (approximately 0.025 in.) in most models. The char caps had adequate adhesion to remain on the cores during model disassembly but were easily cleaved from the core, with part of the char remaining on the model core. The char remaining on the core was scraped off before making length and weight measurements.

Chars on the Modified Purple Blend and the G.E. silicones generally had two types of appearance, depending on the exposure history of the model. At low heat fluxes they were black, carbonized chars that swelled during short exposure times, followed by slow recession at longer run times. Higher heating rates resulted in a grey, fused inorganicappearing surface with the indication that the material was removed from the model by melting and flowing down the sides. The pyrolysis zone on these materials was very narrow (0.020 in.) and the chars could be completely removed from the cores without scraping.

The Avcoat chars usually had a depression in the honeycomb filler material, with the honeycomb web being slightly raised. There were fused droplets of inorganic material at the model periphery. The char had excellent adhesion to the substrate and required moderate scraping to remove. The pyrolysis zone seemed wider in the Avcoat materials than in the other materials, and there was evidence that the honeycomb web preferentially conducted heat to the substrate.

The char densities were calculated for each model and are tabulated in the last column of each table in Appendix D. Analysis of these data indicates that the ablation process and its effect on char properties and char dimensions is a continually changing balance of many competing processes. External variables affecting char properties and thickness are heat flux, stagnation pressure, and run time  $(\dot{\mathbf{q}}, P_{t_2}, t)$ ; gas test composition is also an external variable, but since all NASA round-robin tests were conducted in air, no statements can be made on its effects. Internal or material factors affecting the charring process are the pyrolysis kinetics of the polymer and the thermal and physical properties of the char.

For a given charring material, the char thickness appears to be largely a function of  $\dot{q}$ ,  $P_{t_2}$ , and t. At low  $\dot{q}$ 's and  $P_{t_2}$ 's, the char will continue to increase in thickness as a function of time. At high  $\dot{q}$ 's and  $P_{t_2}$ 's, a constant char thickness is rapidly established, and the front surface recedes as rapidly as the char-virgin material interface. Progressively higher  $\dot{q}$ 's and  $P_{t_2}$ 's result in increasingly thinner char layers (AVCO tests) until the char thickness is effectively zero (Cornell tests).

The char densities were found to increase with increasing  $\dot{\mathbf{q}}$ ,  $P_{t_2}$ , and t. This probably results from the kinetics of the polymer pyrolysis process and the kinetics of coke deposition within the char. The char density was found to increase with front surface temperature and mass loss rate, since both are dependent on  $\dot{\mathbf{q}}$  and  $P_{t_2}$ . At a low  $\dot{\mathbf{q}}$  and  $P_{t_2}$ , the char density was approximately four tenths of the virgin density, and at extremely high  $\dot{\mathbf{q}}$  and  $P_{t_2}$ , the char density approached the virgin material density. Char yield can be calculated from char density in two ways, depending on whether char recession is allowed for. If it is, and if the virgin polymer interface does not recede at the same rate, then char yield will be a function of run time with yield decreasing as more and more recession occurs. On the other hand, if char yield is based on the amount of virgin polymer represented by the char cap, then yield is directly proportional to char density, and the above remarks on effects of the different variables on density refer also to char yield.

A density traverse of the char layer was made using the X-ray techniques described in Sec. III-C, and the results of four of these tests are given in Fig. 30. These curves show a sharp drop in density close to the char-virgin material interface which is at the left side of the plot. This is followed by an additional drop in density, possibly resulting from the volatilization of ablation products. The density then increases, probably resulting from cracking of the gases and redeposition of carbon.

#### 2. Mass Loss Rate Correlations

In view of the success in relating the mass loss rate to a power function of the environmental parameters for Teflon and high-density phenolic-nylon, it was decided to use the same approach for the low-density materials. Stagnation pressure was one of these parameters in almost every case. The other parameters considered are discussed in the following sections.

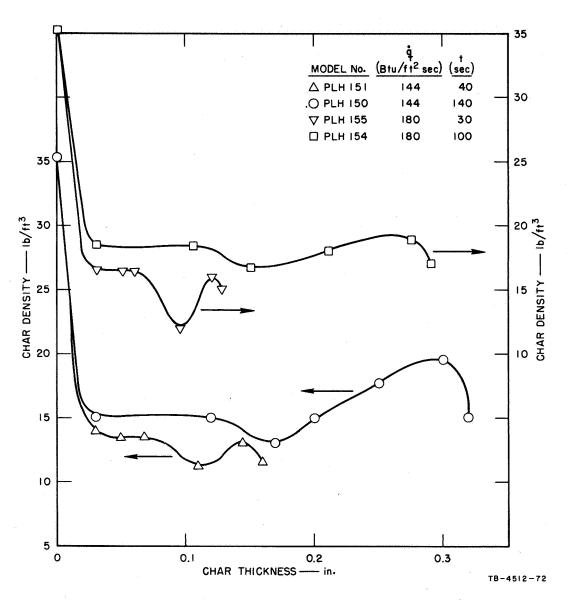


FIG. 30 DENSITY PROFILES OF HUGHES LOW-DENSITY PHENOLIC-NYLON CHARS

### a. SRI Calorimeter Cold Wall Heating Rate

The general form of the relation evaluated was

$$\dot{m}_{t} = a(\dot{q}_{SRI})^{n}(P_{t_{2}})^{m} \qquad (20)$$

using the correlation data from Appendix D for the five low-density materials. The values of the constants found, the degree of correlation, and the percent standard deviation are tabulated below.

MATERIAL	. <b></b>	n	m	MULTIPLE CORRELATION COEFFICIENT	PERCENT STANDARD DEVIATION
PLL	0.00465	0.36	0.26	0.96	15
PLH	0.00388	0.36	0.19	0.94	14
Α	0.00357	0.47	0.33	0.97	16
SP	0.000317	0.81	0.19	0.94	24
SG	0.000188	1.03	0.28	0.92	36
T*	0.0060	0.57	0.25	0.97	10
P*	0.0018	0.55	0.13	0.96	10

Data from Phase I' round robin.

The increased standard deviation for the low-density materials, as compared to the higher density materials, is not surprising. Their composition and ablated appearance is less reproducible and it is more difficult to measure linear dimensions on the charred core. In fact, attempts to correlate the pyrolysis rate,  $\dot{\mathbf{m}}_{p}$ , as a power function of heating rate and stagnation pressure showed a poorer fit of 18, 20, and 24 percent for PLL, PLH, and A, respectively.

Plots of the correlations for the five low-density materials are shown in Figs. 31 through 35. The MSC-Houston subsonic data also shown on each graph so that they can be compared to the supersonic results. The subsonic data were not considered in calculating the intercept, exponents, and standard deviation for the correlations.

It is visually apparent that the correlations are poorest for the silicone materials. Also, the exponents on heating rate and stagnation pressure vary from material to material. This tends to reinforce the suggestion that the Teflon and high-density phenolic-nylon material should not be combined into a single correlation.

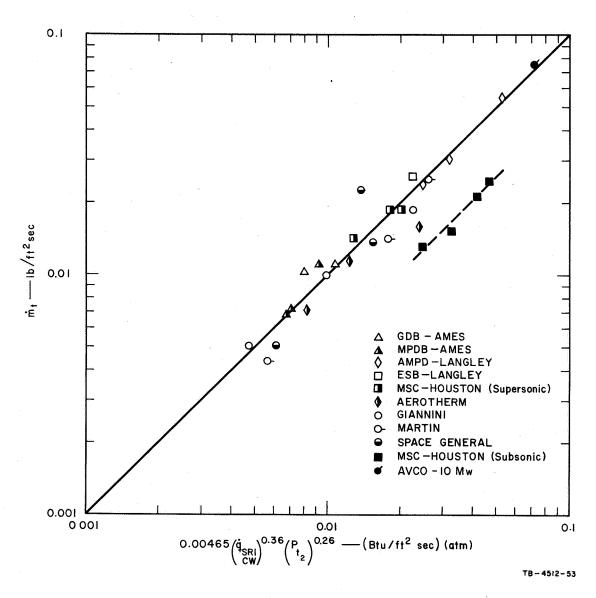


FIG. 31 MASS LOSS RATE CORRELATION FOR LANGLEY LOW-DENSITY PHENOLIC-NYLON

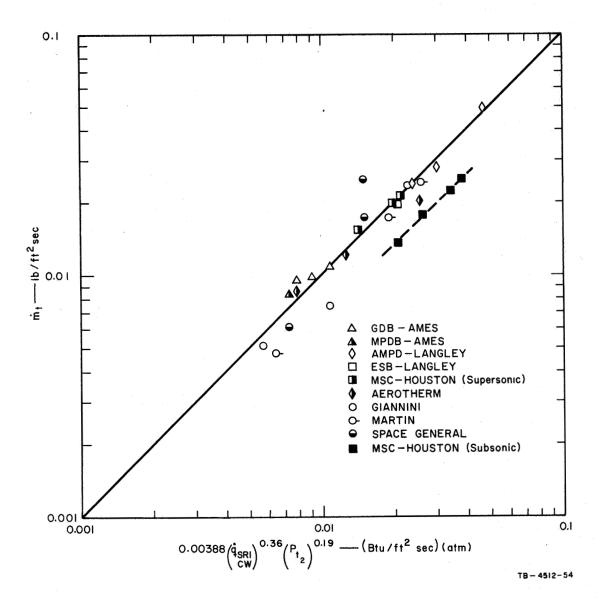


FIG. 32 MASS LOSS RATE CORRELATION FOR HUGHES LOW-DENSITY PHENOLIC-NYLON

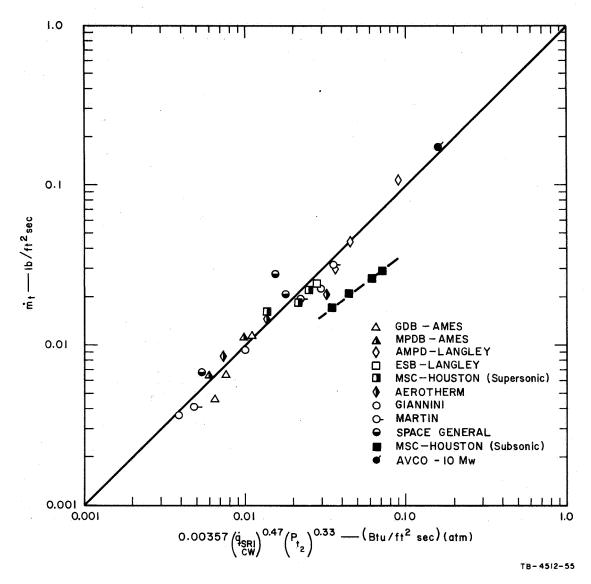


FIG. 33 MASS LOSS RATE CORRELATION FOR AVCOAT

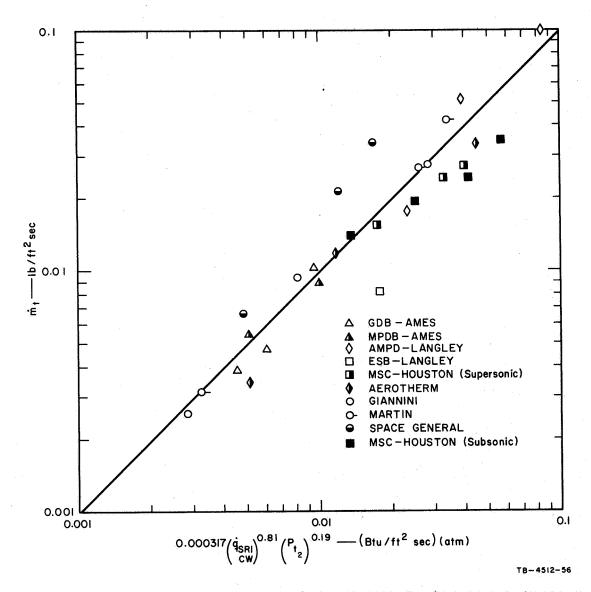


FIG. 34 MASS LOSS RATE CORRELATION FOR MODIFIED PURPLE BLEND SILICONE

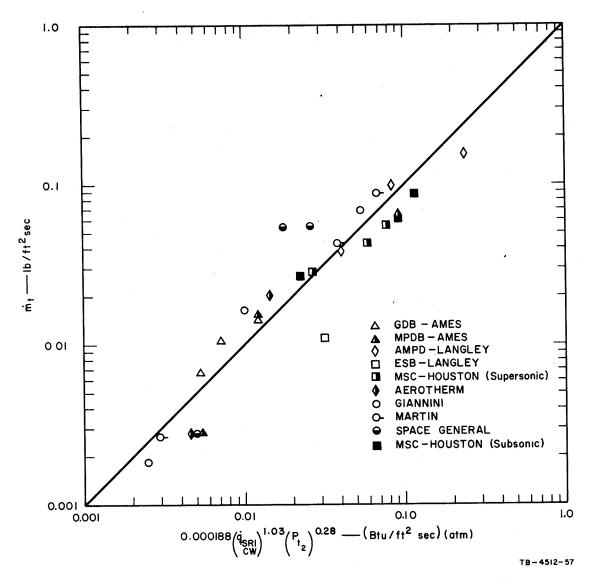


FIG. 35 MASS LOSS RATE CORRELATION FOR GENERAL ELECTRIC SILICONE

## b. Facility Calorimeter Cold Wall Heating Rate

The form of the relation is essentially similar to that used for the SRI calorimeter heating rate correlation:

$$\dot{m}_{t} = a(\dot{q}_{FAC})^{n}(P_{t_{2}})^{m} \qquad (21)$$

This was evaluated using the regression program available on the SRI computer and the data from Appendix D. The results are:

MATERIAL	a	n	m	MULTIPLE CORRELATION COEFFICIENT	PERCENT STANDARD DEVIATION
PLL	0.00538	0.32	0.25	0.95	15
PLH	0.00430	0.35	0.20	0.91	16
Α	0.00414	0.44	0.33	0.97	16
SP	0.000420	0.75	0.18	0.91	26
SG	0.000299	1.00	0.34	0.91	39
T*	0.011	0.48	0.29	0.98	11
P*	0.0034	0.46	0.18	0.97	8

<sup>\*</sup> Data from Phase I round robin.

The data scatter for the low-density materials is nearly the same as when the SRI calorimeter is used, except that the silicone materials show a slightly poorer correlation. As would be expected, the constants are roughly the same for the two correlations.

### c. Measured Enthalpy Potential

The correlation evaluated for this environmental parameter was

$$\dot{m}_{t} = a(\Delta h_{meas})^{u}(P_{t_{2}})^{v} \qquad (22)$$

The data from Appendix D were used in the regression program with the following results:

MATERIAL	a	u	<u>v</u>	MULTIPLE CORRELATION COEFFICIENT	PERCENT STANDARD DEVIATION
PLL PLH A SP SG T * P *	0.000980 0.000650 0.000966 0.0000133 0.00000189 0.0017	0.46 0.48 0.52 0.98 1.32 0.59 0.49	0.47 0.41 0.60 0.64 0.86 0.57 0.41	0.96 0.94 0.90 0.92 0.91 0.92 0.78	16 15 21 28 40 21 30

<sup>\*</sup> Data from Phase I round robin

The data for the two low-density phenolic-nylon materials show nearly as good a correlation as when the cold wall heating rate is used. Greater standard deviations are found for most of the other materials.

## d. Heat of Ablation

A common method of interpreting mass loss data is in terms of the effective heat of ablation,  $H_{\rm eff}$ . This is determined and related to the measured enthalpy potential as shown below:

$$\dot{q}_{SRI}/\dot{m}_{t} = H_{eff} = \alpha + \beta \left(\Delta h_{meas}\right) \tag{23}$$

The coefficient  $\alpha$  is derived to be the heat necessary to raise the material to the ablation temperature and to decompose it, and thus is identical to the term defined earlier as  $\Delta H_D$ , whereas  $\beta$  is a dimensionless number defined as the transpiration shielding factor. A regression analysis of the data from Appendix D, on this basis, leads to the following values for the constants:

MATERIAL	α	β	MULTIPLE CORRELATION COEFFICIENT
PLL	5,654	1.16	0.67
PLH	5,428	1.03	0.68
A	4,248	1.03	0.78
SP	12,580	0.476	0.52
SG	14.130	0.040	0.03

It is apparent that this is not a suitable correlation for the data for a number of reasons. The multiple correlation coefficient is so low as to suggest that a number of sets of  $\alpha$  and  $\beta$  could be used equally well. A plot of the best correlation, that for Avcoat, is shown in Fig. 36.

Equation (23) for Avcoat can also be arranged for logarithmic plotting as

$$H_{eff} - 4248 = 1.03(\Delta h_{meas})$$
 $CW$  (24)

This has been graphed in Fig. 37 and the relation shows a standard deviation of 34 percent. This is over twice the standard deviation of 16 percent for the power function correlation graphed in Fig. 33 and shows the superiority of Equation (20) over Equation (23) or (24).

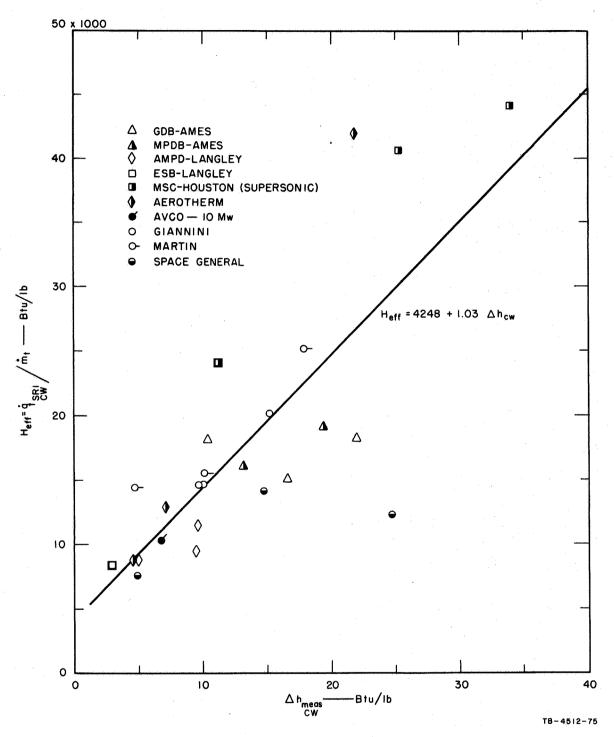


FIG. 36 EFFECTIVE HEAT OF ABLATION CORRELATION FOR AVCOAT

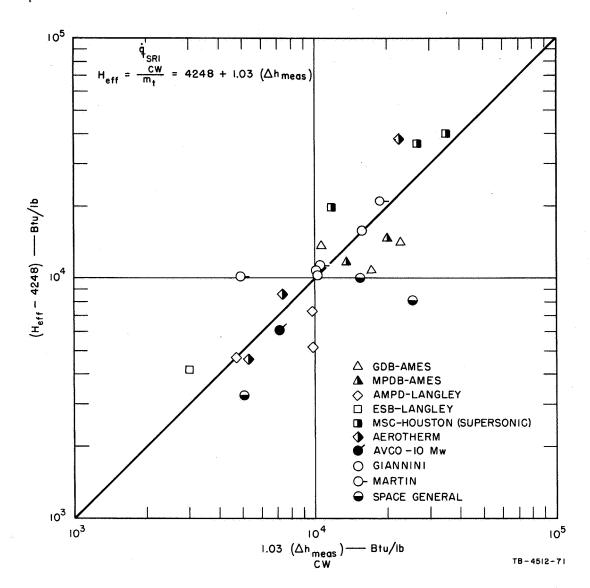


FIG. 37 EFFECTIVE HEAT OF ABLATION CORRELATION FOR AVCOAT (Logarithmic Form)

### e. Dimensionless Forms of Correlations

As with the Phase I round-robin data, the power function correlations can be expressed in dimensionless form. Thus Equations (20) and (21) can be expressed in the same form as Equation (E-9), namely,

$$\pi_{m} = a_{0}\pi_{q}^{n}\pi_{p}^{m} \qquad (25)$$

with the  $\pi$ -groups defined as in Equations (E-4) to (E-6). However, since the values of a  $_0$  and  $\Delta H_D$  are not known for these materials, Equation (25) must be reduced to the dimensional form shown in Equation (E-16A), namely,

$$\dot{m}_{t} = b(R_{eff})^{n+m-1} \dot{q}_{CW}^{n} P_{t_{2}}^{m}$$
 (26)

where

$$b = a(R_{eff})_{RR}^{1-n-m}$$
 with  $(R_{eff})_{RR} = 0.172 \text{ ft}$  (27)

In the case of the measured enthalpy potential correlation, Equation (22) can be expressed as

$$\pi_{\rm m} = a_0 \pi_{\rm h}^{\rm u} \pi_{\rm p}^{\rm v} \tag{28}$$

where

$$\pi_{h} = (\Delta h)_{\substack{m e \text{ a s} \\ CW}} / \Delta H_{D}$$
 (29)

and the other  $\pi$ -groups are as previously defined in Appendix E. Expansion of this in dimensional form leads to

$$\dot{m}_{t} = a(\Delta h_{meas})^{u}(P_{t_{2}})^{v}$$
(30)

where

$$a = a_0 (S_R^2 R_{eff})^{v-1} (\Delta H_D)^{(1/2)-u-v} (K)^{1-2v}$$
(31)

Since  $\Delta H_D$  is not known, this can be converted to

$$\dot{m}_{t} = b_{1}(R_{eff})^{v-1}(\Delta h_{meas})^{u}(P_{t_{2}})^{v}$$
(32)

where

$$b_1 = a(R_{eff})_{RR}^{1-v}$$
 with  $(R_{eff})_{RR} = 0.172 \text{ ft}$  (33)

Equation (32) has the same form as Equation (27) in the Phase I report<sup>1</sup> except that the b given there equals  $b_1(R_{eff})^{v-1}$  here.

### 3. Temperature Correlations

Two distinct sets of temperature measurements were made during the Phase II round robin. These were front surface temperature during exposure and internal temperature of the ablation model during a run. The results of these measurements have been tabulated in Appendix C. A discussion of these data and of correlations based on them follow.

# a. Front Surface Temperature

As pointed out earlier, several optical techniques were used for determining front surface temperature, and in most cases the results were not directly comparable. However, the facility pyrometers were previously calibrated in place and the data from these instruments were used for correlation purposes. Relations involving such factors as mass loss rate, pyrolysis rate, heating rate, and stagnation pressure were evaluated, but the simplest was

$$T_{FS} = a(\dot{m}_p)^w \tag{34}$$

where  $T_{FS}$  is the front surface temperature in degrees Rankine. The regression analysis led to

MATERIAL	a	·w	MULTIPLE CORRELATION COEFFICIENT	PERCENT STANDARD DEVIATION
PLL	12,150	0.26	0.94	5
PLH	12,440	0.27	0.90	7
Α	10,780	0.23	0.88	6
SP	7,820	0.17	0.94	4
SG	5,370	0.11	0.84	5
P*	7,510	0.18	0.84	5

<sup>\*</sup> Data from Phase I round robin.

Plots of the correlations for these six materials are given in Figs. 38 through 40. The correlation and percent standard deviation for the high-density phenolic-nylon were determined with the General Electric data excluded. Their pyrometer differed from the other optical parameters supplied by the various facilities during the Phase I round robin in that it was a special, in-house, two-color design. These correlations were quite satisfactory.

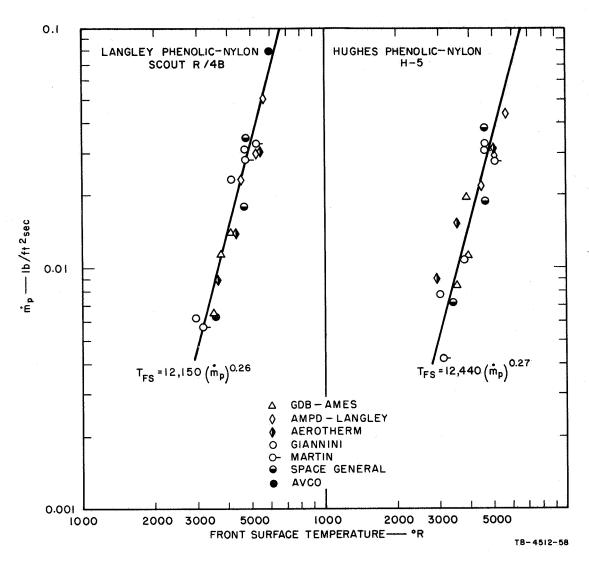


FIG. 38 FRONT SURFACE TEMPERATURE CORRELATION FOR LANGLEY AND HUGHES LOW-DENSITY PHENOLIC-NYLONS

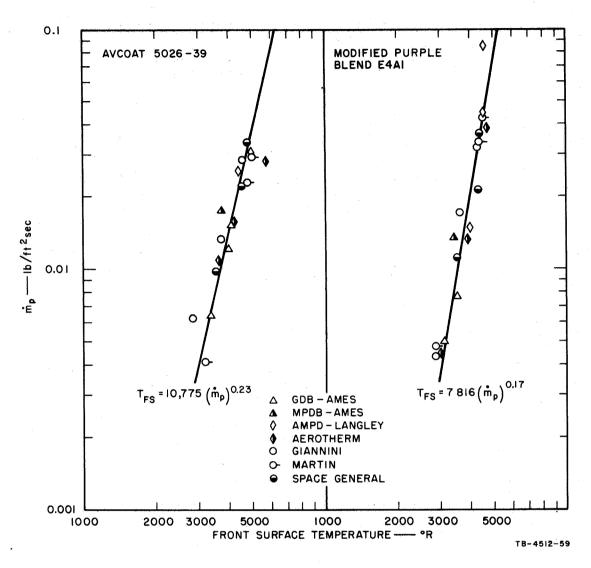


FIG. 39 FRONT SURFACE TEMPERATURE CORRELATION FOR AVCOAT AND MODIFIED PURPLE BLEND SILICONE

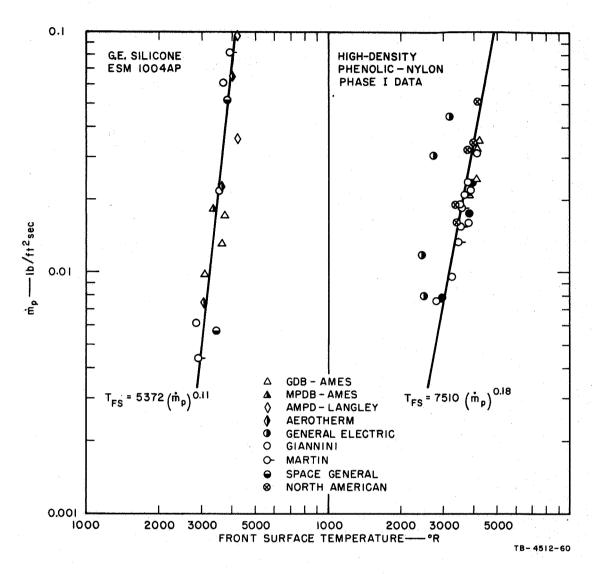


FIG. 40 FRONT SURFACE TEMPERATURE CORRELATION FOR GENERAL ELECTRIC SILICONE AND HIGH-DENSITY PHENOLIC-NYLON

As good a correlation was obtained by replacing  $\dot{m}_p$  with  $\dot{m}_t$ . Other equally good correlations were

$$T_{FS} = a(P_{t_2})^{v}(\dot{m}_{p})^{w} \qquad (35)$$

and

$$T_{FS} = a(\dot{q}_{SRI})^{n}(P_{t_2})^{m}$$
(36)

This was expected since  $T_{FS}$  correlates with  $\dot{m}_p$  (see Equation (21) or  $\dot{m}_t$ , and they correlate with heating rate and stagnation pressure (see Equation (20)). For Equation (35), with the pyrolysis rate,  $\dot{m}_p$ , used, the regression analysis gave

MATERIAL	a	, <b>y</b>	W	MULTIPLE CORRELATION COEFFICIENT	PERCENT STANDARD DEVIATION		
PLL	10,980	0.031	0.21	0.95	5		
PLH	10,710	0.044	0.20	0.92	.6		
Α	10,040	0.039	0.18	0.90	6		
SP	7,660	0.012	0.16	0.95	4		
SG	5,210	0.028	0.072	0.86	5		
P*	7,260	0.0076	0.17	0.84	4		

<sup>\*</sup> Data from Phase I round robin.

#### b. Internal Temperature Rise

A number of the models for all five of the low-density materials were internally instrumented with thermocouples. The method of preparing these models and the information obtained have been described earlier.

(1) Any Temperature Rise. Of considerable interest in ablation design is the thickness of a given material required to prevent the bond line from reaching a given temperature before a given time. It was therefore decided to try correlating thermocouple position, x (in inches), with the time to reach a given temperature, t (seconds for a temperature rise of ( $\triangle T$ ), and the environmental parameters of heating rate and stagnation pressure. The form of the relation evaluated was

$$x = a(P_{t_2})^b (\dot{q}_{SRI})^c (t)^d (\Delta T)^e$$
(37)

Approximately six points from each temperature profile given in Appendix C, covering the range from low temperature rises to values of approximately 1200°F, were used in the regression analysis. The results were

MATERI AL	a	b	с	d	e	MULTIPLE CORRELATION COEFFICIENT	PERCENT STANDARD DEVIATION
PLL	0.034	0.053	0.30	0.63	-0.28	0.93	13
PLH	0.056	0.035	0.15	0.58	-0.24	0.97	9
. <b>A</b>	0.037	0.018	0.27	0.60.	-0.26	0.92	14
SP	0.072	0.022	0.18	0.52	-0.30	0.95	12
SG	0.12	0.031	0.098	0.54	-0.28	0.97	8

A. J. Chapman <sup>14</sup> interpreted model temperature data in a similar way. For a low-density phenolic-nylon prepared at Langley ( $\rho$  = 39 lb/ft<sup>3</sup>), his relation, converted to be comparable to Equation (37), was  $x = 0.013 \ \dot{q}^{0.39} t^{0.89} \Delta T^{-0.39}$ . This is not far different from the values for phenolic-nylon (PLL) when one considers that the experiments were performed in a subsonic facility with an invariant stagnation pressure of one atmosphere. Therefore,  $P_{t_2}$  could not be included in the relation.

Recently, postlaunch reports have become available for several of the unmanned Apollo spacecrafts which used Avcoat (A) for the ablating material on the heat shield. Data from these manned Spacecraft Center Reports 15,16 were used in Equation (37) along with the above constants for Avcoat (A) to predict the positions of the 600 and 1000°F isotherms. These predictions are listed with the NASA predictions and measured depths in Table V. They compare very favorably.

(2) Temperature Rise of  $250^{\circ}F$ . A more limited correlation was tried in which the temperature rise,  $\triangle T$ , was  $250^{\circ}F$ ; for this, the time was designated  $t_{250}$ . The form of this relation was

$$x = a(P_{t_2})^b (\dot{q}_{SRI})^c (t_{250})^d$$
 (38)

The regression analysis, using the time to a  $250^{\circ}F$  temperature rise at each thermocouple position, as tabulated in Appendix C, led to

MATERIAL	ATERIAL a b		c	d	MULTIPLE CORRELATION COEFFICIENT	PERCENT STANDARD DEVIATION
PLL	0.014	0.083	0.18	0.61	0.97	8
PHL	0.017	0.079	0.14	0.60	0.99	5
Α	0.023	0.105	0.15	0.62	0.99	6
SP	0.0082	0.016	0.26	0.54	0.97	9
SG	0.033	0.065	0.046	0.55	0.99	2

PREDICTED AND MEASURED ISOTHERMS FOR AVCOAT MATERIAL ON APOLLO MISSIONS Table V

0	o <sub>F</sub> herm	· :	21	35	35	22	88		lor- n	4	<u>&amp;</u>	0 1	* <u>-</u>	0	4	9,
NASA MEASURED		(in.)	0.47	0.35	0.35	0.37	0.38		Discolor- ation	Dept	1.08	1.00	0.91	0.80	0.84	0.96
NASA M	1000°F Isotherm	(in.)	0.32	0.21	0.21	0.22	0.25		Char Denth	no doc	0.68	0.66	0.59	0.46	0.58	0.75
DICTED	600°F Isotherm	(in.)	68.0	0.42	0.38	0.45	0.45		olated erms	4 <sub>0</sub> 009	1.07	0.95	0.80	0.75	0.57	0.87
NASA PREDICTED	1000°F Isotherm	(in.)	0.25	0.29	0.26	0.32	0.32		Extrapolated Isotherms	1000°F	0.73	0.63	0.50	0.52	0.40	0.63
HIELD	NO C	Y (in.)	0	0	0	39	7.1			0)	0	0 0	.5	0	+38	+69.5
HEAT SHIELD	LOCAT (6)	Z (in.)	11	0	-71	0	0			(10)	02	64	04.0	-67.5	0	0
SRI PREDICTED	600°F Isotherm	(in.) $\Delta T = 500^{O}F$	0.35				0.38	-			0.91					
SRI PRE	1000°F Isotherm	$(in.)$ $\Delta T = 900^{0}F$	0.28				0,31				0.74					
INITIAL	IEMPER- ATURE TioF	(2)	100				100			(6)	100					
STAGNA-	$\begin{array}{c} \text{TION} \\ \text{PRESSURE} \\ \text{P} \\ \text{t} \\ 2 \text{AV} \end{array}$	(atm) (4)	0.408				0.408			(8)	0.061					
HEAT	IRANSFER RATE • AV	(Btu/ft <sup>2</sup> sec) (3)	71.8				104	de la constantina della consta			27					
FLIGHT	I IME	(sec)	85				85			(8)	750					
TOTAL	$\begin{array}{c} \text{HEAI} \\ \text{Q} = \mathbf{qt} \\ \end{array}$	(Btu/ft <sup>2</sup> ) (1)	6,100				8,800			(2)	20,000					
			Misson AS 201		-					_	Misson AS 202		-	-		

(1) From Fig. 7.4-9, pp. 7-83, MSC-A-R-66-4.

(6) From Table 7.4-I, pp. 7-69, MSC-A-R-66-4.

<sup>(2)</sup> From Fig. 6.2-1, pp. 6-19, MSC-A-R-66-4.

<sup>(3)</sup>  $Q/t = \dot{q}_{AV}$ .

<sup>(4)</sup> From Fig. 6.2-1, pp. 6-19, MSC-A-R-66-4. (5) From Fig. 7.4-7, pp. 7-77, MSC-A-R-66-4.

<sup>(7)</sup> From Fig. 7.3-1, pp. 7-58 and pp. 1-2, MSC-A-R-66-5. (8) From Fig. 7.3-2, pp. 7-64, MSC-A-R-66-5.

<sup>(10)</sup> From Table 7.4.2-I, pp. 7-100, MSC-A-R-66-5. (9) From Fig. 7.4.2-4, pp. 7-104, MSC-A-R-66-5.

As expected, the multiple correlation coefficient rises and the percent standard deviation drops, as compared to the correlation given in Equation (37), since one would predict from one-dimensional heat transfer theory that  $\Delta T$  would enter into the relation in a more complex way than a simple power function.

A plot of this correlation was difficult to make with these data because the various thermocouple positions were essentially the same in all models, causing the points to bunch up at these values of x. For this reason, the relation was inverted to make  $t_{250}$  the dependent variable, as follows:

$$t_{250} = a(P_{t_2})^b (q_{SRI})^c (x_{250})^d$$
 (39)

Results of the regression analysis were

MATERIAL	, .a	b	c	d	MULTIPLE CORRELATION COEFFICIENT	PERCENT STANDARD DEVIATION
PLL	1110	-0.26	-0.33	1.54	0.98	13
PLH	930	-0.13	-0.25	1.62	0.99	8
Α	470	-0.166	-0.25	1.58	0.99	9
SP	7250	+0.0074	-0.53	1.70	0.97	16
SG	370	-0.12	-0.087	1.81	0.99	. 5

The constants and exponents are not directly convertible between Equations (38) and (39) because the regression analysis maximizes the multiple correlation coefficient for the dependent variable. The closer this coefficient is to unity, however, the better is the conversion between the constants and the exponents. The difference in percent standard deviation also arises from the fact that it is calculated for the dependent variable, and  $t_{250}$  is more sensitive to  $x_{250}$  (because the coefficient d in Equation (39) is greater than unity) than vice versa.

The correlation given in Equation (39) is graphed in Figs. 41 through 45 for the various materials. It should be remembered that in all cases the thickness represented by  $\mathbf{x}_{250}$  can be converted to weight per unit area,  $\mathbf{W}_{250}$  (in pounds per square foot), by use of the polymer density,  $\rho_{\mathrm{VR}}$  (in pounds per cubic foot).\* Thus,

$$W_{250} = x_{250} (\rho_{VR}/12)^{VR}$$
 (40)

<sup>\*</sup> Polymer densities are given in Table I.

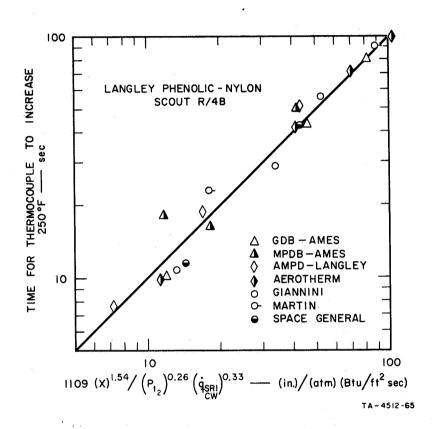


FIG. 41 INTERNAL TEMPERATURE CORRELATION
FOR LANGLEY LOW-DENSITY PHENOLIC-NYLON

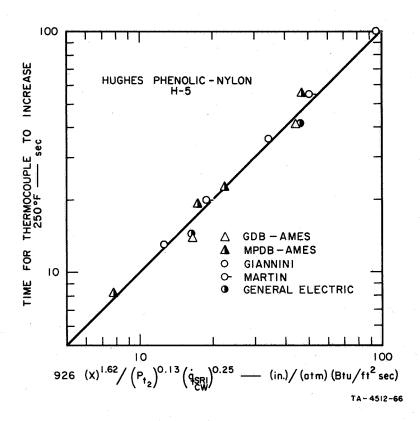


FIG. 42 INTERNAL TEMPERATURE CORRELATION FOR HUGHES LOW-DENSITY PHENOLIC-NYLON

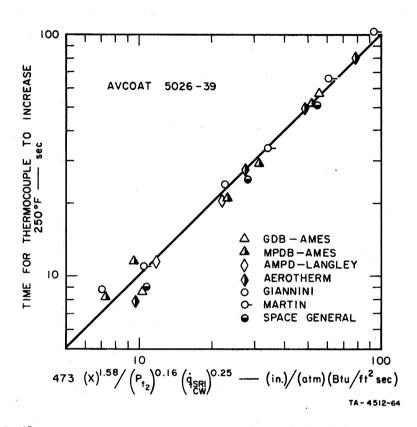


FIG. 43 INTERNAL TEMPERATURE CORRELATION FOR AVCOAT

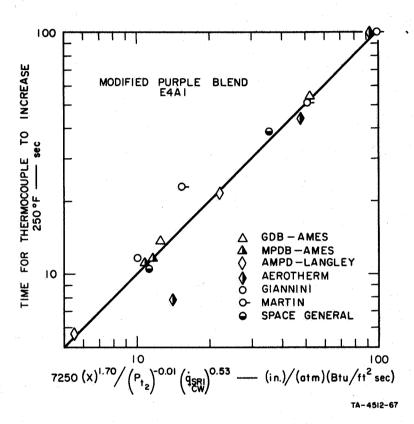


FIG. 44 INTERNAL TEMPERATURE CORRELATION FOR MODIFIED PURPLE BLEND SILICONE

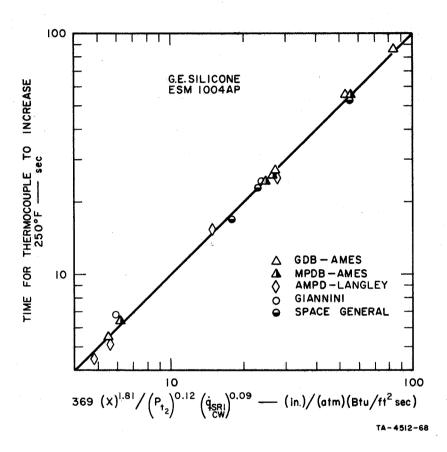


FIG. 45 INTERNAL TEMPERATURE CORRELATION FOR GENERAL ELECTRIC SILICONE

### c. Dimensionless Forms of Correlations

The development of dimensionless forms for the temperature correlations involve additional variables over those considered in the mass loss cases. These additional variables differ for the two cases of concern: front surface and internal temperature.

(1) Front Surface Temperature. The variables considered in the dimensional analysis are similar to those listed in Appendix E, except for the elimination of the mass loss rate,  $\dot{m}_{\rm t}$ , and its dimensionless group,  $\pi_{\rm m}$ , since it is a function of  $\pi_{\rm q}$  and  $\pi_{\rm p}$ . In addition, new variables to be added are the front surface temperature,  $T_{\rm FS}$ , and the emissivity of the ablating surface,  $\epsilon$ . The units for these variables are:

VARIABLES	UNITS	CONVERTED VARIABLES	CONVERTED UNITS
T <sub>FS</sub>	o <sub>R</sub>	T <sub>FS</sub>	o <sub>R</sub>
$\epsilon$	None	$\epsilon$	None

These involve one additional dimension, temperature, and so one additional dimensionless group is required (=2 - 1). This group is

$$\pi_{s} = (\epsilon \sigma T_{FS}^{4} S_{R}^{2} R_{eff}) / (\Delta H_{D}^{1.5} K)$$
 (41)\*

The Stefan-Boltzmann constant,  $\sigma$ , is a conversion factor having the following value

$$\sigma = 4.76 \times 10^{-13} \text{ Btu/ft}^2 \text{ sec } ^{\circ} R^4$$

In its converted variable form, it becomes  $\sigma J_m g_c$  with units of  $lb/sec^3$   $^{\circ}R^4$ . Definitions of  $J_m$  and  $g_c$  are given in Equation (E-3A) of Appendix E.

Neglecting the Fay-Riddell group,  $\pi_{\rm f}$ , for the experiments performed under supersonic flow conditions, the correlation relation might be

$$\pi_{s} = b_{0}\pi_{q}^{u}\pi_{p}^{v} \tag{42}$$

<sup>\*</sup> See Equation (E-3B) of Appendix E for the definition of the conversion factor K.

However, there is already a power function relation between the last two  $\pi\text{-groups}$  and  $\pi_{\rm m}$  (see Equation (20)), and so one of them,  $\pi_{\rm q}$ , can be eliminated to give

$$\pi_{s} = c_{0}\pi_{p}^{z}\pi_{m}^{y} \tag{43}$$

Since  $\Delta H_D$  is still not known for these materials, expansion of this relation into a dimensional form leads to

$$T_{FS} = a(P_{t_2})^{z/4}(\dot{m}_p)^{y/4}$$
 (44)

where

$$a = \left(\frac{c_0}{\epsilon \sigma}\right)^{0.25} \left(S_R^2 R_{eff}\right)^{(z+y-1)/4} \Delta H_D^{(3-2z-y)/8} K^{(1-2z-y)/4}$$
 (45A)

and

$$z = 4v, \qquad y = 4w \tag{45B}$$

in Equation (35). If z/4, or v, is quite small, the pressure term will approach unity and Equation (44) reduces to

$$T_{FS} \stackrel{\sim}{=} a(\dot{m}_{p})^{y/4} \tag{46}$$

where y = 4w in Equation (34) and "a" is the same as in Equation (45A) except that z is set to zero. The exponents given for these cases and their converted values are given below, where it is seen that v is indeed relatively small.

		•	(35 and ( (P <sub>t2</sub> ) (m <sub>p</sub> )		and	ons (34) (46) a(m <sub>p</sub> ) <sup>w</sup>
Material	v	w	z = 4v	y = 4 w	w	y = 4w
PLL	0.031	0.21	0.13	0.84	0.26	1.04
PĻH	0.044	0.20	0.18	0.80	0.27	1.08
Α	0.039	0.18	0.16	0.72	0.23	0.92
SP	0.012	0.16	0.048	0.64	0.17	0.68
SG	0.028	0.072	0.12	0.29	0.11	0.44

The values of a, v, and win Equations (34) and (35) when the mass loss rate,  $\dot{m}_t$ , is used in the correlations are given in Table VI. This table also gives the values of a, n, and m for Equation (36). It should be remembered that these are directly interconvertible with Equation (11) because of the emphasis the regression program puts on the dependent variable in determining the best correlation.

 $\label{thm:constants} Table\ VI$  Constants for additional front surface temperature correlations

### A. Equation (34) $T_{FS} = a(m_t)^w$

MATERIAL	a	w	MULTIPLE CORRELATION COEFFICIENT	PERCENT STANDARD DEVIATION
PLL	11,260	0.23	0.90	6
PLH	13,480	0.28	0.91	6
A	7,990	0.16	0.86	6
SP	7,410	0.15	0.96	3
SG	5,000	0.082	0.89	4

### B. Equation (35) $T_{FS} = a(P_{t_2})^{v(m_t)^{w}}$

MATERIAL	a	v	w	MULTIPLE CORRELATION COEFFICIENT	PERCENT STANDARD DEVIATION
PLL	10,260	0.019	0.19	0.91	6
PLH	12,530	0.017	0.25	0.91	6
Α	8,620	0.026	0.20	0.87	6
SP	7,530	-0.016	0.17	0.96	3
SG	5,010	0.0067	0.076	0.90	4

C. Equation (36) 
$$T_{FS} = a(q_{SRI})^n (P_{t_2})^m$$

MATERIAL	" ā	n	m	MULTIPLE CORRELATION COEFFICIENT	PERCENT STANDARD DEVIATION
PLL	2,870	0.10	0.052	0.94	5
PLH	2,600	0.12	0.055	0.93	5
A	2,280	0.13	0.025	0.90	6
SP	2,110	0.13	0.028	0.93	5
SG	2,680	0.077	0.032	0.88	4
P*	2,530	0.096	0.032	0.80	5
					-

<sup>\*</sup> Data from Phase I round-robin.

(2) Internal Temperature Rise. In this dimensional analysis, two variables—the front surface temperature,  $T_{FS}$ , and its related term,  $\epsilon$ ,—can be dropped since their dimensionless group,  $\pi_s$ , is a function of  $\pi_q$  and  $\pi_p$  and can be eliminated. The five new variables to be considered are the position, x, at which a given temperature rise,  $\Delta T$ , has taken place at a given time, t, the heat capacity,  $C_p$ , and the density of the virgin polymer,  $\rho_{VB}$ . The units of these variables are

Variables	Units	Converted Variables	Converted Units
x	ft	x	ft
$\triangle T$	F	ΔT	$^{\circ}F$
t	sec	t	sec
Cp	Btu/lb °F	$^{\mathrm{C}}_{\mathrm{p}}\mathrm{J}_{\mathrm{m}}\mathrm{g}_{\mathrm{c}}$	$\mathrm{ft}^2/\mathrm{sec}^2$ °F
$ ho_{ m VR}$	lb/ft <sup>3</sup>	$ ho_{ m VR}$	$lb/ft^3$

A net of three new variables has been added without any change in the number of dimensions, and there is one dimensionless group to be replaced,  $\pi_{\rm s}$ ; thus four (3 + 1) new dimensionless groups are required. In their simplest form, these are

$$\pi_{x} = x/R_{eff} \tag{47}$$

$$\pi_{\rm T} = C_{\rm p} \Delta T / \Delta H_{\rm D} \tag{48}$$

$$\pi_{t} = t(\Delta H_{D}J_{m}g_{c})^{0.5}/R_{eff}$$
 (49)\*

$$\pi_{\rm D} = \rho_{\rm VB} S_{\rm B}^2 R_{\rm eff} F_{\rm p} g_{\rm c} \tag{50}$$

The numerator of the second of these (48) represents the heat stored in the virgin polymer per unit mass and is the cause of the temperature rise. Other variables might be considered, such as those to allow for conductive heat flow, but this would require definition of another temperature difference and does not add new information.

The correlation proposed is

$$\pi_{x} = b_{0} \pi_{p}^{b} \pi_{q}^{c} \pi_{t}^{d} \pi_{T}^{e} \pi_{D}^{f} \qquad (51)$$

<sup>\*</sup> See Equation (E-3A) of Appendix E for definitions of conversion factors  $g_c$ ,  $J_m$ , and  $F_p$ .

Again,  $\triangle H_D$  is not known for these materials, and so the expanded dimensional form is

$$x = a(P_{t_2})^b (\dot{q}_{SRI})^c (t)^d (\Delta T)^e$$
(52)

where

$$a = b_0(R_{eff})^{1+b+c-d+f}(S_R^2)^{b+c+f}(\Delta H_D)^{(d-2b-3c-2e)/2}(C_p)^{e}(\rho_{VR})^{f}$$

$$\cdot J_m^{(d-2b-c)/2}F_p^{2b+c+f}g_c^{(2f+d+2b+c)/2} . \qquad (53)$$

In the case of the 250  $^{\circ}\mathrm{F}$  isotherm, the  $\pi_{\mathrm{T}}$  term becomes a constant so that Equation (51) becomes

$$\pi_{x} = g_{0}\pi_{p}^{b}\pi_{q}^{c}\pi_{t}^{d}\pi_{D}^{f} \qquad (54)$$

The expanded dimensional form is then

$$x_{250} = a(P_{t_2})^{b}(\dot{q}_{SRI})^{c}(t_{250})^{d}$$
 (55)

where

$$a = g_0(R_{eff})^{1+b+c-d+f}(S_R^2)^{b+c+f}(\Delta H_D)^{(d-2b-3c)/2}(\rho_{VR})^{f} \cdot J_m^{(d-2b-c)/2}F_p^{2b+c+f}g_c^{(2f+d+2b+c)/2}.$$
 (56)

Also,

$$g_0 = b_0 (C_p \Delta T / \Delta H_p)^e$$
 and  $\Delta T = 250^\circ F$ . (57)

The value of the exponent f is almost impossible to determine since density cannot be changed enough to determine its effect without also affecting other properties of the material. However, the success in using Equation (40) to predict isotherms in the Apollo, which has a very much larger effective radius than the models tested in this program, would suggest that the exponent on  $R_{\rm eff}$  is very small. If it is assumed to be zero, then f is approximately equal to d - l - b - c.

#### 4. Comparative Ablation

The correlations reported in the previous sections permit comparison of the ablation behavior of the materials studied as a function of environment.

#### a. Mass Loss Rate

Equation (20) relates the mass loss rate to the stagnation point heating rate and pressure,

$$\dot{m}_{t} = a(\dot{q}_{SRI})^{n}(P_{t_{2}})^{m} \qquad (20)$$

or by use of the Fay-Riddell relation,  $\pi_{\rm f}$  = 1, in terms of the enthalpy potential,

$$\dot{m}_{t} = b(\Delta h_{calc})^{n}(P_{t_{2}})^{(2m+n)/2}$$
(58)

where

$$b = a(S_R)^{-n}(R_{eff})^{-n/2}$$
 (59)

Figure 46 is a logarithmic plot of Equation (58) based on the values of constants in Equation (20) for the five low-density materials. Stagnation pressure values of 0.03 and 0.3 atm were assumed, and  $R_{\rm eff}$  was taken to be the same as the 1.25-in. flat-face model, i.e., 0.172 ft.

At the lower pressures (i.e., higher altitudes) the silicone materials, SP and SG, show much lower mass loss rates at low enthalpies (i.e., low flight velocities). At higher enthalpies, however, the low-density phenolic-nylons, PLL and PLH, have the lowest rates, followed closely by Avcoat (A). With higher stagnation pressures, representing lower altitudes, the low-density phenolic-nylons are best, across almost the entire range of enthalpies.

The behavior of high-density phenolic-nylon (P) and Teflon (T) is also shown at the higher stagnation pressure. The Teflon shows very poor performance, but the high-density phenolic-nylon is better than the silicones at high enthalpies. One factor not considered here is the threshold stagnation pressures at which mechanical forces markedly increase the mass loss rate. This appears to occur at lower pressures for the low-density materials so that high-density phenolic-nylon may behave better at higher pressures.

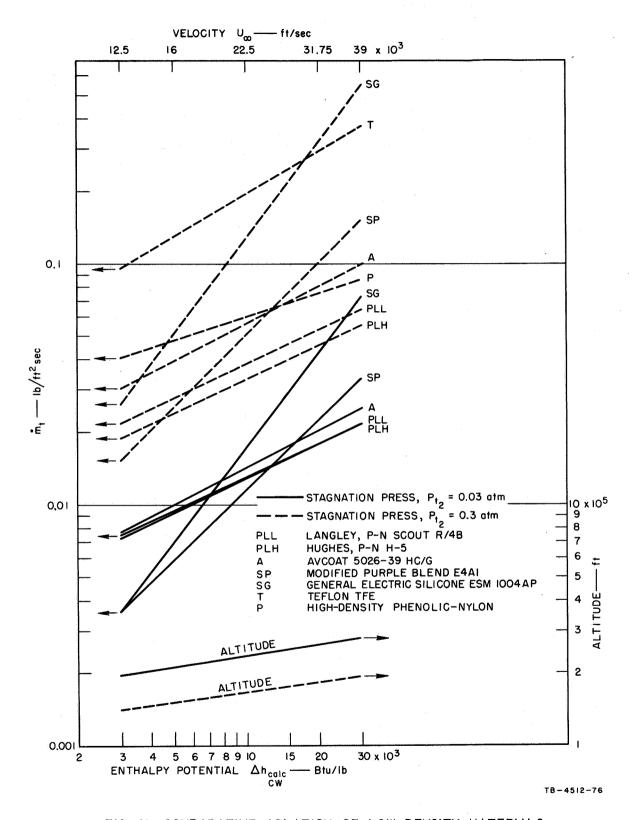


FIG. 46 COMPARATIVE ABLATION OF LOW-DENSITY MATERIALS

#### b. Front Surface Temperature

Relation of pyrolysis rate to front surface temperature is given by Equation (34):

$$T_{FS} = a(\dot{m}_p)^w \qquad . \tag{34}$$

The constants for the various materials can be used to calculate the pyrolysis rate at several values of  $T_{\rm FS}$ . The results are tabulated below.

	m <sub>p</sub> (1b/f	t <sup>2</sup> sec)
MATERIAL	$T_{FS} = 2000^{\circ} F$	$T_{FS} = 4000^{\circ} F$
PLL	0.0021	0.021
PLH	0.0025	0.022
Α	0.0016	0.022
SP	0.0011	0.037
SG	0.00085	0.19
P	0.0021	0.055

This again shows that at the lower thermal environments the silicone materials, SP and SG, perform best; i.e., they have the lowest pyrolysis rates by a factor of two or three. At the higher front surface temperature,  $4000^{\circ}$ F, their behavior is reversed, and they show the highest pyrolysis rates.

The marked change in pyrolysis rate with temperature is undoubtedly related to the chemical reactions involving silicon, oxygen, and carbon. <sup>17</sup> Below the melting point of silica, and this is intermediate to the two temperatures selected for the above tabulation, the surface is protected by silica and some silicon carbide. The latter is formed with the evolution of carbon monoxide. Above the melting point, however, the silica reacts with carbon to form, in addition, liquid silicon and gaseous silicon monoxide which are rapidly removed from the surface.

#### c. Internal Temperature Rise

The heat rejection and insulating power of the various materials is best represented by the internal temperature correlation Equation (39), as follows:

$$t_{250} = a(P_{t_2})^b(q_{SRI})^c(x_{250})^d$$
 (39)

Again using the Fay-Riddell relation,  $\pi_{\rm f}$  = 1, to express this in terms of enthalpy potential, this becomes

$$t_{250} = g(P_{t_2})^{b+(c/2)} (\Delta h_{calc})^{c} (x_{250})^{d}$$
 (60)

where

$$g = a(S_R)^{-c}(R_{eff})^{-c/2}$$
 (61)

The time for the  $250\,^\circ F$  isotherm to reach a position of 0.4 in. back of the front surface has been calculated at a stagnation pressure of 0.03 atm and an  $R_{\rm eff}$  of 0.172 ft for two different enthalpies. The results based on the constants found for Equation (39) are tabulated below.

	$W_{250} (lb/ft^2)$	t <sub>25</sub>	(sec)
MATERIAL	$x_{250} = 0.4 in.$	Δh <sub>calc</sub> = 3000 CW	Δh <sub>calc</sub> = 30,000 CW
PLL	1.19	180	86
PLH	1.18	120	6.9
Α .	1.03	72	41
SP	1.11	190	58
SG	1.22	7 4	60

The best insulator over the range of enthalpies is the Langley low-density phenolic-nylon (PLL); the poorest is Avcoat (A), probably because of conduction along the web.

The lower density of the Avcoat material can be taken into account by selecting material depths that give the same weight loading.

When this is done, the tabulation becomes

	x <sub>250</sub> (in.)	t <sub>25</sub>	0 (sec)
MATERIAL	$W_{250} = 1.03 \text{ lb/ft}^2$	Δh <sub>calc</sub> = 3000 CW	Δh <sub>calc</sub> = 30,000
PLL	0.35	1 40	69
PLH	0.35	97	56
A	0.40	72	41
SP	0.37	167	51
SG	0.34	54	44

The comparative performance of Avcoat improves in this case, but not sufficiently to outrate the other materials at high enthalpies.

#### V CONCLUSIONS AND RECOMMENDATIONS

The results of the Phase II round robin confirm the earlier findings that the mass loss rates of a given material can be correlated in terms of heating rate and stagnation pressure. This is based on more extensive measurements over a wider range of variables. It was also again confirmed that determination of enthalpy by the energy balance method is not satisfactory. This is particularly true when the plasma stream exhibits heating rate and stagnation pressure gradients as was found for many of the facilities used in this study.

The use of a standard calorimeter helped in the interrelating of results, but the use of a standard, calibrated, total radiation pyrometer to measure front surface temperature was successful only when calibrated properly to account for the actual paths and viewing angles. Dimensional analysis shows that the Phase I round-robin data, for both Teflon and high-density phenolic-nylon, can be correlated by a single dimensionless relation in which the values for certain constants vary for each material. One of these, namely, the overall heat of decomposition of the material, cannot easily be determined separately, and for this reason the dimensionless form is converted to a dimensional relation by including this constant in another constant term. The latter relation permits proper allowance for the effective radius of the model in interpreting the data. The success of the dimensional relation in correlating literature data, which cover a thirty-five-fold range of effective radii, confirms this allowance.

The high stagnation pressure runs in Phase II fit the Phase I correlation, in the case of Teflon, up to pressures of 33 atm. The high-density phenolic-nylon data, on the other hand, showed rapidly increasing mass loss rates at pressures above 2.7 atm. The latter data can be correlated successfully, however, in terms of a mechanical stress at which failure of the char occurs. This is confirmed by the fact that these models showed essentially no char layer after testing.

The mass loss rate correlations for the low-density materials show the same form but different exponents than exhibited by the correlations for the high-density materials. The silicone materials do not correlate as well as the low-density phenolic-nylon or Avcoat because there is evidence of an ablation mechanism change over the range of test environments.

Analogous correlations were obtained for front surface temperature and internal temperature rise for the low-density materials, and these can be derived, except for the values of the constants, by dimensional analysis. These three types of correlations can be used for predicting the performance of these materials and also for comparing their ablation behavior under different environments.

Attempts to include subsonic data in these correlations were unsuccessful because only one facility was involved and a broader range of fractional Mach numbers could not be studied.

Based on the results of the Phase II round robin; it is recommended that the following areas be studied further:

- 1. Determination of the critical stagnation pressure at which char failure begins for the low-density materials
- 2. Evaluation of other means than the use of energy balance to obtain accurate enthalpy measurement
- 3. Further correlation of available data and interpretation of the dimensionless correlations in terms of fundamental mechanisms
- 4. Evaluation of techniques for independently obtaining overall heat of decomposition for materials

Perhaps the most important recommendation of all is the following one suggested by success of the present program in showing that ablation results from different hyperthermal, convective test facilities can be interrelated.

5. Establishment of a round-robin program to determine whether ablation results from facilities having combined radiative and convective test devices can be interrelated.

APPENDICES



#### APPENDICES

The appendices contain detailed information about the test facilities and the data gathered at each, plus new correlations for the data from the Phase I round-robin study. The specific appendices are as follows:

- Appendix A Facility Information and Instrumentation Used for Phase II NASA Round-Robin Ablation Tests
- Appendix B Phase II Tunnel Calibration and Test Data
- Appendix C Model Temperature Data
- Appendix D Summary of Phase II Correlation Data
- Appendix E Dimensionless Correlation of Previous Data

The first two appendices are organized primarily by facility, listed in the following order:

- 1. Gas Dynamic Branch, Ames Research Center-NASA (GDB-Ames)
- 2. Magneto Plasma Dynamics Branch, Ames Research Center-NASA (MPDB-Ames)
- 3. Applied Materials and Physics Division, Langley Research Center-NASA (AMPD-Langley)
- 4. Entry Structures Branch, Langley Research Center-NASA (ESB-Langley)
- 5. Manned Spacecraft Center-NASA (MSC-Subsonic)
- 6. Manned Spacecraft Center-NASA (MSC Supersonic)
- 7. Aerotherm Corporation (Aerotherm)
- 8. Avco Corporation (AVCO)
- 9. Giannini Scientific Corporation (Giannini)
- 10. Martin Company (Martin)
- 11. Space General Corporation (Space General)
- 12. Cornell Aeronautical Laboratory (Cornell)

The abbreviation used to designate the facility in tables and graphs is given in the parentheses following each listing.

The next two appendices are organized primarily by test material. Specifically these are:

- a. Langley Phenolic-Nylon, Scout R/4B (PLL)
- b. Hughes Phenolic-Nylon, H-5 (PLH)
- c. Avcoat 5036-39, HC/G (A)
- d. Modified Purple Blend Silicone, E4A1 (SP)
- e. G. E. Silicone, ESM 1004AP (SG)
- f. Teflon, TFE (T)
- g. High-Density Phenolic-Nylon (P)

The symbols in the parentheses are those used to designate these materials and are part of the model number used in the tables.

#### APPENDIX A

# FACILITY INFORMATION AND INSTRUMENTATION USED FOR PHASE II NASA ROUND-ROBIN ABLATION TESTS

Appendix A tabulates, by facility, a description of each plasma arc jet heater. The tables first describe the arc heater and power supply, then nozzle and test chamber dimensions, as well as the vacuum system and insertion capability. The section of the table on instrumentation describes the instruments or procedures used to measure the parameters indicated.

 $\begin{subarray}{ll} Apendix & A\\ \hline FACILITY INFORMATION AND INSTRUMENTATION USED FOR PHASE II NASA ROUND-ROBIN ABLATION TESTS \\ \hline \end{subarray}$ 

	A- 1	A- 2	A-3	A-4
Facility	NASA—Ames Research Center Gas Dynamics Branch	NASA—Ames Research Center Magneto Plasma Dynamics Branch	NASA-Langley Research Center Applied Materials and Physics Division	NASA—Langley Research Center Entry Structures Branch
Location	Moffett Field, California	Moffett Field, California	Langley Station, Hampton, Virginia	Langley Station, Hampton, Virginia
Tunnel Designation	Planetary Entry Ablation Facility	Low Density Constricted-	20-in, Hypersonic Arc Heated Tunnel	Structures 5-Mw Arc Powered Tunnel
Facility Personnel	B. H. Wick R. B. Pope N. S. Vojvodich	H. A. Stine A. F. Okuno	B. Cocke G. D. Walberg R. E. Middan	W. A. Brooks G. M. Stokes R. D. Brown
Arc Heater				
- Design	Giannini MK-4	NASA, Ames design	NASA, Langley design	NASA, Langley design, 3 phase AC
- Electrode Material	Tungsten cathode, copper anode	Thoriated tungsten cathode, Multiple (24) Copper rod anodes	Copper cathode and anode	Water-cooled copper cathode and anode
- Stabilization	Gas vortex stabilized	Constrictor wall stabilizer	Magnetic stabilization	Magnetic, 1500 gauss
- Input Power	44-130 kw DC	100-750 kw DC	500-2000 kw DC	250-4800 kw AC
- Plenum Pressure	0.20-0.50 atm	0.5-3.3 atm	6.8-34.0 atm	0.1-7.0 atm
- Gas Flow Rate	0.0025 lb/sec through heater 0-0.00125 lb/sec, diluent to plenum	0.004-0.025 lb/sec	0.05-0.80 lb/sec	0.05-1.0 lb/sec air
Power Supply				
- Design	40 kw AC to DC selenium rectifiers	Silicon rectifiers, saturable core reactor control, 3 Mw	DC batteries, 1440 at 2.2 v each	AC, 3 phase, 1380 v
- Make	Miller	Temescal	Exide	1
- Maximum Current	2000 amp	4800 amp	3000 amp	4800 amp
Nozzle				
- Throat Diameter, D		0.50 in.	0.538 in.	1.5 in.
- Exit Diameter, D		6.125 in.	2.0 and 3.3 in.	4.0 in.
- Nozzle Expansion Section	Contoured, free jet	Conical, free jet	Conical, free jet	Conical, free jet
. D to D	;	10.5 in.	7.06 and 11.7 in.	6.75 in.
- D to Model Face	1.625 in.	4.25 in.		0.50 in.
- macii no. oi jec	4.1	0.44	nlle 7	0.7
Test Chamber			: : : : : : : : : : : : : : : : : : : :	
- Diameter	48 in.	36 x 36 in. square	24 x 24 in. square	60 in.
- Length	54 in.	35 in.	38 in.	60 in.
- Cooling	None	Water heat exchanger	None	None
Vacuum System	19,700 ft <sup>3</sup> spheres pumped down to 40 microns Hg with Kinney and Stokes vacuum pumps	5.stage steam ejector, 4500 cfm, 20 microns at no flow	77,000 ft <sup>3</sup> sphere, 3 vacuum pumps, 50 microns at no flow	113,000 ft³ Vacuum sphere
Multiple Model Insertion Capability, Maximum per Run	8	9	2	2

	The state of the s			A- 4
	A-1	A-2	A-3	
Tunnel Instrumentation			,	li di
input rower - Voltage	Voltage divider to C.E.C. recorder	Voltage divider to C.E.C. recorder and G.E. voltmeter	Voltage divider to C.E.C.	Westinghouse wattmeter, Hall effect transducer to Beckman magnetic tape recorder
- Current	Hall effect device to C.E.C. recorder	Shunt to transductor to C. E. C. and Simpson Ammeter	Shunts to C.E.C. recorder	Current transducers to Esterline Angus, indicate only
Power Losses - Water Flow	Fisher-Porter rotometers	Differential pressure across "Rinco" Venturi	Turbine type flow meter	Airtronics turbine meter, Potter frequency converter to Beckman recorder
- Temperature Rise	Delta-T.Co differential temperature transducers to C.E.C. recorder	Differential temperature transducer to C.E.C.	Thermocouples to C.E.C. recorder	Ch-Al thermocouples to Beckman recorder
Test Gas - Composition	23% oxygen - 77% nitrogen by	Air	Air	Air
- Gas Flow Rate	weight Sonic flow orifices	Rotometer and pressure gage	Orifice plates	Laminar flow tube, Boonshaft and Fuchs system to Beckman recorder
- Gas Temperature	Thermometers ahead of orifices	Bimetallic well thermometer		Ch-Al thermocouple to Beckman tape recorder
Pressures - Reservoir	Statham ±5 psid transducer to C.E.C.	Statham transducer to C.E.C.	Strain gage type pressure	Statham transducers to Brown and Beckman recorder
- Nozzle Exit		and an book	Transducers to C.E. recorder	Not measured Statham transducers to
- Test Chamber	U to 1 in, water Dwyer Magnehellic gage			Beckman recorders Alnico transducers,
- Model Stagnation	0.75 in. diameter hemisphere to Statham 0 to 1 psia transducer to C.E.C.	Statham transducer to C.E.C.	-	M.B. Electronics, to Beckman, 3/8-in, water cooled hemisphere pitot probe
Model Temperatures - Facility Pyrometer - Model Front Surface Temperature	1. Honeywell total radiation radiometer, 0.37 in. aper- ture, 0.3.3.8 microns, to	;	Temperature camera (NASA TN D-2660)	Not measured
٠.	2. Inst. Dev. Lab., Pyro 650 to C. E.C., 0.653 microns		-	N
- SRI Radiometer	Located inside test chamber, model viewed with front surface mirror recorded on C.E.C.	Located inside test chamber viewed model with front surface mirror recorded on C.E.C.		Not measured
- Internal Temperatures	Ch-Al to C.E.C.		Ch-Al to C.E.C. recorder	Not measured
SRI Calorimeter	SRI Calorimeter No. 13 to C.E.C. recorder	SRI calorimeter to C.E.C.	SR calorimeter to C.E.C.	SRI calorimeter to Beckman recorder
Facility Calorimeter	Transient slug type, similar to	Transient slug type, similar to	Langley design—thin wall,	Transient—thin wall, 0.030 wall thickness Langley design
2. J. J.	SRI design Hemisnhere cylinder	SHI design Hemisphere cylinder	Flat face cylinder	Hemi sphere
- Surface Material		I mil gold plate on copper	Stainless steel	Stainless steel
- Shroud Diameter - Sensing Diameter	0.75 in. 0.3125 in.	1.25 in. 0.375 in.	1.0 in Multiple thermocouples inside shell	1.5 in Multiple thermo- couples inside hemisphere
Run Time	Exposure time automatically controlled, models protected with water cooled shield, time on C.E.C.	Automatically controlled water cooled rotary sting mounts	From C.E.C. recorder	Automatic model withdrawal, limit switches on model holder to Honeywell Visicorder
Camera	Giannini Scientific, 32 mm movie and pulse framing camera	Kodak Cine Special 16 mm	•	Kodak Cine, 16 mm, 48 fps
	HOVIE and pure accounting comments			

Appendix A (Continued)

				α-<
1000	A-5	Ab	A-(	A-0
SA	NASA Manned Spacecraft Center	NASA Manned Spacecraft Center	Aerotherm Corporation	Avco Corporation
Ho	Houston, Texas	Houston, Texas	Palo Alto, California	Wilmington, Massachusetts
WS	MSC 1-Mw Arc Jet Subsonic	c Tunnel	Aerotherm Arc Plasma Facility	10-Mw Arc Facility
വ്വ	D. H. Greenshields D. J. Tillian	J. E. Grimaud D. J. Tillian	D. T. Flood J. J. Reese	H. E. Hoercher R. W. Freeman J. Duggan
Mo	Modified Giannini	Segmented Constricted Arc,	Aerotherm design	AVCO design
ပိ	Copper	Thoriated tungsten cathode silver-plated copper pin anodes	Thoriated tungsten cathode, copper anode	Carbon cathode, copper anode
Ma	Magnetic field and gas vortex	Vortex stabilized	Gas vortex stabilized	Magnetic field and gas vortex
10	1000 kw DC	250-1000 kw DC	60-1000 kw DC	250-10,000 kw DC
1.0.	1.3 atm 0.02.0.05 lb/sec	(Cathode) 0.1 to 2.0 atm 0.0022-0.044 lb/sec	0.06-8.0 atm 0.002-0.05 lb/sec	0.5-28.0 atm 0.08-1.0 lb/sec
lis fi	1500 kw DC silicon recti- fiers, saturable core reactor control	Silicon rectifiers, saturable core reactor control	2 diesel-electric generators: 1-1000 hp, 1-600 hp	2080 - 12 v, 200 amp hr truck batteries
		A. O. Smith		Willard
20	2000 амр	3000 at 500 v, 1500 at 1000 v, 750 at 2000 v	3000 amp	6000 amp
	Subsonic	0.7813 in.	1.05 and 1.00 in.	0.765 and 1.25 in.
· · · · · ·	3.0 in.	5.8 in.	2.98 and 3.50 in.	1.178 and 1.25 in.
- Nozzle Expansion Section   Su	Subsonic	Conical	freejet and con	Conical
i	1.5 in.	5.2 in.	12.0 8.5 in.	1.0 in.
	1.5 and 2.0 in.	4.0 in.	•	3.0 in.
Su	Subsonic	3.8	3.0 and 3.2	2.2
S.	None	72 in.	42 in.	Free jet, no test chamber used for this program
	,	96 in.	180 in	Approximately 48 in.
		Chamber cooling - circulated air	Cooled diffuser and heat exchanger	Water-cooled
N	None	4-stage steam ejector. 0.01 lb/sec at 0.001 atm, 0.035 lb/sec at 0.01 atm, and 0.5 mm Hg at no flow	5-stage steam ejector, 0.036 lb/sec at 200 microns, 0.004 lb/sec at 50 microns, 30 microns Hg at no flow	None used
Multiple Model Insertion Capability, Maximum per Run		2	5	1

	5-4	A.6	A-7	8 - 8
Tunnel Instrumentation Input Power - Voltage	Nobotrol transducers to Bristol recorder and to	Nobotrol transducers to Bristol recorder and Systems Engineering	Vidar voltage to frequency converter 260	Voltage divider to C.E.C. recorder
	digital system My shunt to Bristol	My shunt to Bristol and S.E.L.	Vidar voltage to frequency converter 260	Shunt to C.E.C. recorder
Power Losses - Water Flow	Hydropoise flow trans- ducer to frequency converter and recorder.		ASME orifice and manometer, plus Statham AP transducer to Vidar 260	Not measured
- Temperature Rise	metri tabe record or most	Thermocouples and △T meter Ch-Al thermocouple	Delta T differential temperature transducer to Vidar 260	Not measured
Test Gas - Composition	Nitrogen plus oxygen to equal air	Nitrogen plus oxygen to equal	Nitrogen plus oxygen to equal air	Air
- Gas Flow Rate	Choked orifices, electrical pressure trans-	Choked orifices, Statham transducers to S.E.L.	Fisher Porter rotometers plus orifice and Statham transducers	Fisher Porter flow meters and sonic flow orifices
- Gas Temperature	Thermocouple	Ch-Al thermocouple	I-C Thermocouple to M-H indicating pot.	
Pressures - Reservoir	Not measured	Statham transducer (0-15 psia) to S.E.L.	Statham abs. pressure transducer to C.E.C. 5-119 recorder	Electrical pressure transducer to C.E.C. recorder
- Nozzle Exit	Not measured	Statham transducer (0-1 psia) to S.E.L.	Wallace Tiernan gages	Not measured
- Test Chamber - Model Stagnation	Not measured Not measured	Statham transducer to S.E.L. Statham transducer (0.5 psia) to S.E.L. Wallace-Tiernan gages for checks	Wallace Tiernan gages Statham diff, pressure transducer to C.E.C. 5-119 recorder, 0.375 in diam- eter flat face water cooled pitot probe	Uncooled 1.25 in. diameter pitot probe, to C.E.C. pressure transducer, to C.E.C. recorder
Model Temperatures - Facility Pyrometer, Model Front Surface Temperature	Pyro-optical pyrometer Barnes R4D radiometer	Optical pyrometer-0.65 microns	Infrared Industries-TD9CH 0.8 ± 0.015 microns	Instrument Development Lab. recording pyrometer, 0.653 microns
- SRI Radiometer	SRI radiometer to S.E.L.	Not measured	SRI radiometer to C. P. Inst. 850 m.v. recorder	Not measured
- Internal Temperatures	Ch-Al to tape recorder	Not measured	Ch-Al to C.E.C. 5-119 recorder	Not measured
SRI Calorimeter Facility Calorimeter	Recorded on S.E.L.	Recorded on Bristol and S.E.L. Hy-Cal Engineering (120-1000 Btu/ft2sec)		
- Type	Hy-Cal Engineering	High Temperature, Lab. (120-1000 Btu/ft sec)	Hy-Cal Engineering	AVCO null point transient calorimeters
- Shape	Flat face	Flat face	Flat faced cylinder	1. Flat faced 2. Hemisphe- cylinder rical cone
- Surface Material	Constantan plus carbon coating	Constantan disc plus carbon coating	Constantan plus carbon black	Copper Copper
- Shroud Diameter	1.0 in. calorimeter in 1.25 in. shroud	1.0 in. copper body in 1.25 in. graphite shroud	1.0 in.	1.25 in. 1.0 in.
- Sensing Diameter	0.15 in.	0.060 in.	0.18 in.	0.375 in. 0.25 in.
Run Time	Microswitch on sting to S.E.L.	Microswitches on sting to S.E.L	C.E.C. recorder and stopwatch	From C.E.C. recorder and pressure and current traces
Camera	Milliken	Milliken	Triad Photosonics, 1B, 16 mm	Bolex Reflex, 300 mm lens 32 fps

Appendix A (Concluded)

	9-A	A- 10	A-11	A-12
Facility Location Tunnel Designation	Giannini Scientific Corporation Santa Ana, California 1-Mw Hyperthermal Test Facility	Martin Company Baltimore, Maryland Plasma Arc Laboratory, Facility B	Space General Corporation El Monte, California Electro-Thermal Facility	Cornell Aeronautical Laboratory Buffalo, New York Wave Superheater Hypersonic Tunnel
Facility Personnel	J. P. Todd E. Muchlberger	A. Guido G. J. Schmidt G. Guenterberg R. A. Woodward	S. L. Grindle M. W. Searcy	J. Carpenter R. Clements K. W. Graves
Arc Heater - Design - Electrode Material	Giannini design Tungsten cathode, copper anode	Thermal Dynamics F 5000 Tungsten cathode, copper anode	Space General design Tungsten cathode, copper anode	Device consists of 288 (0.6 x 1.5-in.) shock tubes mounted on periphery of a rogar taing drum. Hot helium driver gas is supplied to the shock
- Stabilization	Gas vortex stabilized	Gas vortex and magnetic field at anode	Gas vortex and magnetic field	tubes from pebble bed heater. Energy is transferred to high pressure air, and forms steady
- Input Power - Plenum Pressure - Gas Flow Mate	35-1000 kw DC 0.02-0.36 atm 0.0005-0.01 lb/sec	40-1500 kw DC 0.03-68 atm 0.0015-0.6 1b/sec	25-1500 kw DC 0.01.40 atm 0.0001-0.2 lb/sec	supersonic jet. 5 lb/sec
Power Supply - Design	Six 150 kw, silicon rectifiers	3 phase full ware recti- fier, saturable reactor control	Two 750 kw AC to DC silicon recti- fiers, moving coil control	Does not apply
- Make - Maximum Current	A. O. Smith 3000 amp	A. O. Smith 4000 amp at 500 v, 2000 amp at 1000 v	Glenn Pacific 3500 amp	
Nozzle - Throat Diameter, Dt - Exit Diameter, De - Nozzle Expansion Section	1.0 in. 3.0 in. Contoured, free jet	1.245 in. 3.0 in. Conical, free jet	1.0 in. 3.0 in. Contoured nozzle, free jet	Free jet has dimensions of 1.5 in. height x 0.6 in. width at jet exit (rotor face).
- Dt to D - D to Model Face - Mach No. of jet	6.18 in. 5.0 in. 3.0	6.625 in. 2.7	2.25 in. 3.0	2.25 and 4.0 in. 2.7 and 3.6
Test Chamber - Diameter - Length - Cooling	30 in. 72 in. Water-cooled chamber and vacuum line	48 in. 96 in. Water-cooled test chamber, andels and inscruments held in separate chamber before insertion in stream	36 in. 84 in. Chamber water-cooled and diffuser heat exchanger	None used for these tests
Vacuum System	Kinney vacuum pumps, 9000 cfm, 0.2 mm Hg at no flow.	5-stage steam ejector, 95,000 scfm, 2 x 10-5 atm at no flow.	2 Roots blowers followed by Stokes vacuum pump, 14,000 cfm, 0.1 mm Hg at no flow	None used
Multiple Model Insertion Capability, Maximum per Run	ന	2	4	1

	A-9	A-10	A+11	A-12
Tunnel Instrumentation Input Power - Voltage - Current	Westinghouse PX 161 Westinghouse PX 161	Greibach Instrument Mo 700 Greibach Instrument Mo 700	Westinghouse PX 161 Westinghouse PX 161	A computation is made for the shock process as it occurs inside the rotor using the driver helium
Power Losses		Date of Distance	Datter timbine meteore to a pertronic	and charge all input temperatures and pressures.
- Water Flow	Hydropoise turbine meter to Erie electronic counter	meters to Cox counter	Counter	
- Temperature Rise	I-C thermocouples to Brown multipoint recorder	Rosemont Engineering platinum resistance thermometer	C-C thermocouples to DeVar recorders	
Test Gas				
- Composition	Nitrogen plus oxygen to equal	Nitrogen plus oxygen to equal air	79% Nitrogen, 21% oxygen by volume	Air
- Gas Flow Rate	Standard orifice plates to Heise gage	Cox Instruments turbine meter and counter	Critica I flow orifices to Heise gages	Not measured
- Gas Temperature	I-C thermocouples to Brown	Tri-R thermister		Thermocouple monitors
Pressures - Reservoir	Wallace and Tiernan gages	Hg manometer, Heise gage	Wallace and Tiernan gages	100 atm maximum C.E.C. transducer and recorder
- Nozzle Exit	Wallace and Tiernan gages	Transsonic differential	Wallace and Tiernan gages	Not measured
- Test Chamber	Wallace and Tiernan gages	ruments - 77MXRP-3	Wallace and Tiernan gages	Not measured
- Model Stagnation	Wallace and Tiernan differen- tial pressure gage. Statham pressure transducer to Texas Instruments recorder	MKS differential pressure, 0.625 in. diameter water cooled pitot probe	Statham Transducer to DeVar recorder, 0.025 in. diameter water cooled pitot probe	10.85 atm, C.E.C. trans- ducer and recorder
Model Temperatures - Facility Pyrometer, Model Front Surface Temperature	1. Infrared Industries, Thermodot TIGBT Recording Pyrometer (1.6-2.3 microns) to Texas Instruments F4W	Instrument Development Lab., Pyro 650	1. Leeds Northrup optical Pyrometer 8632F	Infrared Industries, Thermodot TDA-6
	recorder  2. LN optical Pyrometer 8622C (0,655 microns)			
- SRI Radiometer	SRI radiometer to Midwest 1500B oscillograph	SRI radiometer to Offner oscillograph	SRI radiometer to LN Speedomax H recorder	Not measured
- Internal Temperatures	Ch.Al thermocouples to Midwest 1500B oscillograph	Ch-Al thermocouples to Systrac 160E2	Ch-Al to Texas Instruments F4W recorder	Not measured
SRI Calorimeter	SPI calorimeter to Midwest oscillograph			SRI calorimeter to C.E.C. recorder
Facility Calorimeter - Type	1. Giannini 2. Giannini steady state transient water temp- slug	l. Thermogage, 2. Martin, Gardon sient	HyCal Engineering asymptotic calorimeter	Cornell design
- Shape	Hemispherical Flat face	Flat face Flat cylinder face	Flat face cylinder	Hemispherical
- Surface Material - Shroud Diameter	Copper Copper 0.625 in. 1.25	Constantan Copper 1.25 in.	Constantan 1.25 in.	Copper 0.50 in.
- Sensing Diameter	0.625 in. 0.25 in.	0.125 in. 0.25 and 0.625	0.10 in.	0.125 in.
Run Time	Stop watch plus electric timer with switch on sting	Automatic run time record and arc shut off	Stop watch plus electronic timer activated by microswitch on sting	Test is started with model in place, run termination is automatically controlled
Camera	Bolex H16 movie camera	None used	Bolex H16 reflex movie camera	Two Photosonics

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#### APPENDIX B

#### PHASE II TUNNEL CALIBRATION AND TEST DATA

This appendix contains separate tables of the data reported by each participating facility, plus the measurement data on all models that were determined at the Institute. The latter data constitute the last five columns of the tables.

The calibration runs were assigned numbers by the Institute so that they could be identified in the text. Pertinent remarks applicable to specific columns of data are indicated in the footnotes.

Table B-1

## TUNNEL CALIBRATION AND TEST DATA REPORTED BY GAS DYNAMICS BRANCH, AMES RESEARCH CENTER—NASA

Ref: Letter Report by C. A. Syvertson, Ames Research Center, December 2, 1966

	MODEL	AVER	AGE ENTH	LPY	HEAT	TRANSI	ER RA	ATE	MODEL	PLENUM
43.5	NO.	h <sub>t</sub>	(Btu/1b	) .	$\mathfrak{q}_{CW}$	(Btu/f	t <sup>2</sup> sec	;)	STAGNATION PRESSURE	PRESSURE P <sub>t</sub>
						Calorim	eter		P <sub>t</sub> <sub>2</sub>	
		4.5	(0)	(8)		acility		SRI	(atm)	(atm)
		(1)	(2)	(3)	(4)	(5)	(6)			
Langley Phenolic-Nylon Scout R/4B	PLL54 PLL57 PLL96	10,430 10,430 10,670	10,170 10,170 10,670		143 143 145	67.6 67.6 68.5	84 84 89	77 77 85	0.0106 0.0106 0.0109	0.235 0.235 0.248
	PLL58 PLL59	25,700 25,700	22,830 22,830		240 240	113.5 113.5	128 128	118 118	0.0106 0.0106	0.300 0.300
	PLL60 PLL61	15,360 15,360	16,590 16,590		300 300	141.9 141.9	190 190	178 178	$0.0182 \\ 0.0182$	0.435 0.435
Avcoat 5026-39	A53 A54 A61 A93	9,984 9,984 10,707 10,449	10,170 10,170 10,670 10,750		147 147 140 132	69.5 69.5 66.2 62.4	78 78 85 86	80 80 81 89	0.0108 0.0108 0.0109 0.0108	0.234 0.234 0.248 0.250
	A55 A56	23,920 23,920	22,120 22,120		259 259	122.5 122.5	143 143	121 121	$0.0102 \\ 0.0102$	0.314 0.314
	A57 A60	15,330 15,330	16,670 16,670		349 349	165. 165.	183 183	177 177	0.0185 0.0185	0.438 0.438
Modified Purple Blend Silicone E4Al	SP48 SP49 SP96	10,134 10,134 10,678	10,170 10,170 10,670		148 148 133	70. 70. 62.9	80 80 85	73 73 87	0.0105 0.0105 0.0109	0.234 0.234 0.248
	SP51 SP52	23,340 23,340	21,850 21,850				$\frac{146}{146}$	114 114	0.0097 0.0097	0.312 0.312
	SP50	15, 970	16,480		338	159.8	196	172	0.0185	0.431
G. E. Silicone ESM 1004AP	SG35 SG39	10,153 10,749	10,069 10,670		152 146	71.9 69.	81 86	89 87	0.0106 0.0106	0.231 0.248
	SG36 SG37	24,010 24,010	22,120 22,120		259 259	122.5 122.5	138 138	120 120	0.0098 0.0098	0.314 0.314
	SG38 SG44	15,510 15,510	16,590 16,590		358 358	169.3 169.3	183 183	170 170	0.0183 0.0183	0.435 0.435
Hughes Phenolic-Nylon H-5	PLH6 PLH98	10,216 10,322	10,170 10,730		$144 \\ 142$	68.1 67.1	78 89	78 85	0.0106 0.0111	0.234 0.248
	PLH42 PLH43	24,970 24,970	22,750 22,750		273 273	129. 129.	147 147	125 125	$0.0102 \\ 0.0102$	0.316 0.316
	PLH24 PLH50	15,870 15,870	16,620 16,620		345 345	163.1 163.1	188 188	166 166	0.0185 0.0185	0.436 0.436
Calibration Runs	C1 C2 C3 C4 C5 C6 C7 C8 C9 C10 C11	11,254 15,571 18,025 22,120 24,023 25,580 8,194 12,770 16,530 10,594	11,030 15,400 17,500 20,220 21,580 23,490 8,100 12,690 17,090 10,390	6,915 9,210 9,770 11,760 11,410 11,740 5,800 8,070 9,720 6,340 11,180	143 173 196 211 225 224 160 265 352 135 263	67.7 81.5 92.8 99.9 106.5 115.5 75.8 125.5 166.7 63.9 124.6	82 103 120 134 142 148 94 154 208 84 154	82 96 108 117 124 134 89 150 183 80 126	0.0105 0.0098 0.0097 0.0097 0.0094 0.0101 0.0140 0.0162 0.0189 0.0106 0.0099	0.235 0.260 0.275 0.290 0.301 0.312 0.326 0.329 0.428 0.240 0.309

<sup>(1)</sup> Enthalpy measured by energy balance method, average enthalpy at reservoir entrance, i.e., arc heater exit.

<sup>(2)</sup> Enthalpy of free jet at test position by frozen sonic flow method and using Z. Ref: NASA TN D2233 and NASA TR R-50. Also see Column (7).

<sup>(3)</sup> Enthalpy by energy balance method, average enthalpy at nozzle exit.

<sup>(4)</sup> Ames transient calorimeter, 0.75-in. diameter hemisphere cylinder, copper slug 0.3125-in. diameter supported in shroud with sapphire microspheres.

<sup>(5)</sup> Calorimeter described under (4), calculated to 1.25-in. diameter flat face stagnation value:  $q_{FAC} = 1.112 \ q_{FAC} \left[0.75/(3.3 \times 1.25)\right]^{0.5}$ . The term 1.112 corrects the average q over sensing area to the stagnation point value.

<sup>(6)</sup> Calorimeter described under (4), sprayed with Teflon coating to give noncatalytic surface.

7.	GAS FLOW RATE (1b/sec)		FRONT SUPERATURE	2	RUN TIME (sec)	CORE WEIGHT LOSS (g)	CORE CHAR WEIGHT (g)	CORE RECESSION (in.)	CHAR THICKNESS (in.)	CHAR DENSITY (1b/ft <sup>3</sup> )
(7)		Facility (8)	(9)	SRI						
1.56	0.0025	2,850	2,580	2,559	19.3	0.202	0.076	0.023	0.065	14.5
1.56	0.0025	2,990	2,710	2,785	38.6	0.349	0.107	0.058	0.093	14.3
1.57	0.0025	3,030	2,810	2,846	75.4	0.587	0.160	0.107	0.122	16.3
2. 22	0.0025	3,260	2,960	2,918	15.6	$0.215 \\ 0.414$	0.079	0.026	0.069	14. 2
2. 22	0.0025	3,150	3,200	3,226	38.2		0.134	0.064	0.018	14. 1
1.86 1.86	0.00375 0.00375	3,660 3,650	3, 280 3, 510	3,247 3,432	11.2 28.2	0.153 0.335	$0.084 \\ 0.150$	0.019 0.052	$0.074 \\ 0.122$	14.1 15.3
1.55	0.0025	2,870	2,560	2,600	19.4	0.174	0.110	0.024	0.079	17.3
1.55	0.0025	2,860	2,600	2,590	19.4	0.203	0.118	0.028	0.087	16.8
1.57	0.0025	2,890	2,680	2,682	38.2	0.294	0.181	0.052	0.121	18.5
1.56	0.0025	2,920	2,650	2,692	75.5	0.445	0.264	0.101	0.167	19.6
2.15	0.0025	3,360	3,080	3,000	15.5	0.220	0.110	0.027	0.081	16.9
2.15	0.0025	3,550	3,230	3,196	38.4	0.366	0.188	0.078	0.138	16.9
1.86	0.00375	3,520	3,270	3,226	11.2	0.177	0.120	0.027	0.077	19.3
1.86	0.00375	3,660	3,420	3,350	28.2	0.369	0.157	0.085	0.120	16.2
1.55	0.0025	2,620	2,360	2,446	19.5	0.153	0.110	+0.023	0.093	14.7
1.55	0.0025	2,680	2,480	2,446	38.4	0.249	0.145	+0.065	0.155	11.6
1.57	0.0025	2,550	2,350	2,323	75.5	0.375	0.200	+0.052	0.205	12.1
2.14	0.0025	2,910	2,760	2,764	15.4	$0.146 \\ 0.263$	0.126	+0.031	0. 102	15.3
2.14	0.0025	3,080	3,070	2,826	40.6		0.198	+0.038	0. 166	14.8
1.87	0.00375	3,330	3, 130	3,144	28.1	0.287	0.160	0.019	0.116	17.1
1.55	0.0025	2,640	2,390	2,477	38.5	0.147	0.345	+0.007	0.134	31.6
1.57	0.0025	2,610	2,390	2,405	75.5	0.387	0.458	+0.005	0.199	28.6
2. 15 2. 15	0.0025 0.0025	2,990 3,210	2,770 2,880	2,785 2,846	15.5 38.2	0.097	0.207	+0.001 0.069	0.086 0.112	29.9 28.5
1.87	0.00375	3,290	2,910	2,898	11.3	$0.187 \\ 0.422$	0.142	0.035	0.055	32.0
1.87	0.00375	3,270	2,880	2,846	28.3		0.184	0.118	0.060	38.1
1.55	0.0025	2,890	2,690	2,723	38.4	0.318	0.117	0.044	0.097	15.0
1.56	0.0025	3,090	2,850	2,857	75.3	0.653	0.165	0.115	0.132	15.5
2.18	0.0025	3,260	3,020	2,949	15.6	0.198	0.081	0.017	0.071	14. 2
2.18	0.0025	3,500	3,250	3,206	38.5	0.414	0.135	0.056	0.120	14. 0
1.87	0.00375	3,470	3,200	3, 165	11.4	0.177	0.071	0.011	0.067	13.2
1.87	0.00375	3,470	3,200	3, 165	11.7	0.201	0.070	0.012	0.066	13.2
1.62 1.87 1.98 2.10 2.15 2.23 1.45 1.67 1.92 1.56 2.19	0.0025 0.0025 0.0025 0.0025 0.0025 0.0025 0.00375 0.00375 0.00375 0.0025									

<sup>(7)</sup>  $Z = 1 + \alpha$ ,  $\gamma = (4 + 3Z)/(4 + Z)$ ,  $\sqrt[*]{P_{t_1}}A^* = c \{\gamma[2/(\gamma + 1)]^{(\gamma+1)/(\gamma-1)}\}^{0.5}(ZT_t)^{-0.5}$ ;  $T_t$  is calculated from the preceding equation and is used to get enthalpy under (2). Ref: JANAF Interim Thermochemical Tables, Dow Chemical Co., Dec. 31, 1960.

<sup>(8)</sup> Instrument Development Lab, Pyro 650 recording pyrometer,  $\epsilon = 1$ ,  $T_T = \left[ (1/T_B) + 2.52 \times 10^{-5} \ln \epsilon_{\lambda} \right]^{-1}$ .
(9) SRI radiometer, 0.375-in. aperture,  $\epsilon = 1$ ,  $T_T = T_B/\epsilon^{0.25}$ .  $T_T$  is true temperature and  $T_B$  is brightness temperature.

Table B-2 TUNNEL CALIBRATION AND TEST DATA REPORTED BY MAGNETO PLASMA DYNAMICS BRANCH, AMES RESEARCH CENTER-NASA

Ref: Data Reported by A. Okuno, MPDB, Ames Research Center

	MODEL NO.		RAGE ALPY tu/1b)		HEAT TR			MODEL STAGNATION PRESSURE	PLENUM PRESSURE
		(1)	(2)	Faci	Calo lity (4)	rimeter	SRI	P t 2 (atm)	P 1 (atm)
Langley Phenolic-Nylon Scout R/4B	PLL70 PLL69 PLL87	6,736 6,736 6,736	12,664 12,664 12,664	160.3	103.8	117.1		0.00572 0.00572 0.00572	1.035 1.035 1.035
· -	PLL72 PLL71 PLL89	12,508 12,508 12,508	19,638 19,638 19,638	281.5	182.4	212.2		7.7	0.986 0.986 0.986
Avcoat 5026-39	A7 2 A7 1 A8 4	6,973 6,973 6,973	13, 109 13, 109 13, 109	162.5	105.3	105.3		0.00582 0.00582 0.00582	0.990 0.990 0.990
	A76 A75 A85	12,320 12,320 12,320	19,342 19,342 19,342	313.	202.8	215. 2		0.00947 0.00947 0.00947	0.997 0.997 0.997
Modified Purple Blend Silicone E4Al	SP66 SP65 SP85	6,451 6,451 6,451	12,128 12,128 12,128	150.6	97.6	103.5			0.990 0.990 0.990
	SP68 SP67 SP89	12,561 12,561 12,561	19,721 19,721 19,721	286.9	185.9	221.0		0.00847	$1.014 \\ 1.014 \\ 1.014$
G. E. Silicone ESM 1004AP	SG56 SG55 SG51	6,903 6,903 6,903	12,978 12,978 12,978	150.6	97.6	104.4			0.993 0.993 0.993
*	SG58 SG57 SG53	12,253 12,253 12,253	19,232 19,232 19,232	280.5	181.8	210.8		0.00827 0.00827 0.00827	1.007 1.007 1.007
Hughes Phenolic-Nylon H-5	PLH62 PLH61 PLH97	6,442 6,442 6,442	12,111 12,111 12,111	142.8	92.5	92. 2			1.049 1.049 1.049
	PLH65 PLH64 PLH63	12,162 12,162 12,162	19,094 19,094 19,094	293.2	190.	223.9		0.00787 0.00787 0.00787	1.017 1.017 1.017
Teflon	T117 T116	6,177 6,177	11,613 11,613	147.9	95.9		• :	0.00597 0.00597	0.983 0.983
	T115 T112	12,052 12,052	18,922 18,922	294.	190.5	221.1		. مرت	0.968 0.968
Phenolic-Nylon (75 lb/ft <sup>3</sup> )	P11A6 P11A5	7,025 7,025	13,207 13,207	170.6	110.5	116.1			1.028 1.028
(. 5 = 2) 2 3	P12B3 P12A7	11,927 11,927	18,725 18,725	289.6	187.7	223.2		0.00812 0.00812	0.990 0.990
Calibration Runs	C1 C2	12,214 12,289	19,176 19,294			224.9 229.2	231.3 140.7 <sup>(5)</sup>		4
	C3	6,919	13,007			109.1	64.6 (6)		

- (1) Enthalpy measured by energy balance method.
- (2) Enthalpy calculated by facility from facility heat flux data: h = 24 q<sub>FAC</sub>(R<sub>eff</sub>/P<sub>t2</sub>)<sup>0.5</sup>.
   (3) Ames transient slug calorimeter, 1.25-in. diameter hemisphere cylinder, gold-plated copper slug 0.375-in. diameter supported with sapphire microspheres.
- (4) Facility calorimeter results under (3) corrected by facility to 1.25-in. flat face stagnation condition with relation:  ${}^{\circ}_{FAC} = {}^{\circ}_{FAC} [(0.84 \times 1.25)/(2 \times 1.25)]^{0.5}$ .
- (5) SRI calorimeter as received by facility.
- (6) SRI calorimeter sprayed with Teflon coating to give noncatalytic surface.

TEST CHAMBER PRESSURE (atm)	POWER TO ARC (Mw)	AIR FLOW RATE (1b/sec)	ARGON CATHODE SHIELD FLOW RATE (lb/sec)	$\begin{array}{c} \text{MAXIMUM} \\ \text{FRONT} \\ \text{SURFACE} \\ \text{TEMPERATURE} \\ T_{FS} \ \epsilon \ = \ 1 \\ \text{(°F)} \end{array}$	RUN TIME (sec)	CORE WEIGHT LOSS (g)	CORE CHAR WEIGHT (g)	CORE RECES- SION (in.)	CHAR THICK- NESS (in.)	CHAR DENSITY (1b/ft <sup>3</sup> )
0.00072 0.00072 0.00072	0.0924 0.0924 0.0924	0.01018 0.01018 0.01018	0.000403 0.000403 0.000403		12.4 24.6 50.6	0.144 0.231 0.400	0.063 0.089 0.143	0.008 0.019 0.064	0.054 0.079 0.117	14.5 14.0 15.2
0.00072 0.00072 0.00072	0.1462 0.1462 0.1462	0.0079 0.0079 0.0079	0.000494 0.000494 0.000494		6.2 12.0 24.8	0.122 0.193 0.321	0.051 0.078 0.129	0.012 0.020 0.038	0.047 0.071 0.109	13.5 13.6 14.7
0.00072 0.00072 0.00072	0.0958 0.0958 0.0958	0.01015 0.01015 0.01015	0.000403 0.000403 0.000403		12.2 25.1 50.2	0.112 0.196 0.352	0.087 0.129 0.207	0.005 0.017 0.069	0.067 0.098 0.143	16.1 16.3 18.0
0.00072 0.00072 0.00072	$0.147 \\ 0.147 \\ 0.147$	0.00806 0.00806 0.00806	0.000494 0.000494 0.000494	2,900 3,100 3,330	$6.1 \\ 12.0 \\ 24.8$	0.098 0.158 0.301	0.073 0.112 0.164	0.004 0.011 0.045	0.051 0.070 0.125	17.8 19.8 16.3
0.00072 0.00072 0.00072	0.0908 0.0908 0.0908	$\begin{array}{c} 0.01021 \\ 0.01021 \\ 0.01021 \end{array}$	0.000403 0.000403 0.000403		12.7 24.7 50.5	0.102 0.192 0.307	0.080 0.142 0.232	+0.018 +0.027 +0.018	0.069 0.163 0.170	14.4 10.8 16.9
0.00072 0.00072 0.00072	0.1487 0.1487 0.1487	0.00809 0.00809 0.00809	0.000494 0.000494 0.000494	2,700 2,900 2,980	6.2 12.1 25.0	0.115 0.161 0.278	0.057 0.111 0.163	+0.043 +0.029 0.004	0.079 0.093 0.121	9.0 14.8 16.7
0.00072 0.00072 0.00072	0.0964 0.0964 0.0964	0.01037 0.01037 0.01037	$\begin{array}{c} 0.000403 \\ 0.000403 \\ 0.000403 \end{array}$		12.0 24.8 50.0	0.082 0.106 0.187	0.158 0.238 0.392	+0.005 +0.019 +0.005	0.072 0.121 0.171	27.2 24.4 28.4
0.00072 0.00072 0.00072	0.148 0.148 0.148	0.00821 0.00821 0.00821	0.000494 0.000494 0.000494	2,700 2,800 2,870	6.2 12.1 24.9	0.071 0.161 0.352	0.121 0.144 0.174	0.005 0.020 0.067	0.049 0.071 0.081	30.6 25.2 26.7
	0.0914 0.0914 0.0914	$\begin{array}{c} 0.01048 \\ 0.01048 \\ 0.01048 \end{array}$	$\begin{array}{c} 0.000403 \\ 0.000403 \\ 0.000403 \end{array}$		12.0 $24.9$ $50.2$	0.140 0.255 0.451	0.049 0.096 0.129	0.008 0.022 0.042	0.053 0.087 0.125	11.5 13.7 12.8
	0.1416 0.1416 0.1416	0.00806 0.00806 0.00806	0.000494 0.000494 0.000494		$6.1 \\ 12.1 \\ 24.9$	0.107 0.178 0.303	0.049 0.067 0.115	0.004 0.010 0.027	0.044 0.069 0.111	13.8 12.1 12.9
0.00072 0.00072	0.0875 0.0875	0.01045 0.01045	0.000358 0.000358		11.9 50.2	0.214 1.037		0.017 0.087		
0.00072 0.00072	0.1447 0.1447	0.00806 0.00806	0.000474 0.000474		$\frac{6.1}{24.9}$	0.169 0.736		0.013 0.061		
0.00072 0.00072	0.0964 0.0964	$0.01012 \\ 0.01012$	0.000358 0.000358		12.3 50.5	0.165 0.570	0.031 0.121	0.001 0.023	0.019 0.070	20.2 21.5
0.00072 0.00072	$0.142 \\ 0.142$	0.00802 0.00802	0.000494 0.000494		6.2 24.9	$0.127 \\ 0.411$	0.021 0.087	$0.001 \\ 0.015$	0.014 0.053	$\begin{array}{c} 18.6 \\ 20.4 \end{array}$
0.00072 0.00072	0.1462 0.1481	0.00805 0.00806								
0.00072	0.0968	0.01045								

Table B-3

TUNNEL CALIBRATION AND TEST DATA REPORTED BY APPLIED MATERIALS AND PHYSICS DIVISION, LANGLEY RESEARCH CENTER—NASA

Ref: Letter Report by P. F. Korycinski, Langley Research Center, January 9, 1967

	MODEL NO.	TOTAL ENTHALPY h <sub>t</sub> (Btu/1b)	HEAT TRANSF q <sub>CW</sub> (Btu/ Caloria	ft <sup>2</sup> sec)	MODEL STAGNATION PRESSURE P
		(1)	Facility	SRI	(atm)
Langley Phenolic-Nylon Scout R/4B	PLL30 PLL32 PLL93 PLL33	4,900 4,900 4,900 4,900 4,900		267 261 256 271	0.284 0.284 0.284 0.284
	PLL29 PLL31	9,700 9,700	:	460 592	$\substack{0.293\\0.293}$
	PLL34 PLL35	9,700 9,700		1,150 1,080	0.735 0.735
Avcoat: 5026-39	A30 A90 A34	4,900 4,900 4,900		260 280 269	$0.284 \\ 0.284 \\ 0.284$
	A29 A33 A31 A32	9,700 9,700 9,700 9,700		479 541 531 541	0.293 0.293 0.293 0.293
and the second s	A45	9,700		1,090	0.735
Modified Purple Blend Silicone E4Al	SP30 SP93 SP32	4,900 4,900 4,900		260 273 275	$0.284 \\ 0.284 \\ 0.284$
	SP29 SP31	9,700 9,700		481 539	$0.293 \\ 0.293$
	SP34 SP35 SP33	9,700 9,700 9,700		1,065	$0.735 \\ 0.735 \\ 0.735$
G. E. Silicone ESM 1004AP	SG22 SG49 SG24	4,900 4,900 4,900		259 263 249	$0.284 \\ 0.284 \\ 0.284$
	SG21 SG23	9,700 9,700		478 551	0.293 0.293
	SG25	9,700		1,110	0.735
Hughes Phenolic-Nylon H-5	PLH26 PLH93 PLH28	4,900 4,900 4,900		260 250 293	$0.284 \\ 0.284 \\ 0.284$
	PLH25 PLH27	9,700 9,700		492 607	$\begin{array}{c} 0.293 \\ 0.293 \end{array}$
	PLH29 PLH30	9,700 9,700		1,134 1,145	0.735 0.735
Teflon	T124 T126	9,700 9,700		516 1,055	0.293 0.735
Phenolic-Nylon (751b/ft <sup>3</sup> )	P11A2	9,700		524	0.293
rachoric-myron (1315/10 /	P11A2 P11A3 P11A4	9,700 9,700 9,700		524  996	0.735 0.735
		,	(5)		

<sup>(1)</sup> Enthalpy calculated by facility from  $q_{\mathrm{SRI}}$  and  $P_{\mathrm{t}_2}$  using Fay-Riddell relation.

<sup>(2)</sup> Model front surface temperature measured with facility photographic pyrometer. Ref: NASA TN D-2660.

<sup>(3)</sup> SRI radiometer located outside tunnel; viewed model through 1 in, thick glass window; radiation from model redirected inside tunnel with mirror, 30° to mirror surface. Data not corrected for window or mirror losses.

MAXIN	IUM FRONT TEMPERATI		RUN TIME (sec)	CORE WEIGHT	CORE CHAR	CORE RECESSION	CHAR THICKNESS	CHAR DENSITY
T <sub>FS</sub>	<i>ϵ</i> = 1	(°F)	(sec)	LOSS (g)	WEIGHT (g)	(in.)	(in.)	(lb/ft <sup>3</sup> )
Facility		SRI						
(2)		.( 3:)						
4,100 4,010 3,940 3,960		2,930 2,950 2,900 2,915	14 30 30 60	0.288 0.592 0.510(4) 1.287	0.115 0.131 0.115 0.115	0.042 0.126 0.137 0.381	0.091 0.111 0.099 0.097	15.7 14.6 14.4 14.7
4,810		3,620 3,730	11 30	0.280 0.835	$0.141 \\ 0.164$	0.036 0.181	$0.103 \\ 0.151$	17.0 13.5
5,170 5,230		3,540 4,245	10 20	0.484 1.021	0.127 0.103	0.111 0.300	0.098 0.080	16.1 16.0
3,920 3,900	-	2,380 1,845 2,840	14 20 30	0.350 0.510 0.823	0.112 0.085 0.105	0.089 0.177 0.276	0.077 0.061 0.062	18.1 17.3 21.0
4,600 4,500		3,520 3,455 3,430	11 20 30 30	0.417 0.822 1.243 1.542	0.075 0.077 0.087 0.055	0.106 0.281 0.377 0.453	0.059 0.049 0.070 0.062	15.8 19.5 15.4 11.0
4,680		3,400	5	0.566	0.028	0.242	0006	
3,580 3,600 		2,560 2,505 2,510	14 30 60	0.212 0.548 1.012	$0.151 \\ 0.150 \\ 0.135$	+ 0.004 0.070 0.214	0.097 0.117 0.097	19.3 16.0 17.3
4,070 4,140		3,075 2,910	11 30	0.424 1.386	0.056 0.032	0.082 0.412	0.049 0.022	14.2 18.1
3,940 3,980 4,170		2,580 2,970	5 5 10	0.870 0.727 1.228	0.005 0.011 0.008	0.253 0.190 0.382	0.004 0.022 0.003	15.5
3,830 3,740		2,660 2,570 2,540	14 30 40	0.205 0.830 1.168	$0.278 \\ 0.154 \\ 0.159$	0.032 0.201 0.309	0.081 0.070 0.037	42.6 27.3 53.3
3,720	:	2,600 2,580	$\begin{array}{c} 11 \\ 20 \end{array}$	0.726 1.588	$0.066 \\ 0.053$	0.181 0.448	0.018 0.013	45.5 50.6
3,780		2,855	5	0.760	0.033	0.218	0.010	41.0
4,040 3,940 3,960		2,835 2,800 2,940	14 30 60	0.295 0.582 <sup>(4)</sup> 1.308	$0.103 \\ 0.139 \\ 0.114$	0.036 0.119 0.355	0.085 0.115 0.101	15.0 14.0
4,730 4,680		3,590 3,610	11 30	0.301 0.813	0.139 0.183	0.032 0.168	0.103 0.147	16.7 15.4
5,320 5,120		4,330 3,980	10 20	0.496 0.970	0.127 0.116	0.108 0.260	0.098 0.094	16.1 15.3
			21	2.460		0.201		10.0
			10	2.165		0.180		
4,530		3,350	21	0.750	0.267	0.040	0.099	33.5
5,050 5,280		4,030 4,180	10 20	0.701 1.298	0.238 0.280	0.039 0.135	0.098 0.109	30.2 31.8

<sup>(4)</sup> Mass loss estimated from linear recession data.

 <sup>(4)</sup> Mass loss estimated from fineal recession data.
 (5) Facility thin wall transient flat face calorimeter, 1 in. total diameter, instrumented with multiple thermocouples. Corrected to 1.25 in. flat face with relation \$\dag{4}\_{1.25} = (1.0/1.25)^{0.5}\$ qrace FF meas

#### Table B-3 (Continued)

# TUNNEL CALIBRATION AND TEST DATA FOR VARIOUS DIAMETER MODELS REPORTED BY APPLIED MATERIALS AND PHYSICS DIVISION,

#### LANGLEY RESEARCH CENTER-NASA

Ref: Data sheets on AMPD runs 1108 to 1138

	MODEL NO.	TOTAL ENTHALPY  h t (Btu/lb) (1)	MODEL STAGNATION PRESSURE P <sub>t2</sub> (atm)	MODEL DIAMETER AND SHAPE (in.)	CALORIMETER DIAMETER AND SHAPE (in.)
Teflon	7164	8200	0.0192	1.0 Hemi.	1.25 FF
	7165	9100	0.0192	1.0 Hemi.	1.25 FF
	T160	9700	0.0192	1.25 FF	1.25 FF
	T161	9100	0.0192	1.25 FF	1.25 FF
•	T156	9100	0.0192	2.5 FF	2.5 FF
	T157	9800	0.0192	2.5 FF	2.5 FF
	T153	8700	0.0192	4.0 FF	4.0 FF
	T154	9800	0.0192	4.0 FF	4.0 FF
Hughes Phenolic-Nylon H-5	PLH164	8500	0.0192	1.0 Hemi.	1.25 FF
	PLH165	10300	0.0192	1.0 Hemi	1.25 FF
	PLH160	8700	0.0192	1.25 FF	1.25 FF
	PLH161	9100	0.0192	1.25 FF	1.25 FF
	PLH156 PLH157	9000 8800	0.0192	2.50 FF 2.50 FF	2.50 FF 2.50 FF
	PLH153	9100	0.0192	4.0 FF	4.0 FF
	PLH154	9900	0.0192	4.0 FF	4.0 FF
	I		1	J	1

HEAT TRA	NSFER RATE <sup>q</sup> (		esec) <sup>(2)</sup>
No. 9 Calorimeter (2)	No. 1	O Calorime	ter
1.25 FF	1.25 FF	2.5 FF	4.0 FF
115.7 132 125 130 112 164 136 136 130	107 156 137 128 134	94.6 95.0	76.7 76.0

HEAT TR/ RAT <sup>Q</sup> CV (Btu/ft (3)	E V	RUN TIME (sec)	CORE DIAMETER (in.)	CORE WEIGHT LOSS (g)	CORE CHAR WEIGHT (g)	CORE RECESSION (in.)	CHAR THICKNESS (in.)	CHAR DENSITY (lb/ft <sup>3</sup> )
116 130 137 130 91.6 93.2 70 78	236 264	49 49 97 97 140 140 ~57 63	0.5 0.5 0.625 0.625 1.25 1.25 2.5	1.461 1.510 2.764 2.846 10.667 11.378 19.422 33.739		0.207 0.217 0.242 0.247 0.236 0.247 0.100 0.184		(5) (6)
120 147 124 129 90.5 88.6 129 78.5	244 298	49 49 97 97 140 140 168 180	0.5 0.5 0.625 0.625 1.25 1.25 2.5 2.5	0.491 0.535 0.799 0.908 3.377 3.175	0.106 0.119 0.227 0.227 1.037 1.032 5.209	0.154 0.162 0.118 0.160 0.110 0.104	0.171 0.168 0.207 0.197 0.223 0.225	12.1 <sup>(7)</sup> 13.7 13.6 14.3 14.4 14.2 <sup>(8)</sup> (9) 14.9

- (1) Enthalpy calculated from measured heating rates and pressures.
- (2) All flat-faced models had peripheral shoulder radius equal to one-tenth model radius.
- (3) All heating rates measured with SRI 1.25-in. calorimeters. A shroud was added to the SRI calorimeter to equal the 2.5-in. and 4.0-in.-diameter models.
- (4) Heating rate calculated from (3) with the relation:  $\dot{q}_{CW} = q_{meas} (3.3 \times 1.25/1.0)^{0.5}$
- (5) Tunnel unstarted after  $\sim$ 55 sec, pressure and heating rate increased.
- (6) Tunnel unstarted after 60 sec.
- (7) Calorimeter may be in error.
- (8) Electrode burn out-model was wet.
- (9) Tunnel unstarted when model entered stream-model destroyed.

Table B-4

## TUNNEL CALIBRATION AND TEST DATA REPORTED BY ENTRY STRUCTURES BRANCH, 5Mw FACILITY, LANGLEY RESEARCH CENTER—NASA

Ref: Letter Report by P. F. Korycinski, Langley Research Center, December 23, 1966

	MODEL NO.	TOTAL ENTHALPY h <sub>t</sub> (Btu/lb)		HEAT TRANSFER RATE q <sub>CW</sub> (Btu/ft <sup>2</sup> sec) SRI CALORIMETER (3)	MODEL STAGNATION PRESSURE P <sub>t</sub> (atm) (4)	CHAMBER PRESSURE Pt1 (atm)
Langley Phenolic-Nylon	PLL54	3160	3250	203	0.281	1.389
Scout R/4B	PLL52	2970	3250	203	0.281	1.389
Avcoat 5026-39	A7	3050	3250	203	0.278	1.376
	A48	3100	3250	203	0.279	1.381
Modified Purple Blend	SP45	3100	3250	203	0.280	1.385
Silicone E4A1	SP46	3090	3250	203	0.281	1.392
G.E. Silicone	SG34	3120	3250	203	0.281	1.389
ESM 1004AP	SG33	3060	3250	203	0.281	1.389
Hughes Phenolic-Nylon	PLH41	3070	3250	203	0.281	1.389
H-5	PLH40	3140	3250	203	0.281	1.389

<sup>(1)</sup> Enthalpy measured by energy balance method.

<sup>(2)</sup> Enthalpy calculated from pressure and heat transfer to 1.5-in. diameter hemispherical facility calorimeter, and from theory of Fay and Riddell. This method of calculating enthalpy is considered standard and is preferred by the Entry Structures Branch.

<sup>(3)</sup> Heat transfer to SRI calorimeter determined during separate run.

<sup>(4)</sup> Model stagnation pressure calculated from chamber pressure  $P_{t_2}/P_{t_1} = 0.202$ .

GAS FLOW RATE (lb/sec)	POWER TO ARC (Mw)	RUN TIME (sec)	CORE WEIGHT LOSS (g)	CORE CHAR WEIGHT (g)	CORE RECESSION (in.)	CHAR THICKNESS (in.)	CHAR DENSITY (lb/ft <sup>3</sup> )
0.2	2.15	15.3	0.202	0.076	0.023	0.065	14.5
	2.13	30.4	0.578	0.131	0.131	0.108	15.0
0.2	2.14	15.3	0.357	0.104	0.102	0.075	17.2
	2.15	30.4	0.710	0.103	0.243	0.067	19.0
0.2	2,15	15.3	0.162	0.190	+0.020	0.106	22.2
	2,13	30.4	0.282	0.276	0.006	0.147	23.2
0.2	2.15	15.3	0.149	0.275	0.009	0.101	33.7
	2.12	30.4	0.310	0.300	0.018	0.151	24.6
0.2	2.15	15.3	0.318	0.116	0.037	0.098	14.6
	2.14	30.4	0.604	0.109	0.125	0.108	13.1

Table B-5

## CALIBRATION AND TEST DATA REPORTED BY MANNED SPACECRAFT CENTER, SUBSONIC FACILITY—NASA

Ref: MSC Letter Report ES5/10-17/81L, October 19, 1966

	MODEL	TOTAL	HEAT TRAN	SFER	DISTANCE	MODEL	GAS FLOW
	NO.	ENTHALPY h <sub>t</sub>	RATE		NOZZLE EXIT TO	STAGNATION	RATE (1b/sec)
		"t (Btu/lb)	q <sub>CW</sub>		MODEL	PRESSURE P <sub>t</sub>	(ID/Sec)
			(Btu/ft <sup>2</sup>		FACE (in.)	(atm)	
			Calorime Facility	SRI			
		(1)	(2)	DILL		9	
Langley Phenolic-Nylon Scout R/4B	PLL38 PLL39	8223 8280	404 401		2.0 2.0	$\begin{array}{c} 1.0 \\ 1.0 \end{array}$	0.04 0.04
	PLL40 PLL41	9975 8930	590 600		1.5 1.5	1.0 $1.0$	0.04 0.04
	PLL36 PLL37	4500 4622	100 100		$\frac{2.0}{2.0}$	$\begin{array}{c} 1.0 \\ 1.0 \end{array}$	0.04
	PLL95	3888	101		2.0	1.0	0.04 0.04
	PLL42	5475	205		2.0	1.0	0.04
Avcoat 5026-39	A38 A39	8440 8198	406 399		2.0 2.0	$\begin{array}{c} 1.0\\ 1.0 \end{array}$	0.04 0.04
	A40 A41	9078 9483	590 590		1.5 1.5	1.0 $1.0$	0.04 0.04
	A36 A37 A91	4822 4375 4275	106 103 104	· . :	2.0 2.0 2.0	$egin{array}{c} 1.0 \ 1.0 \ 1.0 \end{array}$	0.04 0.04 0.04
	A42 A92	4853 4950	210 207		2.0 2.0	$\begin{array}{c} 1.0 \\ 1.0 \end{array}$	0.04 0.04
Modified Purple Blend Silicone E4Al	SP42 SP39	8440 8218	405 396		2.0 2.0	1.0 1.0	0.04 0.04
	SP40 SP41	9595 9792	600 590		1.5 1.5	$\begin{array}{c} 1.0 \\ 1.0 \end{array}$	0.04 0.04
:	SP36 SP37 SP95	4400 4445 4113	104 104 105		2.0 2.0 2.0	$\begin{array}{c} 1.0 \\ 1.0 \\ 1.0 \end{array}$	0.04 0.04 0.04
	SP38	4850	205		2.0	1.0	0.04
G. E. Silicone ESM 1004AP	SG28	8050	400		2.0	1.0	0.04
25.0 1004/1	SG29 SG30	9300 9050	613 599		1.5 1.5	$\begin{array}{c} 1.0 \\ 1.0 \end{array}$	0.04 0.04
	SG26 SG27 SG50	4674 4625 4609	105 105 106		2.0 2.0 2.0	$egin{array}{c} 1.0 \ 1.0 \ 1.0 \end{array}$	0.04 0.04 0.04
Hughes Phenolic-Nylon H-5	PLH32 PLH33	8130 8040	410 406		2.0 2.0	1.0 1.0	0.04 0.04
:	PLH34 PLH35	9250 9300	590 608		1.5 1.5	$egin{smallmatrix} 1.0 \ 1.0 \end{smallmatrix}$	0.04 0.04
:	PLH31	4493	101	:	2.0	1.0	0.04
	PLH36	4980	205		2.0	1.0	0.04
Calibration Runs	C1 C2 C3 C4 C5 C6 C7 C8	8640 6838 5205 3565 4050 5300 8700 4100	396 314 251 93 94 276 400 108	388 297 226 101 103 251 366 100	2.0 2.0 2.0 2.0 2.0 2.0 2.0 2.0 2.0	1.0 1.0 1.0 1.0 1.0 1.0 1.0	0.04 0.04 0.04 0.04 0.04 0.04 0.04
	C9 C10	5350 8650	276 398	261 360	2.0 2.0	1.0 1.0	0.04 0.04

<sup>(1)</sup> Enthalpy measured by heat balance method.

<sup>(2)</sup> Facility Hy-Cal asymptotic calorimeter.

<sup>(3)</sup> Measured with an optical pyrometer.

POWER TO ARC (Btu/sec)	MAXIMUM SURFAC TEMPERA <sup>T</sup> FS	Œ	RUN TIME (sec)	CORE WEIGHT LOSS (g)	CORE CHAR WEIGHT (g)	CORE RECESSION (in.)	CHAR THICKNESS (in.)	CHAR DENSITY (1b/ft <sup>3</sup> )
	Facility (3)	SRI						
687.5 688.0	4380	4350 4450	8.0 20.4	0.226 0.479	0.155 0.260	0.017 0.065	0.109 0.175	17.6 18.4
744.1 767.5	4155 4420	4280 4330	10.7 20.5	0.325 0.557	0.196 0.260	0.043 0.081	0.132 0.181	18.0 17.8
330.0 332.4 327.4	3000 2935 3050	2750 2980	15.5 30.5 60.8	0.191 0.374 0.762	$0.165 \\ 0.238 \\ 0.373$	0.008 0.034 0.070	0.114 0.160 0.267	17.9 18.4 17.3
373.1	3105	3250	31.0	0.453	0.262	0.063	0.185	11.4
671.3 647.4		4210 4310	8.6 20.4	0.269 0.572	0.142 0.186	0.051 0.146	0.097 0.114	18.2 20.2
752.4 768.2	4175 4005	4100 4040	10.7 20.3	0.394 0.656	0.133 0.147	0.108 0.202	0.096 0.100	17.2 18.2
314.1 315.5 320.	3085 3105 3105	3300 3280 3400	16.7 30.4 60.4	0.201 0.382 0.913	0.237 0.333 0.975	0.023 0.051 0.137	0.145 0.212 0.276	20.3 19.5 22.8
385.5 385.0	3295 3430	3160 3240	30.2 60.5	$0.519 \\ 1.169$	$0.255 \\ 0.243$	0.121 0.378	0.164 0.150	19.3 20.1
690.6 685.8		4200 4150	7.8 20.5	0.267 0.572	0.154 0.286	0.004 0.101	0.099 0.110	19.3 32.2
764.6 752.4	3790 3850	3940 3820	11.5 20.1	0.489 0.783	$0.152 \\ 0.314$	0.111 0.218	0.071 0.086	26.5 45.3
316.0 315.9 315.1	3105 3105 3160	3320 3300 3400	15.5 30.6 60.1	0.190 0.424 1.078	$\begin{bmatrix} 0.271 \\ 0.394 \\ 0.528 \end{bmatrix}$	+0.037 +0.035 0.038	0.153 0.228 0.320	22.0 21.4 20.5
384.0	3260	3220	30.4	0.575	0.273	0.005	0.205	16.5
658.6	3320	3480	20.5	1,213	0.183	0.355	0.043	52.8
765.0 740.0	3140 3390	3210 3070	10.5	0.931 1.752	0.127	0.273 0.521	0.026 0.036	60.6 54.8
319.5 320.0 318.4	3000 3105 3105	3240 3350 3300	20.6 50.9 60.8	0.182 0.732 1.210	0.419 0.810 0.626	+0.005 0.085 0.225	0.168 0.234 0.204	30.9 42.9 38.1
662.2 665.2		4300 4420	9.0 20.3	0.249 0.489	0.159 0.265	0.023 0.072	0.113 0.171	17.4 19.2
765.0 765.0	4030 4135	4330	10.6 20.4	0.343 0.574	0.199 0.282	0.045 0.094	0.137 0.189	18.0 18.5
326.3	2980	2800	20.3	0.267	0.169	0.016	0.117	17.9
384.7	3295	3200	30.8	0.516	0.280	0.069	0.178	19.5

 $\begin{tabular}{llll} Table B-6 \\ \hline TUNNEL CALIBRATION AND TEST DATA REPORTED BY MANNED SPACECRAFT CENTER, \\ SUPERSONIC FACILITY—NASA \\ \end{tabular}$ 

Ref: MSC		Report ES5/2-27/35L	, March 1,	1967		
	MODEL NO.	AVERAGE ENTHALPY (Btu/lb)	q <sub>CW</sub> (Btu/ft <sup>2</sup>	HEAT TRANSFER RATE  qCW  (Btu/ft <sup>2</sup> sec) Calorimeter		
) 		. (1)	Facility (2)	SRI	(3)(4)	
Langley Phenolic-Nylon	PLL65	11,500	436		0.625	
Scout R/4B	PLL66	24,500	800		0.700	
	PLL67	34,000	988		0.976	
Avcoat 5026-39	A62	11,500	431		0.625	
	A63	25,466	780		0.710	
	A64	34,000	954		0.976	
Modified Purple Blend	SP59	11,500	426		0.625	
Silicone E4Al	SP60	24,500	794		0.704	
	SP61	34,000	958		0.976	
G. E. Silicone ESM1004AP	SG59 SG60	11,500 11,500	436 433		0.625 0.625	
	SG61	23,500	780		0.682	
	SG62	34,150	954		0.976	
Hughes Phenolic-Nylon	PLH54	11;500	431		0.625	
H- <sup>2</sup> 5	PLH55	24,600	797			
	PLH56	34,150	1000		0.976	
Teflon	T129 T130	11,500 11,500	438 433		0.625 0.625	
,	T1 31	24,500	780	1	0.678	
	T133	34,500	928		0.976	
Phenolic-Nylon (75 lb/ft <sup>3</sup> )	Pl 1Bl Pl 1B2	11,500 11,500	431 433		0.625 0.625	
	Pl 1B3	26,500	818		0.720	
	P11B5	34,300	954	4.	0.976	
	C1	11,000	424	394	1	
	C2	22,500	824	839		

- (1) Enthalpy measured by energy balance method.
- (2) Facility Hy-cal Engineering asymptotic flat-face calorimeter, 0.060-in. sensing diameter, 1.25-in. total diameter.
- (3) Models run at h = 11,000 had model stagnation pressures ranging from 0.011 to 0 0125 atm; at h = 25,000,  $P_{\rm t}$  = 0.0163 to 0.0178 atm; at h = 34,000,  $P_{\rm t}$  = 0.0192 atm. Model stagnation pressures were measured with a Grey-Rad enthalpy probe.
- (4) Nozzle exit pressures ranged from 0.00065 to 0.0012 atm for the above tests and chamber test section pressures were controlled within the same range of values.

GAS FLOW RATE (1b/sec)	MODEL FRONT SURFACE TEMPERATURE $T_{FS}  \epsilon = 1$ $(^{o}_{F})$ Facility $(5)$	RUN TIME (sec)	CORE WEIGHT LOSS (g)	CORE CHAR WEIGHT (g)	CORE RECESSION (in.)	CHAR THICKNESS (in.)	CHAR DENSITY (1b/ft <sup>3</sup> )
0.013	3890	31.0	0.430	0.256	0.039	0.184	17.3
0.013	3950	30.0	0.555	0.346	0.051	0.243	17.7
0.013	3810	30.0	0.540	0.351	0.047	0.246	17.7
0.013	3880	29.5	0.474	0.229	0.073	0.169	16.8
0.013		30.0	0.563	0.254	0.110	0.200	15.8
0.013	3880	20.0	0.430	0.208	0.068	0.171	15.1
0.013	3525	30.5	0.462	0.141	0.029	0.131	13.4
0.013	3900	30.0	0.719	0.061	0.187	0.063	159
0.013	3890	20.4	0.546	0.085	0.102	0.082	12.9
0.013 0.013	2975 3160	31.0 30.0	0.819 0.830	0.093 0.098	0.252 0.251	0.041 0.044	28.1 27.6
0.013	3380	30.0	1.259	0.076	0.408	0.036	26.2
0.013	3575	20.0	1.042	0.097	0.329	0.034	35.4
0.013	3870	30.0	0.448	0.240	0.044	0.171	17.4
0.013	3805	30.0	0.581	0.344	0.055	0.232	18.4
0.013	3840	20.0	0.409	0.207	0.027	0.151	17.0
0.013 0.013		30.0 29.5	1.724 1.409		0.151 0.123		
0.013		30.0	1.903		0.168		
0.013		20.0	1.528		0.136		
0.013 0.013	2975 2920	31.0 29.5	0.656 0.626	0.211 0.205	0.021 0.018	0.102 0.100	25.7 25.4
0.013	2950	30.0	0.818	0.396	0.030	0.148	33.2
0.013	3840	20.2	0.522	0.163	0.006	0.081	24.9
0.013							
0.013							

<sup>(5)</sup> Facility optical pyrometer, 0.65 microns.

Table B-7 TUNNEL CALIBRATION AND TEST DATA REPORTED BY AEROTHERM CORPORATION

Ref: Aerotherm Report No. 66-6

	MODEL NO.	h.	AL ENTHAI t Btu/lb)	.PY	HEAT TRA RAT <sup>q</sup> C (Btu/ft Càlori	E W <sup>2</sup> sec)	MODEL STAGNATION PRESSURE Pt <sub>2</sub> (atm)	PLENUM PRESSURE Pt1 (atm)
		(1)	(2)	(3)	Facility (4)	SRI		
Langley Phenolic-Nylon Scout R/4B	PLL43 PLL46 PLL97	4,748 4,085 4,748		5,583 4,146 5,583	92.1 82.3 92.1	80.1 57.4 80.1	$0.0204 \\ 0.0190 \\ 0.0204$	0.0693 0.0704 0.0693
	PLL45 PLL48	7,612 7,612		10,783 11,304	$270.0 \\ 270.0$	188.0 188.0	0.0301 0.0301	0.1034 0.1034
	PLL15 PLL18	21,206 21,598		46,535 44,625		973.0 921.0	0.0433 0.0422	0.3547 0.3547
Avcoat 5026-39	A74 A44 A98	4,783 4,085 4,783		5,644 4,608 5,644	86.2 82.3 86.2	76.5 57.4 76.5	0.0182 0.0190 0.0182	0.0682 0.0704 0.0682
	A58 A59	7,227 7,227		11,304 11,530	$267.0 \\ 267.0$	196.0 196.0	0.0298 0.0298	0.1034 0.1034
. :	A15 A17	21,973 21,925		45,224 45,007		941.0 941.0	0.0429 0.0433	0.3536 0.3558
Modified Purple Blend Silicone E4Al	SP56 SP53 SP97	4,566 4,242 4,566		5 469 10,783 5,469	83.2 77.4 83.2	77.7 63.8 77.7	0.0200 0.0190 0.0200	0.0704 0.0716 0.0704
	SP55 SP57	7,158 7,158		11,530 10,111	$263.0 \\ 263.0$	200.0	0.0298 0.0298	0.1046 0.1046
	SP15 SP18	21,871 21,650		45,751 46,197		952.0 958.0	0.0429 0.0426	0.3570 0.3536
G. E. Silicone ESM 1004AP	SG41 SG42	4,158 4,158		4,839 4,839	80.3 80.3	69.1 69.1	0.0202 0.0202	0.0682 0.0682
	SG1 4 SG4 3	7,667 7,667	=	10,111 9,919	261.0 261.0	174.0 174.0	0.0294 0.0294	0.1046 0.1046
	SG12 SG15	21,412 21,729		43,684 46,952		909.0 969.0	0.0429 0.0422	0.3536 0.3536
Hughes Phenolic-Nylon H-5	PLH44 PLH45	4,328 4,328		4,125 4,125	87.2 87.2	58.2 58.2	0.0197 0.0197	0.0716 0.0716
	PLH46 PLH48	8,133 8,133		9,919	264.0 264.0	$172.0 \\ 172.0$	0.0298 0.0298	0.1046 0.1046
·	PLH16 PLH17	21,442 21,168		46,678 46,808		960.0 974.0	0.0419 0.0429	0.3524 0.3536
Calibration Runs	C1 C2	4,947 4,279	4,500 4,500	5,711 6,085	83.2 85.4	80.1 84.5	0.0195 0.0191	0.0659 0.0693
	C3 C4	8,016 7,133	9,600 9,600	10,326 10,791	230.0 236.0	178.0 185.0	0.0294 0.0291	0.1023 0.1023
	C5 C6	21,255 21,394	32,800 29,500		1,033.0 1,094.0	837.0 876.0	0.0409 0.0404	0.3490 0.3331
	C7 C8	4,518 4,500			86.2	72.3 70.1	0.0196 0.0196	
	C9 C1 (	4,624 4,796			(7) 62.8 59.1	(7) 51.0 52.8	0.0189 0.0196	

<sup>(1)</sup> Enthalpy measured by energy balance method.

<sup>(2)</sup> Enthalpy calculated by Aerotherm Corporation using sonic flow technique

<sup>(3)</sup> Enthalpy calculated by Aerotherm Corporation using heat flux method:  $h = 24 \, q_{CW} \, (P_{t_2}/R_{eff})^{0.5}$ .

<sup>(4)</sup> Aerotherm calorimeter "Gardon" type, steady state, Hy-Cal Engineering; sensing area 0.20-in.
Constantan, shroud diameter, 1.5-in. flat face. Values corrected to 1.25-in. diameter flat face: q<sub>1.25 in.</sub> = 1.095 q<sub>1.5 in.</sub>

NOZZLE EXIT PRESSURE Pe (atm)	GAS FLOW RATE (lb/sec)	MAXIMUM SURFA TEMPERA TFS	ACE	RUN TIME (sec)	CORE WEIGHT LOSS (g)	CORE CHAR WEIGHT (g)	CORE RECESSION (in.)	CHAR THICKNESS (in.)	CHAR DENSITY (1b/ft <sup>3</sup> )
		Facility (5)	SRI (6)		:				
0.0016	0.0042	3,050	2,450	25.4	0.254	0.101	0.033	0.090	13.9
0.0014	0.0042	3,010	2,380	60.5	0.542	0.147	0.096	0.126	14.5
0.0016	0.0042	3,210	2,520	100.8	0.778	0.178	0.189	0.156	14.2
0.0021	0.0048	3,840	3,100	12.4	0.240	0.095	0.021	0.083	14.2
0.0021	0.0048		3,100	45.3	0.607	0.195	0.105	0.154	15.7
0.0027	0.0084	4,460	4,160	5.2	$0.164 \\ 0.290$	0.111	0.008	0.082	16.8
0.0027	0.0084	5,090	4,620	13.3		0.212	0.021	0.152	17.3
$\begin{array}{c} 0.0014 \\ 0.0014 \\ 0.0014 \end{array}$	0.0042	2,930	2,380	25.3	0.235	0.144	0.027	0.111	16.1
	0.0042	2,930	2,510	59.7	0.467	0.203	0.103	0.146	17.3
	0.0042	3,190	2,490	99.8	0.852	0.239	0.251	0.171	17.3
0.0021	0.0048	3,720	3,090	11.9	0.210	0.105	0.038	0.080	16.3
0.0021	0.0048	3,800	2,900	45.1	0.695	0.164	0.210	0.116	17.5
0.0027	0.0084	4,820	4,340	5.0	0.147	0.080	0.017	0.076	13.1
0.0027	0.0084	5,240	4,540	13.4	0.325	0.143	0.058	0.127	14.0
0.0014	0.0042	2,600	2,350	24.6	0.155	0.123	+0.044	0.120	12.3
0.0014	0.0042		2,280	60.4	0.280	0.225	+0.051	0.194	14.3
0.0014	0.0042		2,280	100.0	0.404	0.276	+0.025	0.217	15.8
0.0021	0.0048	3,200	2,840	12.4	0.199	0.128	0.002	0.092	17.2
0.0021	0.0048	3,470	2,840	45.3	0.578	0.177	0.120	0.111	19.7
0.0027	0.0084	4,140	3,720	5.3	$0.197 \\ 0.451$	0.039	0.015	0.044	11.0
0.0027	0.0084	4,310	3,720	13.0		0.078	0.086	0.065	14.9
0.0014	0.0042	2,570	2,320	25.0	0.112	0.278	+0.010	0.116	29.7
0.0014	0.0042	2,610	2,300	99.7	0.315	0.616	+0.036	0.286	26.7
0.0021	0.0048	3,220	2,740	12.5 $45.4$	0.262	0.130	0.051	0.058	27.8
0.0021	0.0048	3,200	2,640		0.921	0.170	0.259	0.066	31.9
0.0027	0.0084	3,590	3,170	5.2	0.339	0.057	0.066	0.043	16.4
0.0027	0.0084		3,140	13.1	0.831	0.065	0.226	0.026	31.0
0.0014	0.0042	2,950	2,460	24.8	0.260	0.093	0.026	0.084	13.7
0.0014	0.0042	3,250	2,470	99.7	0.893	0.187	0.185	0.157	14.8
0.0021	0.0048	3,640	3,090	$\substack{12.4\\45.0}$	0.238	0.087	0.018	0.078	13.8
0.0021	0.0048	3,800	2,710		0.623	0.207	0.103	0.163	15.8
0.0027	0.0084	3,840	4,040	$\frac{5.2}{13.5}$	0.159	0.114	0.009	0.083	17.0
0.0027	0.0084	4,920	4,570		0.330	0.231	0.025	0.155	18.5
0.0014 0.0014	0.0042 0.0042			-					
0.0021 0.0021	0.0048 0.0048							:	
0.0027 0.0027	0.0084 0.0084								

<sup>(5)</sup> Infrared Industries "Thermodot" Mo TD9CH; viewed model through front quartz port, approximately  $45^\circ$  to model surface plane,  $0.80\pm0.015$  microns.

<sup>(6)</sup> SRI radiometer located inside test chamber; viewed model with front surface mirror. Data has been corrected: 1.12  $T_{\text{meas}} = T_{\text{COR}}$ .

<sup>(7)</sup> The Aerotherm and SRI calorimeters were sprayed with a thin coat of Teflon to determine the effect of reduced surface catalyticity.

Table B-8

TUNNEL CALIBRATION AND TEST DATA REPORTED BY AVCO CORPORATION

Ref: AVCO Report R720-HEH-66-105, August 2, 1966

<u> </u>		<u> </u>					
	MODEL NO.	TOTAL ENTHALPY h <sub>t</sub> (Btu/lb)	RAT <sup>Q</sup> C (Btu/ft	HEAT TRANSFER RATE		PLENUM PRESSURE P <sub>t</sub> 1 (atm)	NOZZLE THROAT DIAMETER (in.)
		(1)	Facility (2)	SRI (3)	(4)		
Teflon	T110	2,430		1,640	5.62	10.91	0.765
	T111	6,820		3,340	4, 92	10.40	0.765
	T113 T114	7,260 7,260		1,850 1,850	1.93 1.93	2.06 2.06	$\begin{array}{c} 1.25 \\ 1.25 \end{array}$
Phenolic-Nylon	Pl 2A2	2,430		1,640	5.62	10.91	0.765
$(75 \text{ lb/ft}^3)$	P12A3	6,690		3,340	4.92	10.37	0.765
	P12A4 P12A5	7,260 7,260		1,850 1,850	1.93 1.93	2.05 2.05	1.25 1.25
Langley Phenolic- Nylon Scout R/4B	PLL47	7,000		1,850	1.93	2.03	1.25
Avcoat 5026-39	A73	7,000		1,850	1.93	2.03	1.25
Tunnel Calibration Runs	C1 C2 C3 C4	2,470 2,430 2,500 2,400	1,640	1,640	5.62 5.62	10.91 10.94 10.91 10.87	0.765 0.765 0.765 0.765
	C5 C6 C7 C8	6,700 7,000 7,000 7,000	3,400	3,340	4.91 4.93	10.34 10.40 10.40 10.40	0.765 0.765 0.765 0.765
	C9 C10 C11	7,260 7,260 7,260	1,960	1,850	1.93	2.06 2.06 2.06	

<sup>(1)</sup> Enthalpy determined by sonic flow method.

<sup>(2)</sup> AVCO null point transient calorimeter, 0.375-in. (sensing diameter) copper slug by 1.5-in. long, mounted in a 1.25-in. flat face shroud.

<sup>(3)</sup> SRI calorimeter values taken from tunnel calibration runs.

<sup>(4)</sup> AVCO uncooled copper pitot probe, 0.375-in diameter. Test data taken from tunnel

NOZZLE EXIT DIAMETER (in.)	AIR MASS FLOW (lb/sec)	MAXIMUM FRONT SURFACE TEMPERATURE $T_{FS} \qquad \epsilon = 1$ $(^{\circ}F)$ (5)	RUN TIME (sec)	CORE WEIGHT LOSS (g)	CORE CHAR WEIGHT (g)	CORE RECESSION (in.)	CHAR THICKNESS (in.)	CHAR DENSITY (1b/ft <sup>3</sup> )
1.178	0.440		4.02	2.376		0.208		
1.178	0.273		4.03	2.679		0.228		
1.25 1.25	0.141 0.141		2.97 10.05	1.176 3.999		0.104 0.358		
1.178	0.440	4,220	4.06	0.657	0.033	0.085	0.024	17.0
1.178	0.274	4,540 - 4,840	4.00	1.217	0.016	0.183	0.010 <sup>(6)</sup>	19.8
1.25	$0.140 \\ 0.141$	5,280 5,440	2.97 10.01	0.396 1.053	$0.101 \\ 0.174$	0.016 0.114	0.057 0.076	22.0 28.4
1.25	0.141	5,550	6.02	0.440	0.069	0.144	0.055	
1.25	0.141	4,900	6.01	1.040	0.015	0.416	0.014 <sup>(7)</sup>	
1.178 1.178 1.178 1.178	0.437 0.442 0.442 0.441		0.88 0.93 0.46 0.52				- '	
1.178 1.178 1.178 1.178	0.273 0.270 0.270 0.270		0.71 0.84 0.45 0.41	:		-		
	0.141 0.141 0.141	:	1.13 0.36 0.37				-	

<sup>(5)</sup> Instrument Development Lab recording pyrometer (0.653 microns); viewed model directly.

 $<sup>\</sup>begin{tabular}{ll} \begin{tabular}{ll} \be$ 

<sup>(7)</sup> Model A73 lost considerable side shroud material because of incomplete honeycomb cells;  $\overset{\bullet}{q}$  may have increased during run.

Table B-9

TUNNEL CALIBRATION AND TEST DATA REPORTED BY GIANNINI SCIENTIFIC CORPORATION

Ref: G.S.C. Test Report No. FR076-332, July 1966

	MODEL NO.	TOTAL ENTHALPY	HEAT TRAN	SFER	MODEL STAGNATION PRESSURE	PLENUM PRESSURE	NOZZLE STATIC PRESSURE
		(Btu/lb)	q <sub>CW</sub> (Btu/ft <sup>2</sup> sec)		P t2 (atm)	P t1 (atm)	P e (atm)
			Calorim	eter	`		
		(1)	Facility	SRI	(4)		
		(1)	(2)				
Langley Phenolic-Nylon Scout R/4B	PLL1 PLL2 PLL90	10,200 10,200 10,200	145 146 145		0.0197 0.0197 0.0199	0.084 0.084 0.086	0.00155 0.00158 0.00156
·	PLL3 PLL4	10,090 10,080	65 64		0.0041 0.0041	0.0194 0.0195	0 00035 0 00035
	PLL5 PLL6 PLL7	15,390 15,400 15,400	457 456 457		0.094 0.095 0.095	0.625 0.630 0.624	0.0107 0.0106 0.0106
Avcoat 5026-39	A1 A2 A94	10,200 10,200 10,200	145 146 144		0.0199 0.0197 0.0199	0.086 0.085 0.086	0.00156 0.00157 0.00157
	A3 A4	10,100 10,090	66 65		0.0041 0.0041	0.0197 0.0195	0.00036 0.00035
	A5 A6	15,380 15,400	455 456		0.093	0.626 0.625	0.0104 0.0105
Modified Purple Blend Silicone E4Al	SP1 SP2 SP90	10,180 10,200 10,200	145 146 145		0.0197 0.0199 0.0199	0.085 0.086 0.086	0.00156 0.00156 0.00155
	SP3 SP4	10,100 10,100	66 64	:	0.0041 0.0041	0.0194 0.0196	0.00035 0.00036
	SP5 SP6	15,360 15,400	455 457		0.093	0.620 0.626	0.0105
G. E. Silicone ESM 1004AP	SG1 SG45	10,200 10,190	145 144		0.0197 0.0199	0.084 0.085	0.00155 0.00155
	SG2 SG3	10,080 10,100	65 66		0.0041 0.0041	0.0195 0.0197	0.00036 0.00036
	SG4 SG5	15,380 15,400	456 457		0.093 0.095	0.624 0.626	0.0105 0.0106
Hughes Phenolic-Nylon H-5	PLH1 PLH90	10,200 10,200	146 144		0.0199 0.0199	0.085	0.00156 0.00156
	PLH2 PLH3	10,100 10,100	65 65		0.0041 0.0041	0.0196 0.0195	0.00035 0.00035
	PLH4 PLH5	15,400 15,380	457 456		0.095 0.094	0.625 0.622	0.0106 0.0105
Calibration Runs	C1	10,200	(3)	-	0.0197	0.085	0.00157
Satisfacton Runs	C2	10,190		135	0.0197	0.084	0.00157
1	C3 C4	10,100	67	57	0.0041 0.0041	0.0197 0.0193	0.00036 0.00035
	C5 C6	15,400 15,400	458	457	0.095 0.095		

<sup>(1)</sup> Enthalpy measured by energy balance method

<sup>(2)</sup> Giannini steady state calorimeter, 0.625-in. diameter hemispherical shape, copper surface, water temperature rise type This calorimeter was calibrated with calorimeter described under (3).

<sup>(3)</sup> Giannini transient calorimeter used to calibrate (2), 0.25-in. diameter by 0.25-in. long copper slug set in graphite shroud with shape same as models.

<sup>(4)</sup> Giannini pitot probe, water-cooled, 0.625-in. diameter.

GAS FLOW RATE (lb/sec)	MAXIMUM TE T <sub>FS</sub>	MPERATUI		RUN TIME (sec)	CORE WEIGHT LOSS	CORE CHAR WEIGHT	CORE RECESSION (in.)	CHAR THICKNESS (in.)	CHAR DENSITY (lb/ft <sup>3</sup> )
	Facility (5)	(6)	SRI (7)						
0.0043	3,000	2,820	2,900	9.4	0.142	0.073	0.010	0.063	14.4
0.0043	3,260	3,280	3,180	20.9	0.229	0.119	0.022	0.101	14.6
0.0043	3,680	3,800	3,620	34.7	0.384	0.214	0.073	0.159	16.7
0.00084	2,300	2,300	2,140	33.0	0.208	0.092	0.024	0.084	13.6
0.00084	2,480	2,610	2,350	79.1	0.436	0.172	0.075	0.131	16.3
0.0189 0.0189 0.0189	3,950 4,230 4,020	4,350	4,180 4,000	$5.1 \\ 10.8 \\ 5.2$	0.147 0.249 0.148	0.092 0.156 0.093	0.010 0.030 0.009	0.075 0.114 0.075	15.2 16.9 15.4
0.0043 0.0043 0.0043	3,200 3,370 3,300	3,300 3,400 3,500	3,100 3,300 3,290	$11.1 \\ 20.8 \\ 34.7$	0.148 0.251 0.361	$\begin{bmatrix} 0.112 \\ 0.162 \\ 0.212 \end{bmatrix}$	0.016 0.037 0.072	0.077 0.107 0.144	18.0 18.8 18.3
0.00084	2,350	2,500	2,280	32.5	0.206	0.139	0.030	0.094	18.3
0.00084	2,390	2,600	2,300	78.9	0.373	0.236	0.064	0.173	17.0
0.0189	3,550	4,240	4,100	5.7	0.176	0.096	0.027	0.078	15.3
0.0189	4,150	4,360	4,250	10.7	0.286	0.127	0.062	0.099	15.9
0.0043 0.0043 0.0043	2,780 3,170 3,060	2,820 3,340	2,650 3,050 3,000	10.4 20.8 35.0	$\begin{bmatrix} 0.132 \\ 0.181 \\ 0.369 \end{bmatrix}$	$\begin{bmatrix} 0.102 \\ 0.150 \\ 0.199 \end{bmatrix}$	+0.016 +0.030 0.004	0.075 0.125 0.127	16.9 14.9 19.1
0.00084	2,380	2,450	2,180	32.7	0.169	0.116	+0.036	0.109	13.2
0.00084	2,400	2,485	2,250	79.0	0.286	0.197	+0.048	0.177	13.8
0.0189	3,850	3,340	3,850	5.9	0.207	0.055	0.026	0.045	15.1
0.0189	3,850		3,850	10.7	0.338	0.074	0.070	0.056	16.4
0.0043	3,020	3,240	2,950	20.9	0.255	0.181	0.043	0.077	29.2
0.0043	3,100	3,300	3,050	34.8	0.478	0.248	0.065	0.107	28.8
0.00084	2,400	2,470 2,455	2,250	32.6	0.108	0.295	+0.011	0.115	31.8
0.00084	2,380		2,230	79.2	0.193	0.488	+0.015	0.200	30.2
0.0189 0.0189	3,250 2,900	3,320 3,300	3,250	5.8 11.5	0.268	0.105	0.075 0.182	0.030 0.030	43.5
0.0043 0.0043	3,300 3,350	3,360 3,460	3,250 3,250	20.8 34.8	0.229	0.108 0.163	0.014 0.025	0.094 0.135	14.2
0.00084 0.00084	2,490 2,550	2,400 2,590	2,320 2,330	32.8 78.8	0.222	0.088	0.023 0.073	0.085 0.130	12.8 15.5
0.0189	3,850	4,010	3,800	4.9	0.125	0.084	0.004	0.068	15.3
0.0189	4,200	4,300	4,150	10.8	0.260	0.156	0.025	0.113	17.1
0.0043 0.0043 0.00084 0.00084	2.5								

<sup>(5)</sup> Thermodot Mo. TD-6BT radiation thermometer (1.6-2.7 microns); viewed model through front port, approximately  $40^\circ$  to surface plane of model.

<sup>(6)</sup> L-N optical pyrometer (0.655 microns).

<sup>(7)</sup> SRI radiometer; viewed model through front port, approximately  $40^{\circ}$  to surface plane of model.

Table B-10(a)

TUNNEL CALIBRATION AND TEST DATA REPORTED BY MARTIN COMPANY
Ref: Martin Report ER 14356, August 26, 1966

		<del></del>	<del>نىسىدىدۇم</del>	<del></del>	<del></del>		<del></del>	diagram of the second
	MODEL NO,	TOTAL ENTHALPY		AT TRAN			MODEL STAGNATION PRESSURE	PLENUM PRESSURE Pt1
		(Btu/1b)	,	Calori	meter		t <sub>2</sub> (atm)	(atm)
-		(1)	(2)	Facility (3)	(4)	SRI (5)		
Langley Phenolic-Nylon Scout R/4B	PLL12 PLL13 PLL14 PLL91	5,143 4,996 6,170 5,140	46.5 42.6 44.0 43.2			59.8	0.0071 0.0069 0.0070 0.0070	0.0303 0.0303 0.0301
	PLL10 PLL11	18,117 18,117	456. 456.				0.0333 0.0333	0.0448 0.0448
	PLL8 PLL9	10,137 10,947	417. 418.				$0.140 \\ 0.144$	$0.263 \\ 0.264$
Avcoat 5026-39	A12 A13 A95	5,143 4,996 5,140	46.5 42.6 43.2			59.8	0.0071 0.0069 0.0070	0.0303 0.0303 0.0301
	A10 A11	18,117 18,445	456. 475.				0.0333 0.0340	0.0448 0.0433
	A8 A9 A1 4	10,137 10,947 10,387	417. 418. 417.	:		i	0.140 0.144 0.140	0.263 0.264 0.260
Modified Purple Blend Silicone E4Al	SP13 SP14 SP91	4,933 5,226 5,180	44.4 42.3 44.2			57.5	0.0070 0.0070 0.0070	0.0303 0.0264 0.0316
	SP10 SP11 SP12	17,950 18,445 18,642	475. 475. 455.	3			0.0340 0.0340 0.0341	0.0435 0.0433 0.0632
	SP8 SP9	10,647 10,158	417. 408.			547. 512.	0.139 0.144	0.267 0.263
G. E. Silicone ESM 1004AP	SG6 SG46	5,226 5,180	42.3 44.2			55.9	0.0070 0.0070	0.0264 0.0316
4	SG10 SG7	17,950 10,387	475. 417.				0.140	0.0435
	SG8 SG9	10,387 10,038 10,158	408. 408.		-	512.	0.140 0.144 0.144	0.257 0.263
Hughes Phenolic-Nylon H-5	PLH12 PLH91	4,933 5,180	44.4 44.2		-	.57,.5	0.0070 0.0070	0.0303 0.0316
	PLH7 PLH10 PLH11	18,642 17,950 18,445	455. 475. 475.				0.0341 0.0340 0.0340	0.0632 0.0435 0.0433
	PLH8 PLH9	10,137 10,647	417. 417.			547.	0.140 0.139	0.263 0.267
Calibration Runs	C1	4,824	42.8	40.7	40.0	56.8	0.0070	0.0290
	C2 C3	18,370 10,820	485.	457. 409.	438. 387.	530.	0.0340	0.0435
L	1	1 -0,020	L	1	1	<u> </u>		1

<sup>(1)</sup> Enthalpy measured by energy balance method.

<sup>(2)</sup> Facility steady state calorimeter, "Gardon" asymptotic type by Thermogage Inc., 1.25-in. diameter, flat face, sensing diameter 0.10-in. constantan.

<sup>(3)</sup> Martin design slug calorimeter, 0.25-in. diameter copper slug by 0.125-in. long set in 1.25-in. diameter flat face asbestos-phenolic body.

<sup>(4)</sup> Martin design slug calorimeter, 0.625-in. copper slug by 0.125-in. long set in 1.25-in. diameter flat face asbestos-phenolic body.

<sup>(5)</sup> Facility pitot probe, 0.625-in. diameter, water-cooled.

TEST CHAMBER PRESSURE (atm)	NOZZLE EXIT PRESSURE P <sub>e</sub>	GAS FLOW RATE (1b/sec)	MAXIMUM FRONT SURFACE TEMPERATURE $T_{FS} \in = 1 \text{ (°F)}$		RUN TIME (sec)	CORE WEIGHT LOSS	CORE CHAR WFIGHT (g)	CORE RECESSION (in.)	CHAR THICKNESS (in.)	CHAR DENSITY (1b/ft <sup>3</sup> )
	(atm)		Facility (6)	SRI (7)						
0.00096 0.00126	0.00106 0.00136	0.0030 0.0030	2690 2540	2600 2360	70 35 70	0.401 0.210 0.398	0.160 0.102 0.165	0.054 0.023 0.060	0.126 0.087 0.128	15.7 14.5 16.0
0.00065		0.0030	2700	2440	120	0.574	0.240	0.107	0.178	16.7
0.00379 0.00379	0.00459 0.00459	0.0040 0.0040	4310 4070	4230 3930	11 5	0.210 0.128	$0.138 \\ 0.075$	0.017 0.004	0.108 0.063	15.8 14.8
0.00921 0.00854	0.00921 0.00854	0.0251 0.0250	4770 4440	4520 4300	17 7	0.431 0.192	$0.219 \\ 0.127$	0.069 0.020	0.156 0.094	17.4 16.8
0.00096 0.00126 0.00965	0.00106 0.00136	0.0030 0.0030 0.0030	2780 2600 2620	2500 2430 2470	70 35 120	0.377 0.197 0.537	0.237 0.128 0.338	0.048 0.024 0.060	0.168 0.096 0.236	17.5 16.5 17.8
0.00379 0.00369	0.00459 0.00479	0.0040 0.0040	4340 4200	4280 4150	11 5	0.256 0.140	0.123 0.083	0.048 0.022	0.092 0.064	16.6 16.1
0.00921 0.00854 0.00914	0.00921 0.00854 0.00914	0.0251 0.0250 0.0249	4460 4580	4350 4400 4390	17 7 17	0.551 0.266 0.601	0.106 0.096 0.095	0.169 0.057 0.188	0.079 0.076 0.076	16.6 15.7 15.5
0.00125 0.00100 0.00105	0.00135 0.0009 0.00115	0.0030 0.0030 0.0030	2400 2430 2400	2330 2350 2040	35 70 120	0.184 0.275 0.442	0.123 0.187 0.253	+0.033 +0.049 +0.048	0.107 0.153 0.200	14.3 15.2 15.7
0.00369 0.00369 0.00380	0.00443 0.00479 0.00482	0.0040 0.0040 0.0040	3980 3910 3920	3940 3830 3670	11 5 13.4	0.286 0.129 0.344	0.057 0.054 0.078	0.042 0.000 0.053	0.052 0.052 0.063	13.6 12.9 15.4
0.00909 0.00789	0.00934 0.00789	0.0251 0.0250	4100	3840 3860	17 7	$0.710 \\ 0.294$	0.030 0.033	0.190 0.055	0.032 0.033	11.6 12.4
0.00100 0.00105	0.00090 0.00115	0.0030 0.0030	2430 2380	2320 2290	70 120	0.216 0.344	0.467 0.551	+0.006 +0.065	0.174 0.271	33.3 25.2
0.00369	0.00443	0.0040	3500	3370	11	0.457	0.093	0.129	0.030	38.4
0.00914 0.00855 0.00789	0.00914 0.00878 0.00789	0.0249 0.0250 0.0250	3480	3220 3250 3220	17 17 7	1.350 1.276 0.452	0.077 0.065 0.106	0.415 0.373 0.119	0.011 0.016 0.033	46.8 50.4 39.8
0.00125 0.00105	0.00135 0.00115	0.0030 0.0030	2660 2670	2500 2480	70 120	0.410 0.639	0.147 0.196	0.058 0.078	0.119 0.171	15.3 14.7
0.00380 0.00369 0.00369	0.00482 0.00443 0.00479	0.0040 0.0040 0.0040	4170 4270 4040	3940 4220 3850	11 11 5	0.226 0.225 0.125	0.134	0.025 0.015 0.004	0.103	16.1 14.2
0.00921 0.00909	0.00921 0.00934	0.0251 0.0251	4720	4490 4130	17 7	0.461 0.230	0.204 0.133	0.062 0.019	0.151 0.097	16.8 17.0
0.0010	0.0011									
0.00382	0.00442									:

<sup>(6)</sup> Facility Pyro 650 electronic optical pyrometer; viewed model through front port, approximately  $30^{\circ}$  to model surface plane.

<sup>(7)</sup> SRI radiometer, located inside test chamber; viewed model directly, approximately  $30^{\circ}$  to model surface plane.

Table B-10(b) (Concluded)

### TUNNEL CALIBRATION AND TEST DATA REPORTED BY MARTIN COMPANY FOR VARIOUS DIAMETER MODELS

Ref: Martin Report ER 14 426, November 17, 1966

	MODEL NO.	TOTAL ENTHALPY ht (Btu/lb)	MODEL STAGNATION PRESSURE Pt2	PLENUM PRESSURE P	TEST CHAMBER PRESSURE	GAS FLOW RATE (lb/sec)	MODEL DIAMETER AND SHAPE (in.)	MODEL CORE DIAMETER (in.)
		(1)	(atm)	(atm)	(atm)			
Teflon	T1 63 T1 62	12,149 12,432	0.0205 0.0203	0.241 0.240	0.00132 0.00132	0.04 0.04	1.0 Hemi. 1.0 Hemi.	0.5 0.5
	T158 T159	12,149 12,145	0.0205 0.0203	0.241 0.220	0.00132 0.00132	0.04 0.04	1.25 FF 1.25 FF	0.625 0.625
	T155 T154	12,246 12,246	0.0203 0.0203	$0.241 \\ 0.241$	0.00132 0.00132	0.04 0.04	2.5 FF 2.5 FF	$1.25 \\ 1.25$
	T150 T151	12,610 12,406	0.0203 0.0201	0.250 0.249	0.00132 0.00132	0.04 0.04	5.0 FF 5.0 FF	2.5 2.5
Hughes Phenolic-Nylon H-5	PLH163 PLH162	12,145 12,432	0.0203 0.0203	0.220 0.240	0.00132 0.00132	0.04 0.04	1.0 Hemi. 1.0 Hemi.	0.5 0.5
	PLH159 PLH158	$12,145 \\ 12,432$	0.0203 0.0203	0.220 0.240	0.00132 0.00132	$\substack{0.04\\0.04}$	1.25 FF 1.25 FF	0.625 0.625
	PLH155 PLH154	12,406 12,246	0.0201 0.0203	0.249 0.241	0.00132 0.00132	0.04 0.04	2.5 FF 2.5 FF	1.25 1.25
	PLH151 PLH150	12,610 12,406	0.0203 0.0201	0.249 0.250	0.00132 0.00132	0.04 0.04	5.0 FF 5.0 FF	2.5 2.5
Tunnel Calibration Runs	Cl	12.196	0.0200	0.296	0.00150	0.0401	***	
311221111111111111111111111111111111111	C1 C2 C3 C4 C5 C6	12,196 12,725 12,071 12,367 11,869 11,713	0.0201 0.0195 0.0204 0.0194 0.0194	0.289 0.283 0.265 0.271 0.278	0.00145 0.00132 0.00132 0.00126 0.00122	0.0401 0.04 0.04 0.04 0.04 0.04		

<sup>(1)</sup> Enthalpy determined by heat balance method.

<sup>(2)</sup> Thermogage heat sink calorimeter, 1.25-in. flat face plus adapter to 2.5-in. diameter.

<sup>(3)</sup> Thermogage steady state water-cooled asymptotic calorimeters: (1) 1-in. diameter hemisphere shape, (2) 1.25-in. diameter flat face plus adapters for 2.5-in. diameter and 5-in. diameter flat face. Constantan sensing diameter 0.125 in.

<sup>(4)</sup> SRI calorimeter.

HEAT TF RA q (Btu/f	TE CW		TEMPERATURE $\Gamma_{FS}  \epsilon = 1  (^{\circ}F)$		RUN TIME (sec)		CORE WEIGHT LOSS (g)	CORE CHAR WEIGHT (g)	CORE RECESSION (in.)	CHAR THICKNESS (in.)	CHAR DENSITY (1b/ft <sup>3</sup> )
	2)		Facility (6)	SRI (7)							
149 151					10 35		0.472 1.607		0.068 0.228		
149 155					20 70		0.751 3.055		0,067 0.269		
$\begin{array}{c} 151 \\ 151 \end{array}$					30 100		3.153 11.088		$0.070 \\ 0.248$		
156 156					40 140		12.004 43.032		0.067 0.229		
155 151			3,590 3,960	3,840	10 35		0.185 0.506	0.074 0.158	0.040 0.150	0.094 0.199	14.2 16.2
155 151			3,710 4,110	3,500 4,000	20 70		0.257 0.731	0.128 0.375	$0.027 \\ 0.144$	0.115 0.282	13.8 16.5
156 151			3,480 3,770	3,390 3,550	30 100		1.263 3.094	0.606 1.552	$0.029 \\ 0.120$	0.126 0.278	14.9 17.3
156 156			3,300 3,490	3,200 3,390	40 140	-	5.231 12.898	2.824 6.974	0.032 0.106	0.156 0.318	14.0 17.0
-,,							9	<del></del>			
THERA	MOGAGE	·	THE		RANSFER	RATE (B	tu/ft <sup>2</sup> sec)		ARTIN	······································	
	IMETER	THERMOGAGE CALORIMETER (3)				(4)		CALC	RIMETER (5)		
2.5 FF	1.25 FF	1.0 Hemi.	1.25 FF	2.5 FF	5.0 FF	1.25 FF	1.0 Hemi.	1.25 FF	2.5 FF	5.0 FF	
149 155 151	211	484	246 255	180	144	254	210	217	124	7.5	
171 155	211	500 384					318	240	132	85	
151 156	Ì	438				247	348 345		102		

(5) Martin transient calorimeter, copper slug set in asbestos-phenolic body.

Calorimeter Shape Overall Diameter (in.) Slug Diameter (in.)

alorimeter Shape	Overall Diameter (in.)	Slug Diame
Hemisphere	1.0	0.125
Flat Face	1.25	0.188
Flat Face	2.50	0.250
Flat Face	5.0	0.250

<sup>(6)</sup> Facility pyrometer Pyro 650 electronic optical pyrometer, located outside test chamber; viewed model through front port approximately 38° to plane of model front surface.

<sup>(7)</sup> SRI radiometer, located inside test chamber; viewed model directly, approximately 38° to plane of model front surface.

Table B-11 TUNNEL CALIBRATION AND TEST DATA REPORTED BY SPACE GENERAL CORPORATION Ref: SGC Report 1034-F1, July 1966

A CONTRACTOR OF THE STATE OF TH	MODEL NO.	TOTAL ENTHALPY h <sub>t</sub> (Btu/lb)		HEAT TRA RAT q <sub>CW</sub> (Btu/1	E	MODEL STAGNATION PRESSURE P	PLENUM PRESSURE P
				Calorin	eter	(atm)	(atm)
		(1)	(2)	Facility (3)	SRI	(4)	
Langley Phenolic-Nylon Scout R/4B	PLL22 PLL23 PLL94	14,925 14,920 14,855	14,990 14,850 14,990	99 97 103		0.00510 0.00509 0.00511	0.0266 0.0265 0.0266
	PLL55 PLL56	24,985 24,765	24, 130 25, 340	346 344		0.01947 0.01960	0.1210 0.1234
	PLL24 PLL28	5,117 5,129	5,010 5,020	158 150		0.092 0.092	0.514 0.514
Avcoat 5026-39	A23 A25 A97	15,035 15,100 14,955	14,850 14,850 15,000	101 95 101		0.00511 0.00509 0.00511	0.0265 0.0265 0.0267
	A24 A26	24,880 24,925	25,510 26,180	345 346		0.01974 0.01934	0.1237 0.1250
	A27 A28	5,063 5,134	5,005 5,020	157 155		0.093 0.092	0.512 0.514
Modified Purple Blend Silicone E4Al	SP22 SP24 SP94	14,855 15,050 14,925	15,110 14,990 14,990	98 103 105	8	0.00512 0.00511 0.00510	0.0268 0.0266 0.0266
	SP25 SP26	24,780 24,800	25,510 24,840	344 344		0.01974 0.01960	0.1237 0.1224
	SP27 SP28	5,127 5,093	5,015 5,020	157 155		0.093 0.092	0.513 0.514
G. E. Silicone ESM 1004AP	SG16 SG48	15,035 14,920	15,000 14,990	106 103		0.00511 0.00511	0.0264 0.0266
	SG20	24,888	24,130	345		0.01947	0.1210
	SG18 SG19	5,150 5,174	5,030 5,020	155 153		0.093 0.092	0.510 0.514
Hughes Phenolic-Nylon H-5	PLH19 PLH94	14,855 15,035	14,850 14,990	102 98		0.00510 0.00511	0.0265 0.0266
	PLH20 PLH21	24,910 24,875	24,840 25,510	345 345		0.01968 0.01972	0.1224 0.1237
	PLH22 PLH23	5,129 5,154	5,005 5,020	157 153		0.092 0.092	0.512 0.514
Calibration Runs	C1 C2 C3	15,025 24,925 5,150	14,850 25,510 5,020	101 346 157	98.7 344.8 162.7	0.00509 0.01972 0.0925	0.0265 0.1237 0.514

Enthalpy measured by energy balance method.
 Enthalpy calculated by SGC using sonic flow relationship: h<sub>t</sub> = (280P<sub>t1</sub>A\*/m)<sup>2.5</sup>.
 SGC steady state calorimeter, Hy-Cal Engineering asymptotic type, 0.10-in. diameter constantan sensing area in 1.25-in. diameter flat faced shroud.

<sup>(4)</sup> SGC pitot probe, water cooled, 0.5-in. diameter.

NOZZLE EXIT PRESSURE P e (atm)	GAS FLOW RATE (lb/sec)	MAXIMUM FRONT SURFACE TEMPERATURE T <sub>FS</sub> $\epsilon$ = 1 (°F)		RUN TIME (sec)	CORE WEIGHT LOSS (g)	CORE CHAR WEIGHT (g)	CORE RECESSION (in.)	CHAR THICKNESS (in.)	CHAR DENSITY (1b/ft <sup>3</sup> )
:	-	Facility (5)	SRI (6)						
0.000475 0.000473 0.000475	0.00088 0.00088 0.00088	2930 3100 3010	2580 . 2600	15.0 30.3 50.0	0.158 0.269 0.323	0.082 0.121 0.154	$0.012 \\ 0.031 \\ 0.054$	0.070 0.105 0.124	14.5 14.3 15.4
0.002108	0.00329	4340	3350	15.2	$0.465 \\ 0.264$	0.253	0.055	0.182	17.2
0.002197	0.00329	4340	3450	6.1		0.147	0.021	0.110	16.6
0.00965	0.0259	3980	3620	14.0	0.266	0.162	0.034	0.115	17.5
0.00968	0.0259	4260		33.0	0.515	0.247	0.093	0.170	18.0
0.000476	0.00088	3125	2550	30.2	0.272	0.167	0.044	0.125	16.6
0.000476	0.00088	3050	2420	15.3	0.162	0.118	0.019	0.078	18.7
0.000477	0.00088	3080	2600	50.4	0.392	0.240	0.063	0.168	17.7
0.002237	0.00329	4100	3300	6.1	0.222	0.131	0.035	0.099	16.4
0.001974	0.00329	4385	3200	15.0	0.462	0.205	0.103	0.148	17.2
0.00971	0.0259	4060		14.0	0.310	0.139	0.070	0.098	17.6
0.00968	0.0259	4130		33.0	0.695	0.141	0.242	0.090	19.4
0.000475	0.00088	2980	2460	15.4	0.136	0.108	+0.039	0.097	13.8
0.000475	0.00088	3190	2540	30.2	0.178	0.141	+0.055	0.133	13.1
0.000477	0.00088	2975	2300	50.3	0.349	0.191	+0.033	0.167	14.2
0.002237	0.00329 0.00329	3860 3950	2950 3000	6.1 15.0	0.238 0.537	$0.101 \\ 0.121$	0.005 0.101	0.089 0.091	14.1 16.5
0.00968	0.0259	3835	3540	14.0	0.333	0.104	0.052	0.077	16.8
0.00965	0.0259	3865		33.0	0.732	0.100	0.176	0.077	16.1
0.00475	0.00088	2880	2450	30.0	0.173	0.313	0.008	0.123	31.6
0.00477	0.00088	3010		50.2	0.228	0.369	0.029	0.160	28.6
0.001974	0.00329 0.0259	3520 3435	2800 3080	15.1 14.0	0.810 0.534	0.097 0.113	0.228	0.030	40.1
0.00968	0.0259	3425	3130	33.0	1.526	0.073	0.151 0.451	0.036 0.019	38.9 47.6
0.000476 0.000477	0.00088 0.00088	2800 2980	2540	30.3 50.0	0.259 0.374	$0.115 \\ 0.151$	0.025 0.049	0.098 0.122	14.6 15.3
0.002276	0.00329	4060	3300	6.1	0.208	0.123	0.015	0.095	16.1
0.002237	0.00329	4180	3500	15.1	0.425	0.258	0.047	0.181	17.7
0.00970	0.0259	4035	3640	14.0	0.265	0.154	0.022	0.118	16.2
0.00966	0.0259	4190	.3640	33.0	0.580	0.229	0.090	0.172	16.5
0.000477 0.002237 0.00967	0.00088 0.00329 0.0259								

<sup>(5)</sup> L-N optical pyrometer (0.655 microns); viewed model through front quartz port, approximately  $18\,^{\circ}$  to plane of model front.

<sup>(6)</sup> SRI radiometer located outside test chamber. For test condition 1 (h = 15,000, q = 100) viewed model through front port, approximately 18° to plane of model front surface. For remaining two test conditions, model was viewed through a side port 40° to jet axis, thence off front mirror 40° to plane of model front surface.

Table B-12

TUNNEL CALIBRATION AND TEST DATA REPORTED BY CORNELL, AERONAUTICAL LABORATORY INC.

Ref: Letter Report Dated September 19, 1966

	MODEL NO.	SPECIFIC ENTHALPY (Btu/lb)	HEAT TRANSFER RATE  q <sub>CW</sub> (Btu/ft <sup>2</sup> sec)  Calorimeter			MODEL STAGNATION PRESSURE P (atm) 2	GAS FLOW RATE (1b/sec)	RESERVOIR PRESSURE P <sub>t</sub> (atm)	STAGNATION TEMPERATURE (°R)
			Fac	ility	SRI	. ,		2 m 3 2 m 3 2 m 3 3 m 4 2 m 3 2 m 3 3 3 m 3 3 m 3 3 3 m 3 3 m 3 3 3 m 3 3 m 3 3 3 m 3 3 m	ora i rugi Burg
				(4)	(6)				Maria Maria
Teflon	T118 T119	2060 2080		618 624	1091 1108	10.2 10.3	5.0	101.7 102.5	6100 6120
; 	<b>T</b> 121	1940		1067	1740	29.5		98.4	5840
Phenolic-Nylon (75 lb/ft <sup>3</sup> )	P12B1 P12B6	2120 2080		636 624	1115 1070	10.0 9.6	5.0	100,40 95.8	6200 6120
	P12B4	1870	-	1028	1666	29.3		97.7	5700
Langley Phenolic-Nylon Scout R/4B	PLL44	2060		618	1075	9.9		99.0	6100
Avcoat 5026-39	A43	2040		612	1059	9.8		97.9	6050
Tunnel Calibration Pun	C1	1950			1010	9.8		98.0	5860
:	C2 C3	1870 2100	1700 <sup>(1)</sup> 2720	650 <sup>(5)</sup> 1040		19.1 32.4			
:	C4 C5	2050 1910	5060 <sup>(2)</sup> 5450	1760 1900		53.0 64.0			
	C6 C7	2010 1810	4050 <sup>(3)</sup> 4740	1990 2330		53.0 63.0		)	

- (1) Cornell transient calorimeter, 0.3-in. nose radius, hemispherical shape, 0.090-in. diameter by 0.125-in. long OFHC copper slug potted in 0.125-in. diameter bore with insulating cement, Ch-Al thermocouple spot welded to back face.
- (2) Cornell calorimeter, 0.25-in. nose radius, hemispherical shape, 0.150-in. diameter by 0.506-in. long OFHC copper slug with 0.020-in. long flanges at each end for press fit into 0.160-in. diameter bore. Sheathed Ch-Al exposed junction thermocouple inserted into 0.022-in. diameter hole from rear of slug to within 0.020-in. from gage front face and gold soldered in place. Data reduction is by finite differences scheme on IBM computer.
- (3) Identical to (2) except for 0.5-in. nose radius.
- (4) Estimated from the relation furnished by Cornell:  $\dot{q} = 0.3(H_S H_W)$  at  $P_{t_2} = 10$  atm and  $\dot{q} = 0.55(H_S H_W)$  at  $P_{t_2} = 30$  atm.
- (5) Heat flux adjusted to 1.25-in. flat face: (1)  $0.55 \stackrel{.}{q}_1 0.6/1.25)^{0.5} = 0.382 \stackrel{.}{q}_1$ , (2)  $0.55 \stackrel{.}{q}_2 (0.5/1.25)^{0.5} = 0.348 \stackrel{.}{q}_2$ , (3)  $0.55 \stackrel{.}{q}_3 (1.0/1.25)^{0.5} = 0.492 \stackrel{.}{q}_3$ .
- (6) Estimated from Run Cl and the relation:  $\dot{q}$  = 0.0744  $(P_{t_0}/R_{eff})^{0.5}\Delta h$ .

STAGNATION DENSITY Ps (slugs/ft <sup>3</sup> )	MACH NO. M <sub>∞</sub>	FREE STREAM PRESSURE P <sub>©</sub> (atm)	FREE STREAM TEMPER- ATURE T <sub>∞</sub> (°R)	FREE STREAM DENSITY  Pool (slugs/ft <sup>3</sup> )	DISTANCE ROTOR TO MODEL (in.)	MAXIMUM FRONT SURFACE TEMPER-ATURE TFS $\epsilon = 1$ (°F)	RUN TIME (sec)	CORE WEIGHT LOSS (g)	CORE CHAR WEIGHT (g)	CORE RECESSION (in.)	CHAR THICKNESS (in.)
0.00200 0.00201	3.67 3.67	0.549 0.554	2370 2380	0.00285 0.00286	4 4	1080 1290	2.0 4.7	1.139 2.674		0.103 0.241	
0.0061	2.72	3.00	2360	0.00156	2,25		4.0	(8)			
0.00193 0.00188	3.67 3.67	0.542 0.517	2420 2380	0.00276 0.00267	4 4	4030 4400	2.1 6.0	0.611 1.727	0.000	0.099 0.280	0.000
0.0062	2.72	2.98	2300	0.00159	2, 25	4400	3.1	2.903	0.000	0.470	0.000
0.00194	3.67	0.535	2370	0.00278	4	4200	4.1	(8)			
0.00195	3.67	0.528	2360	0.00275	4	4050	4.0	(8)			
0.00202	3.67	0.530	2280	0.00286	4		2.2				

<sup>(7)</sup> Facility Thermodot optical pyrometer; views model front surface through rotor tubes, 1.6 to 2.7 microns.

<sup>(8)</sup> Model disintegrated during test.

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#### APPENDIX C

#### MODEL TEMPERATURE DATA

This appendix contains internal and external temperature data reported by the participating facilities for the models instrumented with thermocouples. The data were taken from the temperature plots reported by the facilities, and sufficient data have been included to allow reproduction of the original curves. The tunnel operating data for each model may be obtained by consulting the appropriate facility data table and model number in Appendix B. The various materials are designated by the model prefix letters described earlier.

FACILITY	MODEL NO.	INITIAL THERMO- COUPLE DISTANCE FROM MODEL FRONT FACE (in.) (1)	THERMO- COUPLE TEMPERA- TURE RISE (°F)	TIME (sec)	FRONT SURFACE TEMPERA- TURE $\epsilon = 1$ (°F)	CHAR, VIRGIN MATERIAL INTERFACE TEMPERA- TURE (°F) (4)	TIME CHAR INTERFACE PASSED THERMO- COUPLE (sec)
GDB-Ames	A93	0.113	80 370 790 1280 1680 1930 2410	5 10 15 20 25 30 40	2660 2760 2790 2830 2850 2880 2900	1190	20
		0,226	50 90 170 300 450 550	30 40 50 60 70 75	2930 2900 2930	800	60
		0.330	60 80 120	50 60 70			
MPDB-Ames	A84	0.104	30 170 540 1130	5 10 15 20		1380	22
		0.222	1 590 20 75 280	25 10 20 30		1280	50
		0.305	630 1110 10 20 50 120	40 50 10 20 30 40			
	:	0.410	240 5 20 35	50 10 30 50			
MPDB-Ames	A85	0.103	20 230 690 1590	4 8 12 16	2700 3010 3150 3220	1460	16
		0.211	2070 10 15 50	20 5 10 15	3280		
		0.321	1 40 10 15 20 1 20 1 30	20 5 10 15 20 25			
		0.424	5 10 20 20 20 20 25	5 10 15 15 20 25			
AMPD-Langley	A90	0.107	30 -70 150 320 790 1420	4 6 8 10 12 14	3920		

Appendix C (Continued)

FACILITY	MODEL NO.	INITIAL THERMO- COUPLE DISTANCE FROM MODEL FRONT FACE (in.)	THERMO- COUPLE TEMPERA- TURE RISE (°F)	TIME (sec)	FRONT SURFACE TEMPERA- TURE ∈ = 1 (°F)	CHAR, VIRGIN MATERIAL INTERFACE TEMPERA- TURE (°F) (4)	TIME CHAR INTERFACE PASSED THERMO- COUPLE (sec)
	-	(1)	(2)		(3)	<del></del>	
AMPD-Langley	A90	0.209 0.311 0.420	40 70 140 270 580 5 40 70 200 5	6 8 10 12 14 5 10 15 20 10 20		1900	16.8
Aerotherm	A98 :	0., 113	100 480 960 1430 1820	5 10 15 20 25	2700 2810 2860	1490	20
		0.215	40 130 360 810 1400	10 20 30 40 50	2910 2950 3000	1140	46
	:	0.313	20 130 530 1310 10 30 110 300 820	20 40 60 80 20 40 60 80 100	3170 3170	990	72 104
Giannini	A94	0.101	5 30 190 360 810 1210	2 5 8 10 13 15	2450 2800 2950 3100	1160	14.4
		0.213	10 120 290 600 1090	10 20 25 30 35	3210 3360	1070	34
Martin	<b>A</b> 95	0.103	30 200 410 660 910	5 10 15 20 25	2370 · 2500	1190	29
		0.216	910 1200 60 170 330 520 730	30 20 30 40 50	2600 2620	1060	72
		0.314	10 60 200 390 630	20 40 60 80 100 40	2600 2620		
		0.415	10 90 370	80 120			

Appendix C (Continued)

FACILITY	MODEL NO.	INITIAL THERMO- COUPLE DISTANCE FROM MODEL FRONT FACE (in.)	THERMO- COUPLE TEMPERA- TURE RISE (°F)	TIME (sec)	FRONT SURFACE TEMPERA- TURE $\epsilon = 1 (^{\circ}F)$	CHAR, VIRGIN MATERIAL INTERFACE TEMPERA- TURE (°F) (4)	TIME CHAR INTERFACE PASSED THERMO- COUPLE (sec)
		(1)	(2)		(3)		
Space General	A97	0.110	70 290 690 1250 1700 70 140 260 450 670 940 1200	5 10 15 20 25 15 20 25 30 35 40 45 30 40	2950 3000 3050 3070 3100	1000	17.5 41.5
GDB-Ames	PLL96	0.094	100 250 80 250 640 1170 1580 50 200 400	50 50 10 15 20 25 20 40 50	2400 2590 2690 2770 2840 2990 3040	1200	20
:		0.328	620 1140 60 100 110 170 10 50 90	60 70 30 50 60 70 30 50 90	3020 3020		
MPDB-Ames	PLL87	0.095	10 50 170 370 630	5 10 15 20 25 10		800	25.5
		0.220	10 50 80 130 250 5 10 40	20 30 40 50 10 20 30 40 50			
MPDB-Ames	PLL89	0.149	90 20 40 90 220 480			1320	25
		0.234	90 220 480 880 5 15 30	24 5 10 15 20			
		0.332	50 5 10 15 20 5 10 15 20	4 8 12 16 20 24 5 10 15 20 5 10 15 20 5 10			

Appendix C (Continued)

FACILITY	MODEL NO.	INITIAL THERMO- COUPLE DISTANCE FROM MODEL FRONT FACE (in.) (1)	THERMO- COUPLE TEMPERA- TURE RISE (°F)	TIME (sec)	FRONT SURFACE TEMPERA- TURE $ \epsilon = 1 (^{\circ}F) $	CHAR, VIRGIN MATERIAL INTERFACE TEMPERA- TURE (°F) (4)	TIME CHAR INTERFACE PASSED THERMO- COUPLE (sec)
AMPD-Langley	PLL93	0.114	70	4 6		1400	11.6
		0.198	110 280 720 1430 70 100 310 640 1280 30 70 80	8 10 12 10 15 20 22 24 15 20 25	3940	1420	24. 2
Aerotherm	PLL97	0.095	50	5	0750	1060	19
			250 650	10 15	2750		*
		0.220	$1180 \\ 1600 \\ 20$	20 25 10	.2880	980	58
	1	0, 310	70 120 250 510 40 70 150	20 30 40 50 20 40 60	2900 2960 3050 3130	940	89
		0. 399	500 1600 40 70 110 170 820	80 100 20 40 60 80 100	3130		
Giannini	PLL90	0.119	10 180 950	5 10 15	3100 3450 3420 3580	1200	16.2
	:	0.220	1450 15 70 130 360 960	20 10 20 25 30 35	3580 3610 3670 3660	800	33.4
Martin	PLL91	0.111	20 70 130	5 10 15 20	2200 2320	980	35
		0.221	180 290 1150 40 140 300	30 40 20 40 60	2460 2560	840	91
		0.314	560 1010 20 50	80 100 20 40 60	2560 2640 2700		
,		0.415	90 160 380 30 80 170	80 100 40 80 120			

Appendix C (Continued)

FACILITY	MODEL	INITIAL	TUEDVO	TIVE	EDONT	СПУВ	TIME COLD
FACILITY	NO.	INITIAL THERMO- COUPLE DISTANCE FROM MODEL FRONT FACE (in.) (1)	THERMO- COUPLE TEMPERA- TURE RISE (°F)	TIME (sec)	FRONT SURFACE TEMPERA- TURE ∈ = 1 (°F)	CHAR, VIRGIN MATERIAL INTERFACE TEMPERA- TURE (°F) (4)	TIME CHAR INTERFACE PASSÉD THERMO- COUPLE (sec)
Space General	PLL94	0.104		-	(3)	960	21
Space General	PLL94	0.211	50 170 460 800 1170 20 70 210 450 30 50 80	5 10 15 20 25 20 30 40 50 30 40 50	2800 2860 2900 2920 2940 2960 3010	1000	68
GDB-Ames	PLH98	0.115	30 120 290 530 880 1360	5 10 15 20 25 30 40	2440 2650 2740 2800 2840 2880 2940	1170	28
		0.212	2070 110 240 460 850 1540 1740	30 40 50 60 70 75	2980 3040 3060	1100	62
•		0.314	40 70 110 200 10 30 50	40 50 60 70 30 50 70			
MPDB-Ames	PLH97	0.115	20 50 130 270 510 840 1260	5 10 15 20 25 30 35 10		840	30
		0.314	10 25 50 125 200 5 15 40	20 30 40 50 10 30 50			
		-	10 25	30 50			
AMPD-Langley	PLH93	0.114	20 80 210 580 1260 5 40	4 6 8 10 12 10 15	3940	1540 1440	12.6 27.6
		0.309	150 660 10 25 100	15 20 25 20 25 20 25 30			

Appendix C (Continued)

FACILITY	MODEL NO.	INITIAL THERMO COUPLE DISTANCE FROM MODEL FRONT FACE (in.)	THERMO- COUPLE TEMPERA- TURE RISE (°F)	TIME (sec)	FRONT SURFACE TEMPERA- TURE $\epsilon = 1 \text{ (°F)}$	CHAR, VIRGIN MATERIAL INTERFACE TEMPERA- TURE (°F)	TIME CHAR INTERFACE PASSED THERMO- COUPLE (sec)
		(1)	(2)		(3)	(4)	\$ 1000
Giannini	PLH90	0.111	10 110 360 870 1400 10 40 90 160 260	5 10 15 20 25 10 20 25 30 35	2300 2750 2950 3130 3200 3370	960	19.6
Martin	PLH91	0.115	90	10	2160	1130	45
			150 250 380 510 890 1280	15 20 25 30 40 50	2250 2380		
		0.211	30 140 300 590	20 40 60 80	2480	1160	101
		0.313	10 20 70 130 240 10 40 130	20 40 60 80 100 40 80 120	2560 2640		
Space General	PLH94	0.111	110 270 520 850 1280 40 130 240 440	10 15 20 25 30 20 30 40 50	2750 2840 2900 2940 2940 2940 2960 2970	1060	27
GDB-Ames	SP96	0.095	50 170 310 490 840 1120 1280 100 150	5 10 15 20 30 40 50 30 40	2540 2540 2540 2510 2470 2430 2400	1190	41.5
		0.225	200 280 350 400	50 60 70 75	2360 2340		
		0.337	20 60 100 10 20 50	30 50 70 30 50			
MPDB-Ames	SP89	0.120	50 60 210 520 1160 1680	70 5 10 15 20 25	2610 2850 2880 2890 2890	1280	20.5

Appendix C (Continued)

FACILITY	MODEL NO.	INITIAL THERMO- COUPLE DISTANCE FROM MODEL FRONT FACE (in.)	THERMO- COUPLE TEMPERA- TURE RISE (°F)	TIME (sec)	FRONT SURFACE TEMPERA- TURE	CHAR, VIRGIN MATERIAL INTERFACE TEMPERA- TURE	TIME CHAR INTERFACE PASSED THERMO- COUPLE (sec)
		(1)	(2)		(3)	(4)	
MPDB-Ames	SP89	0.241 0.311 0.421	5 10 25 50 75 2 4 6 2 4	5 10 15 20 25 15 25 5 15	•		
			6	25		1000	24
MPDB-Ames	SP85	0.099	60 210 410 680 1020 1300 40 110 240 460	5 10 15 20 35 30 10 20 30 40		1020	24
		0.325	740 10 30 70	50 10 30 50			
		0.416	5 10 20	1 0 30 50			
AMPD Langley	SP93	0.085	20 110 280 580 1000	2 4 6 8 10	2000	1580	12.6
		0.189	1510 20 70 170 410	12 10 15 20 25	3600	1440	30.6
Aerotherm	SP97	0,101	100 350	5 10	2140 2280		
		0.208	600 20 50 100 200 310 420 20	15 10 20 30 40 50 60	2290 2270 2260 2250 2240	1190	38
		0.303	140 250 330	20 40 60 80 100			
		0.409	20 50 100 140 210	20 40 60 80 100			
Martin	SP91	0.097	70 220 340 470 550 610	10 20 30 40 50 60	2230 2290 2320 2380	740	62

Appendix C (Continued

FACILITY	MODEL NO.	INITIAL THERMO- COUPLE DISTANCE FROM MODEL FRONT FACE (in.)	THERMO- COUPLE TEMPERA- TURE RISE (°F)	TIME (sec)	FRONT SURFACE TEMPERA- TURE $\epsilon = 1$ (°F)	CHAR, VIRGIN MATERIAL INTERFACE TEMPERA- TURE (°F) (4)	TIME CHAR INTERFACE PASSED THERMO- COUPLE (sec)
		(1)	(2)	<del> </del>	(3)	(4)	
Martin	SP91	0.198 0.314 0.411	40 160 320 480 20 120 260 10 50	20 40 60 80 40 80 120 40 80 120	2390 2400		
Giannini	SP90	0.099	20 200 500 1000 1400 30 100	5 10 15 20 25 20 30	2600 2800 2850 3000	1340	22.7
Space General	SP94	0.097	80 210 450 770 1070 1300 1460 50	5 10 15 20 25 30 35 20	2720 2800 2840 2870 2900 2920 2940	1460	33.5
GDB Ames	SG39	0.091	140 250 370 250 610 1020 1410	30 40 50 5 10 15 20	2950 2960 2540 2590 2600 2600	1540	26.5
		0.220	1670 130 270 430 590 790 980 1040	25 20 30 40 50 60 70 75	2610 2630 2640 2650 2650		
		0.318	90 150 210 300 390 40 100 160	30 40 50 60 70 30 50			
MPDB-Ames	SG53	0.099	190 560 1080 1620 1960 10 35 90	5 10 15 20 25 5 10 15	2680 2780 2840 2870 2870	1240	14
		0.308 0.411	310 5 30 60 2 10 25	20 25 5 15 25 5 15 25			

Appendix C (Continued)

FACILITY	MODEL NO.	INITIAL THERMO- COUPLE DISTANCE FROM MODEL FRONT FACE	THERMO- COUPLE TEMPERA- TURE RISE (°F)	TIME (sec)	FRONT SURFACE TEMPERA- TURE $\epsilon = 1 \text{ (°F)}$	CHAR, VIRGIN MATERIAL INTERFACE TEMPERA- TURE (°F)	TIME CHAR INTERFACE PASSED THERMO- COUPLE (sec)
		(in.) (1)	(2)		(3)	(4)	
MPDB-Ames	SG51	0.089 0.215	100 590 1100 1440 1650 40	5 10 15 20 25 10	-	1520	19.2
		0.313	130 300 530 800 10 20 60 110	20 30 40 50 10 20 30 40			
		0.405	170 5 10 20 40 60	50 10 20 30 40 50			
AMPD-Langley	SG49	0.110	30 190 430 770	2 4 6 8		1840	12.4
		0.205	1220 5 65 220	10 5 10 15	2570	1440	23.4
		0.288	550 5 25 80 190	20 10 15 20 25			
Martin	SG46	0.114	80 160 250 700	2 4 6 8		1410	47
		0.218	1220 1610 100 210 340	10 15 5 10 15	2240	· ·	
		0.399	490 830 1160 130 310	20 30 40 20	2300 2320		
			550 900 40 180 450	40 60 80 40 80 120	2340 2360		
Space General	:SG48	0.168	80 330 690 1100	10 20 30 40	2700 2780 2850 2920 2980	1300	45
		0.191	1450 50 200 440 710 960	50 10 20 30 40 50	2980	1250	60

Appendix C (Concluded)

FACILITY	MODEL NO.	INITIAL THERMO- COUPLE DISTANCE FROM MODEL FRONT FACE (in.) (1)	THERMO- COUPLE TEMPERA- TURE RISE (°F)	TIME (sec)	FRONT SURFACE TEMP ERA- TURE $ \epsilon = 1 \ (^{\circ}F) $	CHAR, VIRGIN MATERIAL INTERFACE TEMPERA- TURE (°F) (4)	TIME CHAR INTERFACE PASSED THERMO- COUPLE (sec)
Space General	SG48	0.310	50 130 230	30 40 50			
Giannini	SG45	0.101	120 500 940 1500 1900	5 10 15 20 25	2800 2850 2900 2900 3000	1300	17.4
		0.215	20 150 410	10 20 30			-

- (1) Thermocouple distance from original model face determined from X-ray photographs.
- (2) Thermocouple temperature minus original starting model temperature at the time indicated in adjacent column.
- (3) Front surface temperature measured with facility optical pyrometer at the time indicated in the preceding column.
- (4) Char back face and virgin material interface temperature determined by method described in Section IV-C.

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#### APPENDIX D

#### SUMMARY OF PHASE II CORRELATION DATA

This appendix tabulates, by material, information calculated from the data in Appendices B and C. This information was used in preparing the various graphs and correlations appearing in this report. Where multiple runs are shown, by listing more than one model number on the same line, the values represent averages of the available data.

Appendix D

SUMMARY OF CORRELATION DATA

FACILITY MODEL	EL NO. MASS LOSS (1b/ft <sup>2</sup>	RATES sec)	HEAT TRANSFER RATE (Btu/ft <sup>2</sup> sec)		MODEL STAGNATION PRESSURE (atm)	ENTHALPY (Btu/lb)		EFFECTIVE MODEL RADIUS (ft)	FRONT SURFACE TEMPERATURE	gNI/mt CW/ Btu/lb
	• <b>B</b>	• ⊑	°SRI CW	°. GW	Pt2	∆hmeas CW (1)	∆h <sub>calc</sub> SRI CW (2)	Reff	ı	
PLL54,57, PLL58,59 PLL60,61	0.00716 0.0104 0.0111	$\begin{array}{c} 0.00629 \\ 0.0114 \\ 0.0140 \end{array}$	81 118 178	68 113.5 141.9	0.0109 0.0106 0.0182	10,270 22,680 16,440	7,721 11,406 13,130	$\begin{array}{c} 0.172 \\ 0.172 \\ 0.172 \\ 0.172 \end{array}$	3,490 3,720 4,120	11,313 11,346 16,036
PLL70, 69 PLL72, 71	0.00686 0.0111	0.0094 0.0129	117.1	96.3 169.1	0.00572	12,664 19,638	15,408 23,435	$0.172 \\ 0.172$		17,070 19,117
PLL30, 32, PLL29, 30 PLL34, 35	2, 93, 33 0. 0240 0. 0303 0. 0556	0.0238 0.0300 0.0505	263 526 1,115		0.284 0.293 0.735	4,750 9,550 9,550	4, 911 9, 670 12, 940	0.172 0.172 0.172	4,560 5,270 5,690	10,958 17,350 20,100
PLL54,52	0.0257		203	177	0.281	3,100	3,811	0172		7,898
PLL65 PLL66 PLL67	0.0143 0.0190 0.0186		410 810 1,020	436 800 988	0.0117 0.0170 0.0192	11,350 24,350 33,850	37,722 61,820 73,258	0.172 0.172 0.172	4,350 4,410 4,270	28,671 42,632 54,839
PLL38, 39 PLL40, 41 PLL36, 37, 9 PLL42	0.0211 0.0245 0.0131 0.0151	0.0273 0.0263 0.0140		403 595 100 205	0000	8,251 9,452 4,336 5,475		$\begin{array}{c} 0.172 \\ 0.172 \\ 0.172 \\ 0.172 \\ 0.172 \end{array}$	4,840 4,680 3,510 3,565	
PLL43, 46, PLL45, 48 PLL15, 18	$\begin{array}{c c} , 97 & 0.00721 \\ 0.0116 \\ 0.0161 \end{array}$	0.00905 0.0139 0.0303	80.1 188 947	88 270 1,160	0.0204 0.0301 0.0427	4,598 7,462 21,260	5, 581 10, 784 45, 608	$\begin{array}{c} 0.172 \\ 0.172 \\ 0.172 \\ 0.172 \end{array}$	3,670 4,300 5,550	11,110 16,207 58,820
PLL47	0.0756	<del>,</del>	1,850	1,960	1.93	6,850	13,252	0.172	6,010	24,471
,2,90 ,4 ,6,7	0.0100 0.0051 0.0187	0.0234 0.0063 0.0316	138 54 457	145 645 457	0.0198 0.0041 0.0945	10,050 9,935 15,345	9,760 8,392 14,794	$0.172 \\ 0.172 \\ 0.172 \\ 0.172$	4,140 2,940 4,690	13,800 10,588 24,439
PLL13,12,9 PLL11,10 PLL9,8	, 91 0.00435 0.0142 0.0248	0.00576 0.0286 0.0329	500 500 500	44.1 456 417.5	0.0070 0.0333 0.142	4,793 17,817 10,242	7,136 27,268 13,204	0.172 0.172 0.172	3,160 4,770 5,230	13,793 35,211 20,161
PLL22, 23 PLL55, 56 PLL24, 28	, 94 0.0051 0.0228 0.0136	0.00632 0.0345 0.0178	98 344 156	99 345 154	0.0051 0.0195 0.092	14,750 24,720 4,973	13,656 24,497 5,118	$\begin{array}{c} 0.172 \\ 0.172 \\ 0.172 \\ 0.172 \end{array}$	3,560 4,800 4,720	19,216 15,088 11,471
PLH6, 98 PLH42, 43 PLH24, 50	0.00940 0.00973 0.0170	0.00849 0.0113 0.0197	81 125 166	67.5 129 163.1	0.0108 0.0102 0.0185	10,300 22,600 16,470		0.172 0.172 0.172	3,550 3,960 3,930	8,617 12,847 9,764
PLH62, 61, 97 PLH65, 63, 64		0.00838	92.2	85.7	0.00582	12,111 $19,094$		$0.172 \\ 0.172$		11,000 20,731

MATERIAL	FACILITY	MODEL NO.	MASS LOSS RATES (1b/ft <sup>2</sup> sec)	S RATES sec)	HEAT TRANSFER RATE (Btu/ft <sup>2</sup> sec)		MODEL STAGNATION PRESSURE	ENTHALPY (Btu/lb)	LPY 1b)	EFFECTIVE MODEL RADIUS (ft.)	FRONT SURFACE TEMPERATURE (OB) e = 1	q SRI/mt CW Rrn/lb
							Pt.			Reff	TFS	ar /nag
			۵۰	• E	q.SRI	9FAC		Δh megas CW	∆h <sub>calc</sub> SRI			
						<u>.</u>		(1)	(2)	(3)		
Hughes Phenolic-Nylon H-5 (PLH) (continued)	AMPD-Langley	PLH26, 93, 28 PLH25, 27 PLH29, 30	0.0228 0.0280 0.0491	0.0219 0.0280 0.0437	267 549 1,139		0.284 0.293 0.735	4,750 9,550 9,550		0.172 0.172 0.172	4,500 5,190 5,780	11,711 19,607 23,198
	ESB-Langley	PLH41,40	0.0196		203	177	0.281	2,955		0.172		
	MSC-Houston (supersonic)	PLHS4 PLHS5 PLHS6	0.0154 0.0200 0.0211		400 808 1,040	431 797 ,000	0.0117 0.0170 0.0192	11,350 24,450 34,000		0.172 0.172 0.172	4,330 4,265 4,300	25, 947 40, 400 49, 298
	MSC Houston (subsonic)	PLH32, 33 PLH34, 35 PLH31 PLH31	0.0220 0.0245 0.0136 0.0174	0.0280	<del>,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,,</del>	408 600 101 205	0.000	8, 085 9, 275 4, 493 4, 980		0.172 0.172 0.172 0.172	4,880 4,595 3,440 3,660	<del>1 1</del>
	Aerotherm	PLH44, 45 PLH46, 48 PLH16, 17	$0.00877 \\ 0.0122 \\ 0.0203$	0.00916 0.0154 0.0313	58.2 172 967 1,	87.2 264 ,180	0.0197 0.0298 0.0419	4,178 7,983 21,155		0.172 0.172 0.172	2,930 3,550 5,030	6,636 14,098 47,635
	Giannini	PLH1, 90 PLH2, 3 PLH4, 5	0.00750 0.00512 0.0237	0.0109 0.0078 0.0330	139 55 456	146 65 456	0.0199 0.0041 0.0945	10,050 9,950 15,240	<del></del>	0.172 0.172 0.172	3,810 3,010 4,660	18,533 10,742 19,241
	Martin	PLH12, 91 PLH10, 11, 7 PLH9, 8	0.00474 0.0172 0.0240	0.00425 0.0310 0.0280	57.5 510 547	44.3 468.3 417	0.0070 0.0340 0.1395	4,757 18,046 10,102		0.172 0.172 0.172	3,130 4,730 5,180	12, 131 29, 651 22, 792
	Space General	PLH19, 94 PLH20, 21 PLH22, 23	0.00606 0.0249 0.0173	$\begin{array}{c} 0.00720 \\ 0.0387 \\ 0.0189 \end{array}$	99 344 157	100 345 155	0.00511 0.0197 0.092	14, 795 24, 740 4, 990	-	0.172 0.172 0.172	3,440 4,640 4,650	16, 337 13, 815 9, 075
	AMPD-Langley	PLH153, 154 PLH156, 157 PLH160, 161 PLH164, 165	0.0050 0.00605 0.0091 0.0158	,	78.5 89.5 126.5 271		0.0192 0.0192 0.0192 0.0192	9,500 8,900 9,400	<del> </del>	0.550 0.344 0.172 0.0416		15,700 14,800 13,900 17,200
	Martin	PLH162, 163 PLH158, 159 PLH154, 155 PLH150, 151	0.00496 0.00674 0.00980 0.0193		250	144 180 250 451	0.0203 0.0203 0.0202 0.0202	12,300 12,300 12,300 12,500	<del>(111) (111)</del>	0.688 0.344 0.172 0.0416	3,950 4,230 4,570 4,420	25,500
'Avcoat 5026-39, HC/G (A)	GDB-Ames	A53, 54, 61, 93 A55, 56 A57, 60	$\begin{array}{c} 0.0066 \\ 0.0117 \\ 0.00462 \end{array}$	0.0122 0.0153 0.00658	121 177 84	122.5 165 66	0.0102 0.0185 0.0108	21, 970 16, 520 10, 380	11,923 12,950 8,044	0.172 0.172 0.172	4,010 4,120 3,380	18, 333 15, 128 18, 182
	MPDB-Ames	A72, 71, 84 A76, 75, 85	0.00654 0.0113	0.00998 0.0180	105.3 215.2	97.6 188.2	0.00582 0.00947	13, 109 19, 342	13,736 22,000	0.172	3,790	16, 101 19, 044

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gSRI/mc CW/ Bcu/1b		8,940 11,550 9,410		8,423 24,096 44,144		12, 980 42, 773 8, 885	10,335	14,712 14,745 20,133	14,479 15,250 15,576	14,203 12,321 7,523	20,305 23,750 16,226	18,649	14,862 9,732 10,240	24,726
FRONT SURFACE TEMPERATURE (°R) $\epsilon$ = 1	S.	4,380 5,060 5,140		4,340	4,770 4,560 3,860 3,700	4,260 5,700 3,650	5,360	3,760 2,850 4,610	3,240 4,800 5,040	3,585 4,845 4,590	3,140 3,540 3,790	3,440	4,060 4,600 4,630	
S, Æ	<sup>n</sup> eff	0.172 0.172 0.172	0.172	0.172 0.172 0.172	0.172 0.172 0.172 0.172	0.172 0.172 0.172	0.172	0.172 0.172 0.172	0.172 0.172 0.172	0.172 0.172 0.172	0.172 0.172 0.172	$0.172 \\ 0.172$	0.172 0.172 0.172	0.172
LPY 1b)	∆healc SRI CW (2)	5,042 9,615 12,792	3,824	36,800 60,299 70,385		11,299 45,108 5,643	13,252	9,760 8,548 14,769	7,113 27,377 13,251	13,656 24,562 5,170			, , , ,	
ENTHALPY (Btu/lb)	Δhiriess CW (1)	4,750 9,550 9,550	2,925	11,350 25,316 33,850	8,251 9,452 4,491 4,901	7,077 21,800 4,633	6,850	10,050 9,950 15,240	4,793 17,981 10,190	14,880 24,750 4,948	10,270 21,700 16,330	12,128 19,721	4,750 9,550 9,550	2,945
MODEL STAGNATION PRESSURE (atm)	د 2 1	0.284 0.293 0.735	0.279	$\begin{array}{c} 0.0117 \\ 0.0170 \\ 0.0192 \end{array}$	0.000	$\begin{array}{c} 0.0298 \\ 0.0431 \\ 0.0182 \end{array}$	1.93	0.0198 0.0041 0.094	0.0070 0.0337 0.141	$\begin{array}{c} 0.00510 \\ 0.0195 \\ 0.0925 \end{array}$	0.0107 0.0097 0.0185	0.00582 $0.00847$	0.284 0.293 0.735	0.281
HEAT TRANSFER RATE (Btu/ft <sup>2</sup> sec)	°FAC CW		177	431 780 954	403 104 208	267 1,155 84	1,960	145 65.5 455	44.1 465.5 417.3	99 346 156	66.5 120.5 159.8	90.5 172.3		177
HEAT TF RA (Btu/f	gsri Cw	270 523 1,102	203	400 790 980	,	196 941 76.5	1,850	138 55 455	59.8 505 500	98 345 158	80 114 172	$\frac{103.5}{221}$	269 510 1,065	203
S RATES 2 sec)	• E	0.0258 0.0302	0.0228		0.0245 0.0264 0.0129 0.0207	0.0162 0.0283 0.0111		$0.0134 \\ 0.0063 \\ 0.0289$	0.00413 0.0232 0.0297	0.00986 0.0336 0.0223	0.00505	0.00822 0.0136	0.0155 0.0456 0.103	0.0141
MASS LOSS RATE (lb/ft <sup>2</sup> sec)	• <b>£</b>	0.0302 0.0452 0.117	0.0241	$\begin{array}{c} 0.0166 \\ 0.0194 \\ 0.0222 \end{array}$	0.0266 0.0283 0.0175 0.0222	$\begin{array}{c} 0.0151 \\ 0.0220 \\ 0.00861 \end{array}$	0.179	0.00938 0.00373 0.0226	$\begin{array}{c} 0.00413 \\ 0.0200 \\ 0.0321 \end{array}$	0.0069 0.0280 0.0210	0.00394 0.0048 0.0106	0.00555	0.0181 0.0524 0.104	0.00821
MODEL NO.		A30, 90, 34 A29, 33, 31, 32 A35, 45	A7,48	A62 A63 A64	A38,39 A40,41 A36,37,91 A42,92	A74,44,98 A58,59 A15,17	A73	A1,2,94 A3,4 A5,6	A13, 12, 95 A11,10 A9,8,14	A23, 25, 97 A24, 26 A27, 28	SP48, 49, 96 SP51, 52 SP50	SP66, 65, 85 SP68, 67, 89	SP30, 90, 32 SP29, 31 SP34, 35, 33	SP45, 46
FACILITY		AMPD-Langley	ESB-Langley	MSC-Houston (supersonic)	MSC-Houston (subsonic)	Aerotherm	AVCO	Giannini	Martin	Space General	GDB-Ames	MPDB-Ames	AMPD-Langley	ESB-Langley
MATERIAL		Avcoat 5026-39, HC/G (A) (continued)									Modified Purple Blend Silicone, E4Al (SP)			

Appendix D (Continued)

MATERIAL	FACILITY	MODEL NO.	MASS LOSS RATES (1b/ft <sup>2</sup> sec)	RATES	HEAT TRAN RATE (Btu/ft <sup>2</sup> s	SFER	MODEL STAGNATION PRESSURE (atm)	ENTHALPY (Btu/lb)	LPY 1b)	VE	FRONT SURFACE TEMPERATURE	GW / HT CW Btu/1b
			• 8	• 6	• 0	• 5	Pt 2	Δh <sub>Heas</sub> Cw	∆h <sub>calc</sub> SRI	Reff		
			t I	<u>α</u> ,	'SRI CW	4FAC CW		(1)	C≰ (2)	(3)	-	
Modified Purple Blend Silicone, E4Al (SP) (continued)	MSC-Houston (supersonic)	SP59 SP60 SP61	0.0156 0.0248 0.0277		400 800 990	426 794 958	0.0117 0.0170 0.0192	11,350 24,350 33,850		$\begin{array}{c} 0.172 \\ 0.172 \\ 0.172 \end{array}$	3,985 4,360 4,350	25,641 32,258 35,740
	MSC Houston (subsonic)	SP42, 39 SP40, 41 SP36, 27, 95 SP38	0.0249 0.0354 0.0186 0.0196			400 595 104 205	0.000	8, 329 9, 693 4, 317 4, 850		0.172 0.172 0.172 0.172	4,660 4,400 3,860 3,680	
	Aerotherm	SP56, 53, 97 SP55, 57 SP15, 18	0.00352 0.0120 0.0339	0.00457 $0.0135$ $0.0394$	77.7 200 955	83.2 263 1,160	0.020 0.0298 0.0428	4,416 7,008 21,610		0.172 0.172 0.172	3,060 3,930 4,770	22,074 16,667 28,171
	Giannini	SP1, 2, 90 SP3, 4 SP5, 6	0.00952 0.00263 0.0283	0.0173 $0.00443$ $0.0323$	138 55 456	145 65 456	0.0198 0.0041 0.094	10, 050 9, 950 15, 230	-	0.172 0.172 0.172	3,630 2,860 4,310	14,496 20,913 16,113
	Space General	SP13, 14, 91 SP11, 10, 12 SP9, 8	0.00678 0.0348 0.0218	0.0114 $0.0370$ $0.0215$	101 343 158	102 344 156	0.00512 0.0196 0.0925	14,790 24,640 4,935		0.172 0.172 0.172	3,570 4,410 4,325	14,897 9,856 7,247
G.E. Silicone, ESM1004AP (SG)	GDB-Ames	SG51, 52 SG50 SG35, 39	0.00665 0.0106 0.0143	0.00987 0.0129 0.0169	88 120 170	70.4 122.5 169.3	0.0106 0.0098 0.0183	10, 220 21, 970 16, 440		0.172 0.172 0.172	3,100 3,670 3,750	13,233 11,321 11,888
	MPDB-Ames	SG56, 55, 51 SG58, 57, 53	0.00284	0.00965	90.5 168.5	90.5	0.00582	12,978 19,232		0.172 0.172	3,330	36,761 13,600
	AMPD-Langley	SG22, 49, 24 SG21, 23 SG25	0.0382 0.099 0.157	0.0355	257 514 1,110		0.284 0.293 0.735	4,750 9,550 9,550		0.172 0.172 0.172	4,290 4,180 4,240	6,727 5,191 7,070
	ESB-Langley	SG34, 33	0.0110	0.0127	203	177	0.281	2,940		0.172		18,455
	MSC-Houston (supersonic)	SG59, 60 SG61 SG62	0.0280 0.0433 0.0538		400 790 980	435 780 954	0.0117 0.0170 0.0192	11,350 23,350 34,000		0.172 0.172 0.172	3,620 3,840 4,035	14,286 18,245 18,216
	MSC-Houston (subsonic)	SG28 SG29,30 SG26,27,50	0.0612 0.0859 0.0264	0.0891		400 607 105	1.0 1.0	8,050 9,175 4,627		0.172 0.172 0.172	3,780 3,850 3,565	
•	Aerotherm	SG41, 42 SG14, 43 SG12, 15	0.00282 0.0208 0.0644	0.00750 0.022 0.0655	69.1 174 939	80.3 261 1,150	0.0202 0.0294 0.0425	4,003 7,517 21,420		0.172 0.172 0.172	3,070 3,680 4,050	24,504 8,365 14,581
	Giannini	SG1, 45 SG2, 3 SG4, 5	0.0165 0.00186 0.0635	$\begin{array}{c} 0.0216 \\ 0.00617 \\ 0.0605 \end{array}$	138 55 456	145 65.5 456	0.0197 0.0041 0.095	16,050 9,950 15,240		0.172 0.172 0.172	3,560 2,860 3,710	8,363 29,570 6,656

Appendix D (Continued)

MATERIAL	FACILITY	MODEL NO.	MASS LOSS RATES (1b/ft 2sec)	RATES (ec)	HEAT TRAN RATE (Btu/ft <sup>2</sup>	(SFER	MODEL STAGNATION PRESSURE (atm)	ENTHALPY (Btu/lb)	ALPY /1b)	EFFECTIVE MODEL RADIUS (ft)	FRONT SURFACE TEMPERATURE (°R) $\epsilon$ = 1	$\frac{q}{SRI} / \frac{m}{t}$ $\frac{CW}{Btu/lb}$
			E	Q. • ⊟	° gsri Cw	° FAC CW	Pt 2	Δh meas CW (1)	Δh calc SRI CW (2)	Heff	દ- જ	
G.E. Silicone ESM1004AP (SG) (Continued)	Martin	SG6, 46 SG10 SG9, 8, 7	0.00264 0.0430 0.0890	0.00439	55.9 520 512	43.3 475 411	0.0070 0.0170 0.142	4,903 17,650 9,894		0.172 0.172 0.172	2,890 3,960 3,940	21,174 12,093 5,752
Spa	Space General	SG16,48 SG20 SG18,19	0.00282 0.0556 0.0540	0.00568	103 344 156	104 345 154	0.00511 0.0195 0.0925	14,820 24,738 5,012		0.172 0.172 0.172	3, 470 3, 980 3, 885	36,525 6,187 2,888
Teflon, TFE (T) MFD	MPDB-Ames	T117,116 T115,112	0.0214 0.0306		100	88.8	0.00597	11,613 18,922		0.172		4,672
AMP	AMPD-Langley	T124 T126	0.121		516 1,055	,	0.293	9,550		0.172		4,264
OSW	MSC-Houston (supersonic)	T129,130 T131 T133	0.0543 0.0656 0.0790		410 790 960	435 780 928	$\begin{array}{c} 0.0117 \\ 0.0170 \\ 0.0192 \end{array}$	11,350 24,350 34,350		0.172 0.172 0.172		7,550 12,042 12,151
Cor	Cornell	T118 T119	0.590		1,091 1,108	618 624	10.3 10.3	1,910		$0.172 \\ 0.172$		1,849
AVCO	8	T110 T111 T113,114	0.612 0.690 0.412		1,640 3,340 1,850	1,640 3,400 1,960	5.62 4.92 1.93	2,280 6,670 7,110		0.172 0.172 0.172		2,679 4,840 4,490
AME	AMPD'Langley	T153,154 T156,157 T160,161 T164,165	0.0220 0.0203 0.0299 0.0458	·	74 92.4 133.5 250		0.0192 0.0192 0.0192 0.0192	9,250 9,450 9,400 8,650		0.550 0.344 0.172 0.0416		3,363 4,551 5,464 5,458
Mar	Martin	T163,162 T158,159 T155,154 T150,151	0.0200 0.0292 0.0476 0.0683	<del></del>	250	144 180 250 451	0.0204 0.0204 0.0203 0.0202	12,500 12,246 12,145 12,300		0.688 0.344 0.172 0.0416		5,252
High-Density Phenolic- MPI Nylon (P)	MPDB-Ames	1A5 2A7	0.0117	-	116.1 223.7	102.5	0.00597 0.00812	13,207 18,725		$0.172 \\ 0.172$		9, 923 13, 081
<del></del>	AMPD-Langley	P11A2 P11A3, P11A4	0.0369		524 996		0.293 0.735	9,550 9,550		$0.172 \\ 0.172$		14,200 14,821

Appendix D (Concluded)

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	gSRI/mgCW CW Btu/lb		18,264 29,893 36,704	9,761 10,603 13,405 16,972	3,643
	FRONT SURFACE TEMPERATURE (°R) $\epsilon$ = 1	E.		·	
	EFFECTIVE MODEL RADIUS (ft)	Reff (3)	0.172 0.172 0.172	0.172 0.172 0.172 0.172	0.172
	LPY 1b)	Δh calc SRI CW (2)			
	ENTHALPY (Btu/lb)	∆h meas C₩ (1)	11,350 26,350 34,150	2,280 6,540 7,110 7,110	1,950
ranea)	MODEL STAGNATION PRESSURE	P 2	0.0117 0.0170 0.0192	5.62 4.92 1.93 1.93	9.8
Appenaix D (Concraded)	HEAT TRANSFER RATE (Btu/ft <sup>2</sup> sec)	GW	432 818 954	1,640 3,400 1,960 1,960	630 1,028
Appenata	HEAT T R/ (Btu/	°. qSRI CW	400 840 980	1,640 3,340 1,850 1,850	1,093 1,666
	S RATES sec)	Ω. • E			
	MASS LOSS RATES (1b/ft <sup>2</sup> sec)	⊒• •	0.0219 0.0281 0.0267	0.168 0.315 0.138 0.109	0.300 0.970
	MODEL NO.		P11B1, P11B2 P11B3 P11B5	P12A2 P12A3 P12A4 P12A5	P12B1, P12B6 P12B4
	FACILITY		MSC-Houston (supersonic)	AVCO	Cornell
	MATERIAL	igan kanadana (irang nyayayan yaya	High-Density Phenolic- MSC-Houston Nylon (P) (supersonic)		

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#### APPENDIX E

#### DIMENSIONLESS CORRELATION OF PREVIOUS DATA

This appendix provides a dimensional analysis of ablation variables and suggests several approaches to correlating mass loss data. These approaches are used to interpret data from the Phase I round robin and to compare them with data from the ablation literature.

# A. Dimensional Analysis of Mass Loss Data

The correlation data given in Table I of the Phase I report 1 showed some interesting similarities between the power functions for Teflon and high-density phenolic-nylon. For each correlation used, except where the exponents were adjusted (see Ref. 1, Table I, Note 4), the exponents on the heating rate measured by the SRI calorimeter were identical or varied by less than 5 percent. The exponents on the stagnation pressure had greater spread but were less critical. It was therefore considered that a correlation might be the same for both Teflon and high-density phenolic-nylon and that only the proportionality constant would differ because of differences in material parameters.

A dimensional analysis of the type described by Buckingham 18 was therefore undertaken. The analysis considered two types of variables: those pertaining to the model and those pertaining to the environment to which they are exposed. The model variables include:

 $\dot{m}_t$  = total mass loss rate—lb/ft<sup>2</sup> sec

 $R_{eff}$  = effective radius of curvature—ft

 $\dot{\Delta} H_D$  = overall heat of decomposition required to convert the original abalation material to gaseous end products—Btu/lb\*

<sup>\*</sup> The abbreviation lb will always be used to designate pound mass but pound force when combined with the word "force."

The environmental variables are:

 $\dot{q}_{CW}$  = cold wall heat transfer rate—Btu/ft<sup>2</sup> sec

 $P_{t_0}$  = model stagnation pressure—atm

 $\triangle h$  = enthalpy potential; stream enthalpy minus cold wall enthalpy—Btu/lb

 $S_R$  = proportionality constant in Fay-Riddell relation, see equation (5)—ft<sup>1.5</sup> sec atm<sup>0.5</sup>/lb

This last variable is shown by Fay-Riddell<sup>7</sup> to be given by the fluid properties of the gas stream, as follows:

$$S_{R} = \frac{\left(\frac{p}{\rho\mu}\right)_{c}^{0.1} \left(\frac{p}{\rho\mu}\right)_{s}^{0.4} \left(\frac{\rho}{2p}\right)_{s}^{0.25} \left[\frac{1}{1 - (p_{\infty}/p_{s})}\right]^{0.25}}{0.763 \left[\frac{\psi(N_{Le})}{N_{Pr}^{0.6}}\right]}$$
(E-1)

where p,  $\rho$ , and  $\mu$  are pressure, density, and viscosity, respectively; the subscripts c, s, and  $\infty$  denote front edge of char layer, front edge of boundary layer, and free stream, respectively; and

$$\psi(N_{Le}) = [1 + (N_{Le}^{0.52} - 1)(H_D/h_s)]$$
 (E-2)

Here  $N_{Le}$  and  $N_{Pr}$  are the Lewis and Prandtl numbers for the gas,  $H_D$  is the heat of dissociation (and ionization), and  $h_s$  is the stagnation enthalpy of the gas. For the fluid properties of air under the usual range of reentry conditions, the numerical value of  $S_B = 24 (=1/0.0417)$ .

The term  $\Delta H_D$  is actually an average expression of all the thermal and chemical parameters describing the pyrolysis of each specific ablating material. As such it includes, in a complex manner, such variables as char and virgin polymer density, thermal conductivity, specific heat, heats of pyrolysis, thermal properties of the gas products, etc.

The above model and environmental variables must next be expressed in terms of as few dimensions as possible, e.g., mass, length, and time. This can be done by using the appropriate conversion factors, as follows:

VARI ABLE	UNITS	CONVERTED VARIABLE	CONVERTED UNITS
m <sub>t</sub>	$lb/ft^2$ sec	m <sub>t</sub>	$\frac{1}{1}$ lb/ft <sup>2</sup> sec
$R_{eff}$	ft	$R_{eff}$	ft
$\Delta H_{\mathrm{D}}$	Btu/lb	$\Delta H_D J_{m} g_{c}$	$\mathrm{ft}^2/\mathrm{sec}^2$
d <sup>CM</sup>	$Btu/ft^2$ sec	$\mathbf{q}_{CW}\mathbf{J}_{m}\mathbf{g}_{c}$	lb/sec <sup>3</sup>
$P_{t_2}$	atm	$P_{t_2}F_{p}g_{c}$	$lb/ft sec^2$
∆h	Btu/lb	△hJ <sub>m</sub> g <sub>c</sub>	${ m ft}^2/{ m sec}^2$
$S_R$	$\mathrm{ft}^{1.5}$ sec $\mathrm{atm}^{0.5}/\mathrm{lb}$	$S_R^2 F_p g_c$	$\mathrm{ft}^2/\mathrm{lb}$

where

$$g_c = 32.17 \text{ lb ft/lb force sec}^2$$

$$J_m = 778 \text{ lb force ft/Btu}$$

$$F_p = 2116 \text{ lb force/ft}^2 \text{ atm} . \qquad (E-3A)$$

For convenience these conversion factors can be combined as

$$K = (J_{m}g_{c})^{0.5}/F_{p}g_{c} = 2.325 \times 10^{-3} \text{ ft}^{2} \text{ atm sec/lb}^{0.5} \text{ Btu}^{0.5}$$
 (E-3B)

According to the Rayleigh rule, <sup>19</sup> these seven dimensional variables (the conversion constants do not count) can be combined into four dimensionless groups (number of variables, 7, minus number of dimensions, 3). The most convenient forms of these were found to be

$$\pi_{\rm m} = \dot{m}_{\rm t} (S_{\rm R})^2 R_{\rm eff} / (\Delta H_{\rm D})^{0.5} K$$
 (E-4)

$$\pi_{q} = \dot{q}_{CW}(S_R)^2 R_{eff}/(\Delta H_D)^{1.5} K \qquad (E-5)$$

$$\pi_{\rm p} = P_{\rm t_2}(S_{\rm R})^2 R_{\rm eff} / \Delta H_{\rm D}(K)^2$$
 (E-6)

$$\pi_{f} = \dot{q}_{CW} S_{R}(R_{eff})^{0.5} / \Delta h(P_{t_{2}})^{0.5}$$
 (E-7)

The last dimensionless group is actually the Fay-Riddell relation when  $\pi_{\rm f}$  is unity. Thus, this group is equal to unity under supersonic flow conditions.

Using the same type of power function as that used in the correlations in the Phase I report<sup>1</sup>, a simple relation between these groups is

$$\pi_{m} = a_{0} \pi_{q}^{n} \pi_{p}^{m} \pi_{f}^{s} \qquad (E-8)$$

Under supersonic conditions this reduces to

$$\pi_{m} = a_{0} \pi_{q}^{n} \pi_{p}^{m} \qquad (E-9)$$

Expansion of (E-9) in dimensional terms leads to

$$\dot{m}_{t} = a_{0} (S_{R}^{2}R_{eff})^{n+m-1} (\Delta H_{D})^{(1-3n-2m)/2} (K)^{1-n-2m} (\dot{q}_{CW})^{n} (P_{t_{2}})^{m} (E-10)$$

For constant effective radius\* and material, this reduces to

$$\dot{m}_{t} = a(\dot{q}_{CW})^{n}(P_{t_{2}})^{m} \qquad (E-11)$$

with

$$a = a_0 (S_R^2 R_{eff})^{n+m-1} (\Delta H_D)^{(1-3n-2m)/2} (K)^{1-n-2m}$$
 (E-12)

Equation (E-11) is identical to the correlation given in Equation (22) of the Phase I Report.<sup>1</sup> Therefore, the data from that report can be used in the dimensional correlation.

# B. Interpretation of Results for High-Density Ablation Materials

### 1. Combined Correlation for Teflon and Phenolic-Nylon

The similarity between the values of n for the Teflon and phenolic-nylon correlations in the Phase I report led to an attempt to combine these data into a single correlation. This required a two-step process which was iteratively performed on a computer. First, values of n and m were assumed. The calculated values for the SRI calorimeter (cold wall heating rate), and including Giannini and Martin data, were from the Phase I correlation

<sup>\*</sup> The value of  $R_{\rm eff}$  for the Phase I round robin was 0.172 ft.

,	TEFLON	PHENOLIC NYLON	AVERAGE
n	0.57	0.55	0.56
m	0.25	0.13	0.19
a	0.0060	0.0018	

In the first step, the average values of n and m were used in the regression program to calculate the value of "a" in Equation (E-11) for the form showing the highest multiple correlation coefficient for the Teflon data; similarly the best value of "a" for the phenolic-nylon data was calculated. For models of the same effective radius, Equation (E-12) shows that

$$(a)_{p}/(a)_{T} = [(\Delta H_{D})_{p}/(\Delta H_{D})_{T}]^{(1-3n-2m)/2}$$
 (E-13)

Based on Chapman's work<sup>20</sup>, the heat of decomposition,  $\Delta H_D$ , for Teflon was taken to be 940 Btu/lb. Equation (E-13) then permitted calculation of  $(\Delta H_D)_p$  from the two values of "a".

In the second step, the values of  $\pi_{\rm m}$ ,  $\pi_{\rm q}$ , and  $\pi_{\rm p}$  were calculated for both Teflon and phenolic-nylon, using the appropriate values of  $\Delta {\rm H}_{\rm D}$ , and the regression program was used to calculate new values of n and m, and also a<sub>0</sub> for the combined data. If these values of n and m were those initially assumed, the iteration was stopped; otherwise these were used as the new input to the first step of the program.

The results of the completed iteration were

n = 0.54, m = 0.19, 
$$a_0$$
 = 1.01  
 $\triangle H_D$  = 940 Btu/lb for Teflon (assumed)  
 $\triangle H_D$  = 6470 Btu/lb for phenolic-nylon

The last value is intermediate to the range of theoretical values (4300 to 7300 Btu/lb) quoted by Wick  $^{21}$ .

Thus, for the Phase I round-robin data, the correlation obtained by the regression program, namely,

$$\pi_{\rm m} = 1.01 \, \pi_{\rm q}^{0.54} \pi_{\rm p}^{0.19}$$
 (E-14)

showed a standard deviation of 10.7 percent. The data include that from Giannini and Martin\* but exclude runs in which the heating load (heating rate multiplied by the run duration) was less than 2000 Btu/ft<sup>2</sup> for Teflon and 4000 Btu/ft<sup>2</sup> for high-density phenolic-nylon. These exclusions were made to minimize the use of data from the presteady-state period of the runs. A plot of the correlation, with the standard deviation indicated, is shown in Fig. E-1. The range of variables covered was as follows:

$$\dot{m}_{t} = \frac{0.0129 - 0.218 \text{ lb/ft}^2 \text{ sec for Teflon}}{0.00699 - 0.0547 \text{ lb/ft}^2 \text{ sec for phenolic-nylon}}$$

$$\dot{q}_{SRI} = \frac{36 - 726 \text{ Btu/ft}^2 \text{ sec}}{CW}$$

$$P_{t_2} = \frac{0.0066 - 1.18 \text{ atm}}{1215 - 14,960 \text{ Btu/lb}}$$

The range of dimensional variables was

$$\pi_{\rm m} = 3.70 - 303$$
,  $\pi_{\rm q} = 3.03 - 1070$ ,  $\pi_{\rm p} = 15.3 - 23,000$ 

Thus, the correlation, which compares  $(\pi_{\rm m})_{\rm obs}$  with  $(\pi_{\rm m})_{\rm calc}$ , covers nearly a hundredfold range.

The Phase I round robin had only one set of subsonic data, that from Manned Spacecraft Center—NASA, and the Fay-Riddell group,  $\pi_{\rm f}$ , which was calculated using the measured enthalpy, had values ranging from only 0.45 to 0.73. For this reason no attempt was made to determine the value of s in Equation (E-8). In fact there is no evidence that this is the form in which  $\pi_{\rm f}$  should be used in the correlation. Any form involving  $\pi_{\rm f}$  which reduces to Equation (E-9) when  $\pi_{\rm f}$  equals unity is possible.

# 2. Separate Correlation for Teflon and Phenolic-Nylon

An alternative approach is to use the average values of n and m and the resultant values of  $(\Delta H_D)_P$  and  $a_0$  from the first step of the iteration.

In this case the results are

n = 0.56 (assumed), m = 0.19 (assumed, a = 0.93   
 
$$\Delta H_D$$
 = 940 Btu/lb for Teflon (assumed)   
  $\Delta H_D$  = 6040 Btu/lb for phenolic-nylon

The value of q<sub>SRI</sub> was not reported by these two facilities but was estimated from their calibration runs.

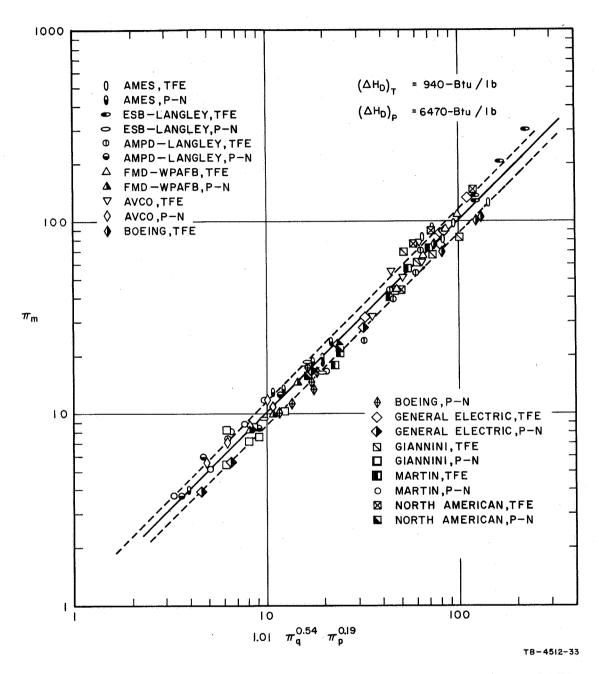


FIG. E-1 DIMENSIONLESS CORRELATION OF PHASE I ROUND-ROBIN DATA

Using the same data as for Equation (E-14) the correlation

$$\pi_{\rm m} = 0.93 \, \pi_{\rm q}^{0.56} \, \pi_{\rm p}^{0.19}$$
 (E-15)

has a standard deviation of 10.9 percent.

This implies that there may be a number of sets of n, m,  $(\triangle H_D)_P$ , and  $a_0$  values having only slightly higher percent standard deviations than those shown for the correlation in Equation (E-14). Calculations in which n and m were varied by  $\pm 0.05$  and  $(\triangle H_D)_P$  by  $\pm 1500$  Btu/lb showed that the standard deviation increased only a few percent. Typical results are given below for calculations based on simple values of n and m.

•	SET 1	SET 2	SET 3	SET 4
n (assumed)	0.5	0.54	0.56	0.6
m (assumed)	0.25	0.19	0.19	0.15
a <sub>0</sub> (obtained by iteration)	0.87	1.01	0.93	0.88
$(\triangle H_D)_p$ (obtained by iteration)	8.390	6470	60 40	6900
Percent standard deviation (calculated)	13.1	10.7	10.9	12.1

It is apparent therefore that an independent source for the value of the overall heat of decomposition,  $\Delta H_D$ , would eliminate the iterative process and would permit more accurate values for  $a_0$  and percent standard deviation to be obtained. In addition, it is not certain that the Teflon and high-density phenolic-nylon data should be forced into the same correlation. If they should not be combined, then it is not so important that the value of  $\Delta H_D$  be known since Equation (E-10), which is based on the same dimensional analysis, can be written as

$$\dot{m}_{t} = b(R_{eff})^{n+m-1} \dot{q}_{CW}^{n} P_{t_{2}}^{m}$$
 (E-16A)

or alternatively,

$$\dot{m}_{t}(R_{eff}) = b(\dot{q}_{CW}R_{eff})^{n}(P_{t_{2}}R_{eff})^{m} \qquad (E-16B)$$

where

$$b = a(R_{eff})_{RR}^{1-n-m}$$
 with  $(R_{eff})_{RR} = 0.172 \text{ ft}$  (E-17)

and with "a" defined as in Equation (E-12). Thus, the constant b will be specific for each material, although its numerical value will depend on the units used for  $\dot{m}$ ,  $R_{eff}$ ,  $\dot{q}_{CW}$ , and  $P_{t_2}$  and on the values of n and m found for that material. Therefore, from the Phase I round robin the values of "a", n, and m are those already tabulated preceding Equation (E-13). For  $\dot{m}_t$  in  $lb/ft^2$  sec,  $\dot{q}_{CW}$  in  $Btu/ft^2$  sec, and  $P_{t_2}$  in atm, the numerical values of b are

(b) 
$$_{T} = 0.0044 \text{ lb } \text{ft}^{-0.68} \text{ sec}^{-0.43} \text{ Btu}^{-0.57} \text{ atm}^{-0.25}$$

(b)<sub>p</sub> = 
$$0.0010 \text{ lb } \text{ft}^{-0.58} \text{ sec}^{-0.45} \text{ Btu}^{-0.55} \text{ atm}^{-0.13}$$

Thus, the mass loss rates for Teflon and high-density phenolic-nylon become

$$(\dot{m}_{t})_{T} = 0.0044 (R_{eff})^{-0.18} (\dot{q}_{CW})^{0.57} (P_{t_{2}})^{0.25}$$
 (E-18A)

$$(\dot{m}_{t})_{p} = 0.0010 (R_{eff})^{-0.32} (\dot{q}_{CW})^{0.55} (P_{t_{2}})^{0.13}$$
 (E-18B)

Another form of these relations is

$$(\dot{m}_t R_{eff})_T = 0.0044 (\dot{q}_{CW} R_{eff})^{0.57} (P_{t_2} R_{eff})^{0.25}$$
 (E-19A)

$$(\dot{m}_{t}R_{eff})_{p} = 0.0010 (\dot{q}_{CW}R_{eff})^{0.55} (P_{t_{2}}R_{eff})^{0.13}$$
 (E-19B)

These dimensionally valid equations, as well as the dimensionless correlation combining the data for the two materials, can be checked with the results from Phase II; this is done in Sec. IV-B.

# C. Inclusion of Literature Data

The open literature on supersonic arc-jet testing was reviewed to locate ablation data on Teflon and high-density phenolic-nylon. A major requirement was that mass loss rate, cold wall heating rate, \* stagnation pressure, and effective radius either be directly tabulated or capable of being calculated from other tabulated data. In addition, the composition of the materials had to be nearly identical to those used in the round robin.

The applicable data and their sources are given in Table E-1. These data were converted into the dimensionless form of Equation (E-14). A plot of this correlation is given in Fig. E-2 which can be compared directly with Fig. E-1. As can be seen, only the Walberg data on phenolic-nylon models do not fit the correlation, this will be considered in more detail in Sec. IV-B. Excluding the Walberg data, the standard deviation for the literature data is 9.6 percent. Combined with the round-robin data the overall standard deviation is 10.3 percent. The range of variables covered was

$$\dot{m}_{\rm t} = \begin{cases} 0.0132 - 1.22 \text{ lb/ft}^2 \text{ sec for Teflon} \\ 0.0233 - 0.596 \text{ lb/ft}^2 \text{ sec for phenolic-nylon} \\ \dot{q}_{\rm CW} = 21.6 - 3000 \text{ Btu/ft}^2 \text{ sec} \\ P_{\rm t_2} = 0.0030 - 33.0 \text{ atm} \end{cases}$$

$$\Delta h_{\rm meas} = 420 - 7470 \text{ Btu/lb} \\ R_{\rm eff} = 0.0156 - 0.55 \text{ ft}$$

The dimensionless variables had a range of

$$\pi_{_{\rm m}} = 2.24 - 1130$$
,  $\pi_{_{\rm q}} = 1.47 - 2510$ ,  $\pi_{_{\rm p}} = 37.1 - 1,260,000$ 

For the literature data the correlation, which compares  $(\pi_{_{\rm m}})_{_{\rm obs}}$  with  $(\pi_{_{\rm m}})_{_{\rm calc}}$ , has nearly a five hundredfold range. In combining the roundrobin and literature data, the range of variables was

$$\dot{m}_{\rm t} = \begin{cases} 94.5\text{-fold for Telfon} & P_{\rm t_2} = 11,000\text{-fold} \\ 85.2\text{-fold for phenolic-nylon} & \Delta h_{\rm meas} = 35.7\text{-fold} \end{cases}$$

$$\dot{q}_{\rm CW} = 139\text{-fold} & R_{\rm eff} = 35.3\text{-fold}$$

The heating rate must be that measured, or calculated, for a calorimeter having the same shape and dimensions as the model used.

Table E-I LITERATURE DATA FOR THE ABLATION OF TEFLON AND HIGH DENSITY PHENOLIC-NYLON

MODEL RADIUS Reff (ft) (ft)	0.206 0.103 0.0686 0.0343	0.0208 0.0208 0.0208 0.0689 0.0345 0.0345 0.0345	$\begin{array}{c} 0.0156 \\ 0.0156 \\ 0.0156 \\ \end{array}$	0.025 0.025 0.025 0.025	0.55	0.55	0.1155 0.1155 0.1155 0.1155 0.1155 0.1155
ENTHALPY  h  (Btu/lb)	2190 2490 2700 2990	2700 2950 3950 6600 2700 44200 4450 4450	1170 1840 2600	1900 1870 1850 1670	2410 3130	940 1550 1450 1010	6380 7616 6888 5992 5460 6720 7644
MODEL STAGNATION P t 2	0.031 0.031 0.031 0.031	88.1 1.866.455 1.866 55.7880 6.758 7.22	444	33.0 21.5 14.4 8.0	5.1	13.0 20.2 8.7 1.46	0.039 0.044 0.041 0.037 0.016 0.021
HEAT TRANSFER  *  *  *  *  *  *  *  *  *  *  *  *  *	21.6 44.7 61.2 87.0	22500 2240 2240 22500 32500 32500 32500	620 1040 1540	2000 1740 1480 1540	340 530	200 410 250 72	148 198 176 132 108 126 158
MASS LOSS RATE (1b/ft <sup>2</sup> sec)	0.0132 0.0315 0.0364 0.057	0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0 0	0.61 0.64 0.92	1.22 0.95 0.67 0.96	0.182 0.229	0.154 0.254 0.174 0.050	0.0538 0.0558 0.0558 0.0510 0.032 0.048
MODEL OR RUN NUMBER			4:29	$\begin{array}{c} 2\\10\\11\\46\end{array}$	$90 (2) \\ 91$	28 (3) 29 30 35	
MATERIAL	Teflon	Teflon	Teflon	Teflon	Teflon	Teflon	Teflon
L ITERATURE SOURCE	Compton, D. L., Winovich, W., Wakefield, R. M. NASA TN D 1332, August 1962	Chapman, A. J. NASA TN'D 1520, April 1963	Farmer, R. W. WADD Tech Report 60-648, November 1960	Graves, K. W. C.A.L. BM-1526-G-8 AD 469965	Winters, C. W. NASA TN D 1500	Winters, C, W. NASA TN D 2383	Vojvodich, N. S. NASA, Ames Research Center

Table E-I (Continued)

MODEL RADIUS Reff (ft)	0.1155 0.1155 0.1155 0.1155 0.1155 0.077 0.077 0.077	00000000000000000000000000000000000000	0.0312 0.0312 0.0312 0.0312	0.0156 0.0156 0.0156 0.0156 0.0156
ENTHALPY ht (Btu/lb)	5796 6860 6860 7168 7168 7560 4519 4519 4519 4519 4524	2840 33000 33000 44000 22130 2540 1540 1540 22250 2250	2420 3080 3310 3040 3210	< 571 744 1038 < 571 744
MODEL STAGNATION P t 2	0.0044 0.0055 0.0057 0.0035 0.0035 0.0030 0.000 0.050 0.050 0.050		0.16 0.045 0.16 0.072 0.16	7.28 7.28 7.28 7.28 7.28
HEAT TRANSFER  * * * * * * * * * * * * * * * * * *	259 652 652 653 1214 203 26	700 743 7453 1075 1051 1040 1187 1275 1305	101 99 176 193 145	306 455 614 308 445 609
MASS LOSS RATE mt (1b/ft <sup>2</sup> sec)	0.0226 0.0245 0.0245 0.0241 0.0241 0.0187 0.0808 0.0564 0.0341	0.385 0.4403385 0.65263347 0.626337 0.626337 0.626337 0.809	0.0394 0.0233 0.0384 0.0368	0.150 0.181 0.230 0.176 0.248
MODEL OR RUN NUMBER	1109 11109 1128 1129 1139 1139 1139	145 (4) 145 (4) 1111 1111 1111 1119 1114 1127	1733951	165321
MATERIAL		Teflon	High-Density Phenolic- Nylon 75 lb/ft <sup>3</sup>	High-Density Phenolic- Nylon 75 lb/ft <sup>3</sup>
LITERATURE SURVEY	Vojvodich, N. S. (Concluded)	Paul, N. J., Falk, F., Kauffman, L. O. Johns Hopkins Report TG881, December 1966	Lundell, J. H. AIAA J. <u>3</u> ,2087-95 (1965)	Walberg, G. D., Crouch, R. K. NASA TN D 3465, August 1966

Table E-I (Concluded)

LITERATURE SOURCE	MATERIAL	MODEL OR RUN NUMBER	MASS LOSS RATE	HEAT TRANSFER RATE 9FAC	MODEL STAGNATION PRESSURE P	ENTHALPY M	MODEL RADIUS Reff
			(lb/ft <sup>2</sup> sec)	(Btu/ft <sup>2</sup> sec)	(atm)	(Btu/lb)	(ft) (4)
Walberg G. D. NASA Langley Research Center	High-Density Phenolic- Nylon 75 lb/ft <sup>3</sup>	1 1 2 2 2 2 2 2 2 2 2 3 3 3 3 3 3 3 3 3	0.181 0.239 0.208 0.218 0.179 0.179 0.596	769 794 874 1087 648 875	7.20 7.31 7.49 7.65 5.17 10.69	1495 1530 1663 1000 934 980	0.0208 0.0208 0.0208 0.0688 0.0688

(1)  $R_{eff} = R_{Hemisphere} = 3.3 R_{flat Face}$ .

(2) Free flight data. Instantaneous rates at 90 and 91 sec after launch; the period of maximum velocity during flight. Enthalpy and stagnation calculated from velocity and air density.

Free flight data, Instantaneous rates at 28, 29, 30, and 35 sec after launch. Enthalpy, stagnation pressure calculated from velocity and air density.

Model stagnation pressure from relation:  $P_{t_2} = 0.75 P_{t_1}$ . Heat Flux calculated from relation:  $q = 0.0417 \left(P_{t_2}/R_{eff}\right) \cdot \frac{5}{h_{ARC}} - 150$ .

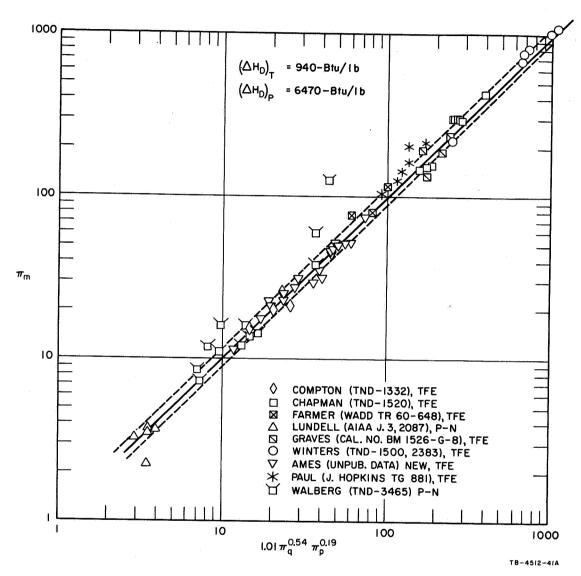


FIG. E-2 DIMENSIONLESS CORRELATION OF LITERATURE DATA

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